

Appendix B

SAPS Design Calculation

STEP 1: Data Collection

Sample	Flow(Q) GPM	pH su	DO mg/l	Fe mg/l	Mn mg/l	Al mg/l	Acidity mg/l *	Non-Mn Acidity mg/l ***
1	6.0	3.5	6.2	26.8	10.7	1.6	62.0	42.5
2	5.0	3.1	6.6	36.6	14.1	0.7	134.0	108.4
3	1.5	3.1	7.0	44.7	19.2	1.0	182.0	147.1
4	0.5	3.0	7.9	25.5	20.9	0.9	186.0	148.0
5	1.5	3.2	7.5	27.2	12.0	1.1	104.0	82.2
6	3.0	4.7	6.5	13.2	5.9	1.5	44.0	33.3
7	1.2	3.8	6.8	12.0	9.7	1.3	74.0	56.4
8	11.8	3.7	6.5	31.9	6.4	0.9	60.0	48.3
9	5.5	3.2	5.3	27.3	5.0	0.8	58.0	48.9
10	4.7	3.3	5.9	42.4	10.2	0.9	104.0	85.5
11	12.0	4.3	7.2	30.1	6.7	0.3	48.0	35.8
12	4.5	3.4	6.2	51.7	14.9	1.1	130.0	102.9

*CaCO₃ equivalent

** Determined by (Acidity - 1.818Mn)

STEP 2: Development of the 95% Confidence limit

	Flow(Q)	pH	DO	Fe	Mn	Al	Acidity	Non-Mn Acidity
$n =$	12	12	12	12	12	12	12	12
$\chi =$	4.77	3.35*	6.63	30.78	11.31	1.01	98.83	78.27
$t_{.05[n-1]} =$	1.796	1.796	1.796	1.796	1.796	1.796	1.796	1.796
$s =$	3.81	NA**	0.71	11.80	5.17	0.35	50.01	41.04

*median value **not applicable

$$L_{upper} = \chi + t_{.05[n-1]} s n^{-0.5}$$

	Flow(Q)	pH	DO	Fe	Mn	Al	Acidity	Non-Mn Acidity
$L_{upper} =$	6.74	NA	7.00	36.90	13.99	1.19	124.76	99.54

STEP 3: Residence Time Calculation**ONE CELL**

$$\ln(t_{r1\text{ cell}}) = (-0.017 \times 36.90) + (0.012 \times 99.54) + 2.21 = 2.78 \text{ ln hours}$$

$$t_{r1\text{ cell}} = e^{\ln(t_{r1\text{ cell}})} = 16.07 \text{ hours}$$

using a bulk void volume of 0.50

$$\text{Volume of limestone} = 0.227 \times 6.74 \text{ GPM} \times 16.07 \text{ hr} \times 0.50^{-1} = 49.19 \text{ m}^3$$

using a bulk density of 0.8 MT/m³

$$\text{Tonnage of Limestone} = 49.19 \text{ m}^3 \times 0.8 \text{ MT/m}^3 = 39.35 \text{ MT}$$

20 YEAR DESIGN LIFE

$$\text{Volume of limestone} = \frac{V_{\text{is}} = 1.99 Q C T}{\rho_b \times}$$

$$\frac{1.99 (6.74 \text{ GPM})(149.54 \text{ mg/l})(20 \text{ yr})}{0.9 \times 1600 \text{ kg/m}^3}$$

$$\text{Tonnage of Limestone} = 27.85 \text{ m}^3 \times 0.8 \text{ MT/m}^3 = 22.28 \text{ MT}$$

$$\text{TOTAL LIMESTONE} = 49.19 \text{ m}^3 + 27.85 \text{ m}^3 = 77.04 \text{ m}^3$$

$$39.35 \text{ MT} + 22.28 \text{ MT} = 61.63 \text{ MT}$$

TWO CELLS

$$\frac{\ln(t)_{2 \text{ cell}}}{t_{2 \text{ cell}}} = \frac{\ln(t)_1 / 2}{e^{\ln(t)_{2 \text{ cell}}}} = 1.39 \text{ ln hours}$$

$$t_{2 \text{ cell}} = 4.01 \text{ hours}$$

using a bulk void volume of 0.50

$$\text{Volume of limestone} = 0.227 \times 6.74 \text{GPM} \times 34.01 \text{hr} \times 0.50^{-1} = 12.27 \text{ m}^3$$

using a bulk density of 0.8 MT/m³

$$\text{Tonnage of Limestone} = 12.27 \text{m}^3 \times 0.8 \text{ MT/m}^3 = 9.81 \text{ MT}$$

20 YEAR DESIGN LIFE

$$\text{Volume of limestone} = \frac{1.99 \text{ Q C T}}{\rho_b \times}$$

$$\frac{1.99 (6.74 \text{ GPM})(149.54 \text{ mg/l})(20 \text{ yr})}{0.9 \times 1600 \text{ kg/m}^3}$$

$$\text{Tonnage of Limestone} = 13.93 \text{m}^3 \times 0.8 \text{ MT/m}^3 = 27.85 \text{ m}^3$$

$$= 22.28 \text{ MT}$$

$$\text{Per Cell Basis} = \frac{27.85}{2} = 13.93 \text{ m}^3$$

$$\frac{22.28}{2} = 11.14 \text{ MT}$$

$$\text{Total Limestone per cell} = 12.27 \text{m}^3 + 13.93 \text{m}^3 = 26.19 \text{ m}^3$$

$$9.81 \text{MT} + 11.14 \text{MT} = 20.96 \text{ MT}$$

$$\text{TOTAL LIMESTONE} = 26.19 \text{m}^3 \times 2 = 52.39 \text{ m}^3$$

$$20.96 \text{MT} \times 2 = 41.91 \text{ MT}$$

STEP 4: Organic Matter Calculation

$$\begin{aligned} \text{Volume of Organic Matter} &= 0.227 Q t_{\text{rom}} V_v^{-1} \\ &= 0.227 \times 6.74 \times 25 \times 0.59^{-1} = 64.83 \text{ m}^{3*} \end{aligned}$$

* Since this calculation is independent of the number of cells, this volume of organic matter should be used in each cell.