

**A Quantitative Analysis of a Non-Eruptive Volcanic Event:  
Mt. Spurr, Alaska, 2002-2006**

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## **ABSTRACT**

Mt. Spurr is a volcano in proximity to Anchorage, Alaska and major airline routes making an eruption or episode of unrest potentially hazardous. Between 2004 and 2006, Mt. Spurr underwent such an episode of unrest involving increased seismic activity, CO<sub>2</sub> emissions, ice melting, and debris flows, which was likely forecasted by the increased seismicity of Oct 2002. The timeline of events provide data to construct a model analyzing the thermal energy release and constraining subsurface magmatic and hydrothermal processes during the period of unrest. The results show that the ice cauldron formation and the increase of meltwater temperature could not have been caused by the observed CO<sub>2</sub> release alone and suggest that enhanced hydrothermal heat transfer related to increased CO<sub>2</sub> output could provide the thermal power necessary to drive the melting event. Scaling hydrothermal convection in terms of its Rayleigh number and using boundary layer analysis suggests that the mean permeability of the volcanic edifice prior to the unrest event was  $\sim 10^{-14}$  m<sup>2</sup>. CO<sub>2</sub> release, most likely related to mechanical fracturing of the edifice by over-pressurized fluids at depth and signaled by increased seismicity likely enhanced the hydrothermal Rayleigh number and heat output by a combination of heating and increased permeability.

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## **1. Introduction**

Eruptive events can cause hazards such as ash plumes, massive degassing, and debris flows which may contain large volumes of meltwater. Non-eruptive events may also result in degassing plumes and debris flows, but typically less meltwater which results in flows with higher concentrations of heavy metals and acids (Schaefer *et al.*, 2012). At 61°N, many of the mountains are covered with a persistent mantle of glacial ice and snow, which makes any thermal event a potential for flood hazards and debris flows.

Understanding volcanic unrest events that do not lead to eruption could provide a means of predicting debris flows and degassing events associated with thermal perturbations and ice melting. Monitoring, data collection, and recording of such events, is crucial to the refinement of ice melt models and eruption prediction. This is particularly important along the Aleutian Arc because its location along a great circle route between the 48 contiguous United States and Asia makes Anchorage, Alaska the third busiest air cargo port in the world. An eruptive event could be detrimental to the safety of passenger aircraft as well as to the potential loss of cargo. Application of this model could be applied to other volcanoes, which may assist in the prediction of volcanic unrest or eruption and guide civil evacuation decisions that would reduce the likelihood of loss of life resulting from an eruptive event or from a debris flow in the case of a non-eruptive thermal event.

### ***1.1. Volcanic Unrest***

Volcanic unrest events that culminate in an eruption cause hazards to humans that are more noticeable, and far more interesting, than those in which volcanic unrest does not culminate in an eruption. As such, it is more than likely that those events that do not culminate in eruption are

under-reported or neglected (Moran *et al.*, 2011). A failed eruption can be further defined as an event in which a magma intrusion gets close to the surface, but does not overcome the hindering forces to reach the surface (Moran *et al.*, 2011).

From the perspective of the magmatic system, there are four possible outcomes of volcanic unrest: 1) stalling at depth, with little or no degassing; 2) stalling at shallow depths within two to three kilometers of the surface, accompanied by obvious degassing; 3) a sluggish, highly viscous lava dome extrusion; or 4) an explosive magmatic eruption (Moran *et al.*, 2011).

## ***1.2. Scope of Study***

This paper investigates a non-eruptive volcanic event at Mount Spurr Volcano, Alaska. The event likely began with a shallow seismic swarm as early as October 2002, was perpetuated by another seismic swarm in July 2004 (Power, 2004; Power *et al.*, 2004), and ended in 2006 (Coombs *et al.*, 2006) with the cessation of ice melting and the re-accumulation of snow and ice at the summit of Mount Spurr (Neal *et al.*, 2007).

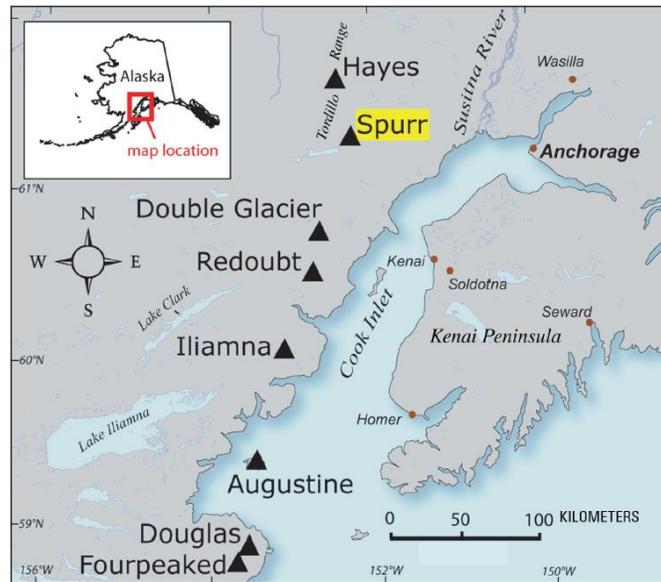
The purpose of this study is to develop a quantitative analysis of this episode of volcanic unrest at Mount Spurr. The timeline of events is marked by the dates of the initial seismic swarm, ice cauldron formation, and the cessation of ice melt. These data, combined with an estimate of the volume of ice melt and rate of CO<sub>2</sub> degassing, provide a means to estimate the thermal energy released, rate of heat flow, and a number of subsurface geophysical parameters that control the rate of heat transfer.

Section 2 discusses the location, geology, and eruptive history of Mount Spurr. In Section 3, AVO data of the non-eruptive event at Mount Spurr in 2002-2006 is analyzed and key constraints are defined. These data provide constraints on the thermal energy and power

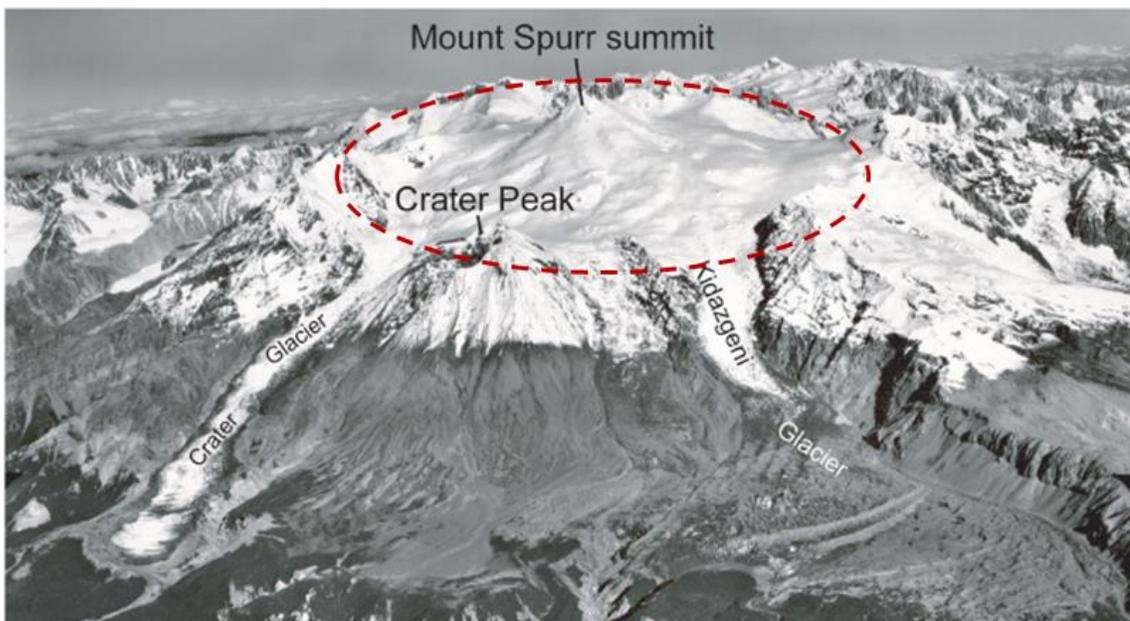
necessary to melt the ice cauldron and heat the resulting meltwater, the duration of the thermal perturbation front, and the maximum distance of heat transport. I then outline a conceptual model of the volcanic unrest event that forms the framework of quantitative analysis. A mechanism of advective heat transport based solely on the mass flow rate and heat output transported by CO<sub>2</sub> degassed as a result of cracking and enhanced CO<sub>2</sub> release is analyzed in Section 4. Section 5 analyzes the potential for a hydrothermal system which extends from the top surface of the magma body to the base of the ice cauldron formed near the summit of Mt. Spurr and offers calculated permeability and planform area of such a system. In Section 6, a hydrothermal system enhanced as a result of degassed volatiles is discussed. A potential state of the pre-enhanced hydrothermal system is developed, changes in conductive boundary layer thickness are modeled, permeability is calculated, and a planform area that agrees with mapped phenomena is deduced. Discussions of the results and implications of this study are found in Section 7, and the conclusions are summarized in Section 8.

## **2. Mount Spurr**

Mount Spurr is a 3,374 m high dominantly andesitic arc strato-volcano in Alaska (Coombs *et al.*, 2012), located approximately 125 km west of Anchorage, Alaska, Figure 1 (Coombs *et al.*, 2006). Mount Spurr is the eastern-most historically active volcano in the Aleutian Arc (Coombs *et al.*, 2006). Mount Spurr summit is approximately 2.5 km north of Crater Peak (Coombs *et al.*, 2006). Both lie within a 5.5 km diameter, mostly ice filled collapsed caldera that formed in the late Pleistocene or early Holocene, Figure 2 (Nye and Turner, 1990). Recent eruptions have occurred at the summit peak of Mount Spurr at approximately 5.2 ka (Riehle, 1985), at Crater Peak in 1953 (Juhle and Coulter, 1955) and in 1992 (Keith, 1995).



**Figure 1.** Cook Inlet region, southern Alaska, showing locations of Mount Spurr and other volcanoes (triangles). Dots, towns and villages. Map courtesy of Janet Schaefer, Alaska Division of Geological and Geophysical Surveys (Coombs *et al.*, 2006).



**Figure 2.** Spurr volcanic complex (see Figure 1 for location). Summit of Mount Spurr sits in an ice-filled collapse caldera, breached to the south; Crater Peak, a historically active satellite vent, sits in the breach. Debris-avalanche deposit in foreground, which was formed during caldera formation is dated at early Holocene (Waythomas and Nye, 2002). View northward; photograph 666-6 by Austin Post, taken September 4, 1964. Modified from (Coombs *et al.*, 2006).

### **3. Analysis of Key Observational Constraints and a Conceptual Model**

Ice melt on summit glacier was tracked through time by using a geographic information system, digital elevation models, and aerial imagery (Coombs *et al.*, 2006). The ice mantle began to show subsidence in June 2004, resulting in the formation of an ice cauldron by 08 July 2004 (Coombs *et al.*, 2006), suggesting that heating began prior to June 2004. By August 2005, the ice cauldron was considered to have expanded to its maximum area of approximately 70,000 m<sup>2</sup> (Coombs *et al.*, 2006; Neal *et al.*, 2007), and depth of approximately 78 to 80 m (Coombs *et al.*, 2006). The estimated melted ice volume was approximately  $5.4 \times 10^6$  m<sup>3</sup> (Coombs *et al.*, 2006). On 14 April 2006, the average temperature of the lake within the ice cauldron was 11 °C; see Table 1 of (Coombs *et al.*, 2006).

Table 1 provides a summarized timeline of events as recorded by the Alaska Volcano Observatory (AVO) and/or its affiliates. This summarized timeline was used to determine the timing of major events for the calculations in this portion of the study.

#### ***3.1. Minimum Thermal Energy Required to Melt the Ice and Heat the Meltwater***

This section is concerned with calculating the thermal energy and heat output necessary to melt the total amount of ice melted within the cauldron, the heating of that meltwater to 11 °C, and the duration of the melting event. These values are considered to be minimum values, as not all of the thermal energy was captured by melting ice; the excess of which was released by fumaroles, radiant heating, or advection into the surrounding environment without measurement.

The amount of thermal energy,  $E$ , needed to melt the ice necessary to form the ice cauldron and heat the melt water is given by:

$$E = \rho_i V_i [c_i (T_{fus} - T_i) + L_{fus} + c_w (T_f - T_{fus})] \quad [1]$$

where  $\rho_i$  is the density of the ice;  $V_i$ , the volume of the ice melted;  $c_i$ , the heat capacity of the ice;  $c_w$ , the heat capacity of water;  $T_{fus}$ , the melting temperature of the ice;  $T_i$ , the initial temperature of the ice;  $L_{fus}$ , the latent heat of fusion of the ice;  $T_f$ , the final temperature of the meltwater. Appendix A contains symbol definitions and values used in calculations.

**Table 1.** Summary timetable of data relating to a thermal event at Mount Spurr, Alaska in 2002-2006.

Event	Date	Reference
seismic swarm, possible magma emplacement and assumed initiation of thermal event	20 Oct 2002	(Dixon <i>et al.</i> , 2003; Power, 2004; Power <i>et al.</i> , 2004)
increased seismicity	Jul 2003	(Power, 2004)
steaming rock below summit, audible gas emission, strong H <sub>2</sub> S noted	19 Mar 2004	(Coombs <i>et al.</i> , 2006)
no deformation noted	03 May 2004	(Coombs <i>et al.</i> , 2006)
obvious subsidence in summit glacier denoting heat transfer to surface geology	20 Jun 2004	(Coombs <i>et al.</i> , 2006)
seismic swarms, possibly contributing to previous magma volume	Jul 2004	(Dixon <i>et al.</i> , 2005; Power, 2004; Power <i>et al.</i> , 2004)
obvious ice cauldron, actual date of ice plug collapse is unknown	08 Jul 2004	(Coombs <i>et al.</i> , 2006)
debris flow on the ESE flank of the summit	10 Jul 2004	(Coombs <i>et al.</i> , 2006)
effective cessation of ice cauldron growth, limited by existing geology	01 Aug 2005	(Coombs <i>et al.</i> , 2006; Neal <i>et al.</i> , 2007)
no appreciable growth in ice cauldron, meltwater temperature recorded as high as 25/45 °C, average temperature of 11 °C	14 Apr 2006	(Coombs <i>et al.</i> , 2006; Neal <i>et al.</i> , 2007)
previously steaming areas on flanks covered in snow, yellow precipitate observed along lakeshore; taken as the definitive end of this thermal event	23 Apr 2006	(Coombs <i>et al.</i> , 2006)

For simplicity, we assume that the glacial ice is initially at its melting point of 0°C, so  $T_i = T_{fus} = 0$  °C. Equation [1], with  $T_2 = 11$  °C and parameter values from Table 2, is then:

$$E = \rho_i V_i [L_{fus} + 11c_w] \quad [2]$$

The minimum thermal energy required to melt  $5.4 \times 10^6 \text{ m}^3$  of ice and heat the resulting meltwater to  $11 \text{ }^\circ\text{C}$  is approximately  $2.1 \times 10^{15} \text{ J}$ ; neglecting the heating of meltwater, the minimum thermal energy,  $E$ , becomes  $1.8 \times 10^{15} \text{ J}$ , showing that a significant majority of the heat input goes into melting the ice.

### ***3.2. Minimum Rate of Heat Release Required to Melt the Ice and Heat the Meltwater***

To calculate heat flow, it is first necessary to determine the dates of melt initiation and melt cessation. Table 2 lists the events from Table 1 which were deemed appropriate for calculating the heat flow related to cauldron growth and meltwater heating. Note that 20 Jun 2004 was selected as the date of cauldron initiation. Although steaming rock was noted on 19 Mar 2004, accompanied by audible gas emission, it was considered too early, as ice had not yet begun melting as far as observation was concerned, as per the entry on 03 May 2004 of no deformation noted. An obvious cauldron had developed by 08 Jul 2004, but was considered as too late a date, due to the fact that by the time the ice plug was able to fall in, creating the obvious cauldron, melting had been previously ongoing prior to that event. Since the thermal properties beneath the ice are unobserved, surface subsidence is the only indication of thermal energy reaching the surface, and so the date of subsidence, 20 Jun 2004 is used to delineate thermal breaching of the rock surface.

**Table 2.** Observed Data for Determining Heat Flow

<b>Description</b>	<b>Date</b>	<b>Reference</b>
obvious subsidence in summit glacier denoting heat transfer to surface geology	20 Jun 2004	(Coombs <i>et al.</i> , 2006)
effective cessation of ice cauldron growth, limited by existing geology	01 Aug 2005	(Coombs <i>et al.</i> , 2006; Neal <i>et al.</i> , 2007)
no appreciable growth in ice cauldron, meltwater average temperature recorded as $11 \text{ }^\circ\text{C}$	14 Apr 2006	(Coombs <i>et al.</i> , 2006; Neal <i>et al.</i> , 2007)

The 21 month duration of the cauldron forming and meltwater heating event is calculated as the difference between the date of cauldron initiation, 20 Jun 2004 (Coombs *et al.*, 2006), and the date which the meltwater temperature was recorded at 11 °C, 14Apr 2006 (Coombs *et al.*, 2006); which is approximately  $5.5 \times 10^7$  s.

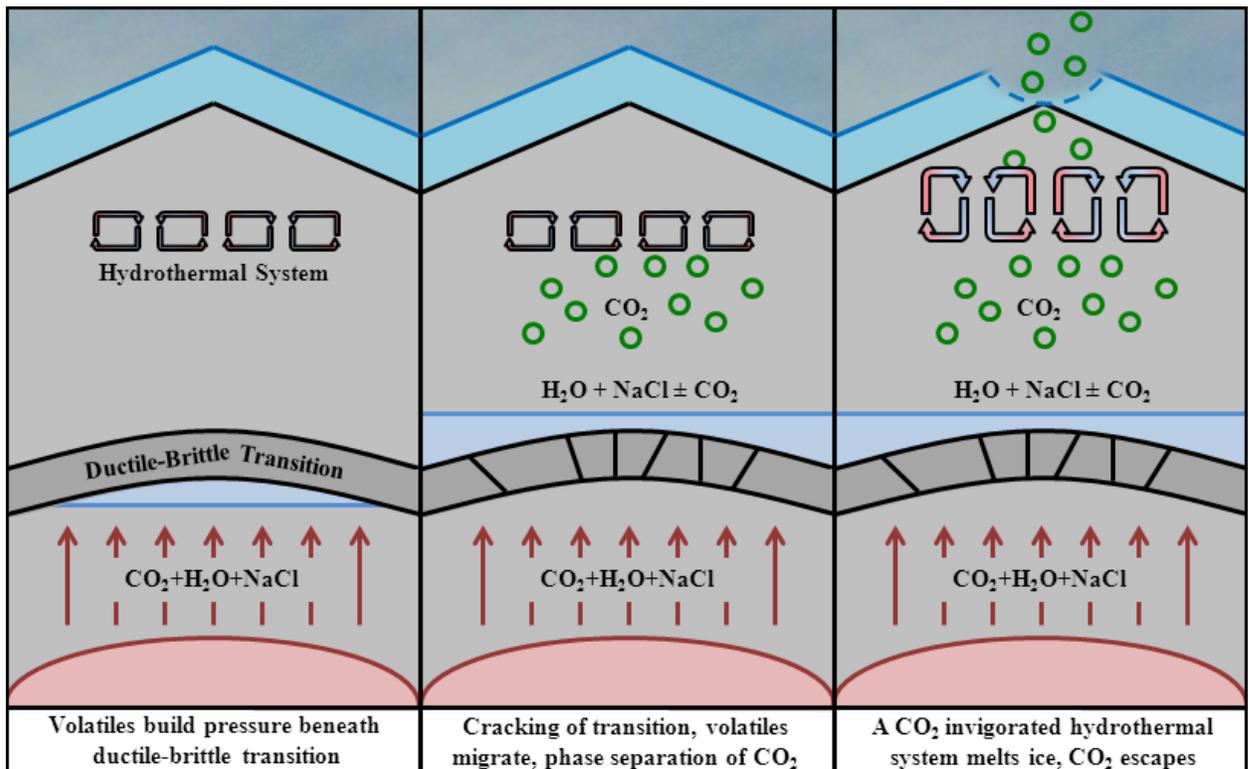
From the minimum thermal energy of  $2.1 \times 10^{15}$  J calculated in Equation [2], and the  $5.5 \times 10^7$  s duration of the event, we calculated the minimum heat flow rate to be approximately 38 MW.

### ***3.3. A Conceptual Model of Thermal Unrest***

Prior to the surface-observable unrest event at Mt. Spurr, the processes within the volcanic complex were in a thermally reduced state, or at least in a state where changes were unnoticed at the surface, suggesting a quasi-steady state of thermal flux. The onset of increased fumarolic activity and the melting of ice suggest that Mt. Spurr had received an influx of thermal energy, which altered that previous state to one of increased activity.

The conceptual model of the volcanic unrest event at Mt. Spurr assumes the presence of a crystallizing subsurface magma body at some depth beneath the Mt. Spurr summit that is separated from a shallower convecting hydrothermal system by a rheological boundary termed the brittle-ductile transition, Figure 3. At temperatures greater than approximately 500 °C, rock tends to deform in a ductile manner, whereas beneath that temperature, rock behaves as a brittle medium (Fournier, 1999). As crystallization of the magma body occurs, a single phase H<sub>2</sub>O-CO<sub>2</sub>-NaCl magmatic fluid exsolves from the melt, migrates to the brittle-ductile transition boundary, and ponds. The accumulating magmatic fluid results in an increase in pressure beneath the brittle-ductile transition, and the rocks eventually fracture when the pressure exceeds

the tensile strength of the overlying rocks (Fournier, 1999). Emplacement of new magma at depth or a tectonic event may also lead to rupture of the brittle-ductile boundary. Once fracturing occurs, the associated pressure drop leads to phase separation to produce a dense H<sub>2</sub>O-NaCl brine that ponds at depth, and a CO<sub>2</sub>-rich vapor phase that migrates upward into the bottom of a hydrothermal system. The drop in pressure as a result of breaking the seal, also results in a pressure pulse that diffuses through the overlying rock (Phillips, 1991).



**Figure 3.** This figure shows the conceptual model of the Mt. Spurr system. Left: A cooling and crystallizing magma body at depth degasses volatiles, the majority of which are CO<sub>2</sub>, H<sub>2</sub>O, and assorted metallic chlorides. These migrate upward until they reach the mostly impermeable brittle-ductile transition and begin to collect and build pressure. Middle: Volatile pressure exceeds lithostatic pressure and ruptures the boundary, temporarily increasing permeability, and allowing the escape of the volatiles. Right: Changes in pressure from lithostatic to hydrostatic cause phase separation in the CO<sub>2</sub>-H<sub>2</sub>O-NaCl system allowing CO<sub>2</sub> gas to escape to the surface, invigorating the preexisting hydrothermal system and escaping to the atmosphere.

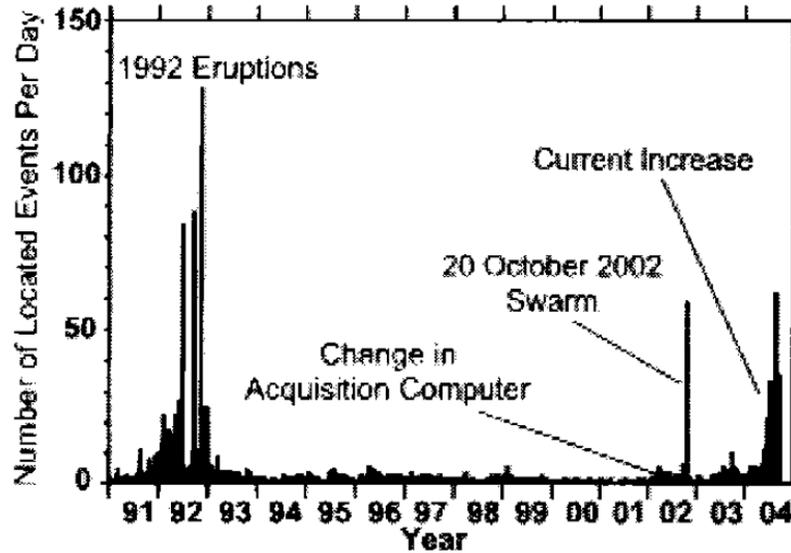
The hydrothermal system is assumed to extend from the brittle-ductile transition to the base of the ice sheet, though the most vigorous circulation is thought to extend only to a depth of a few kilometers (Motyka and Nye, 1993) as suggested schematically in Figure 3. The pressure pulse, together with the CO<sub>2</sub> phase, transfers thermal energy and mechanical energy from depth into the shallow hydrothermal system. The increased thermal energy and likely increase in rock permeability likely enhances the vigor of hydrothermal heat transfer to the base of the ice layer. Some of the CO<sub>2</sub> dissolves into the convecting hydrothermal fluid, but some of the CO<sub>2</sub> moves as a separate phase through the hydrothermal system and transfers heat to the ice layer at the Mt. Spurr summit.

The unrest event discussed in this study began with seismic activity (Coombs *et al.*, 2006; Power *et al.*, 2004), continued with increases in fumarolic activity and outgassing (Coombs *et al.*, 2006; Doukas and McGee, 2007), the melting of  $5.4 \times 10^6 \text{ m}^3$  of ice which caused debris flows (Coombs *et al.*, 2006), and ended with the cooling of the edifice, evidenced by the re-accumulation of snow and ice (Neal *et al.*, 2007). To investigate the mechanism of heat transport, we focus on that portion of the event which altered the thermal state of Mt. Spurr volcano; from the date of the initiating seismic swarm to the date of noticeable surface ice melt.

Figure 4 shows a bar chart of seismic activity beneath Mt. Spurr as a function of time from 1991 to July 2004. The figure shows significant increases in seismicity associated with the 1992 eruption as well as other seismic swarms in October 2002 and July 2004 (Power, 2004).

As suggested by Power (2004), I consider the seismic swarm in October 2002 to be the initiation of the unrest event, which I interpret to be a manifestation of a cracking event and release of volatiles. The seismic swarms in 2003 and 2004 may be associated with additional crustal cracking associated with magma degassing. The timeframe of the thermal perturbation

front is then the second fundamental constraint, beginning in Oct 2002 and affecting the surface by Jun 2004, a duration of 20 months, approximately  $5.3 \times 10^7$  s.



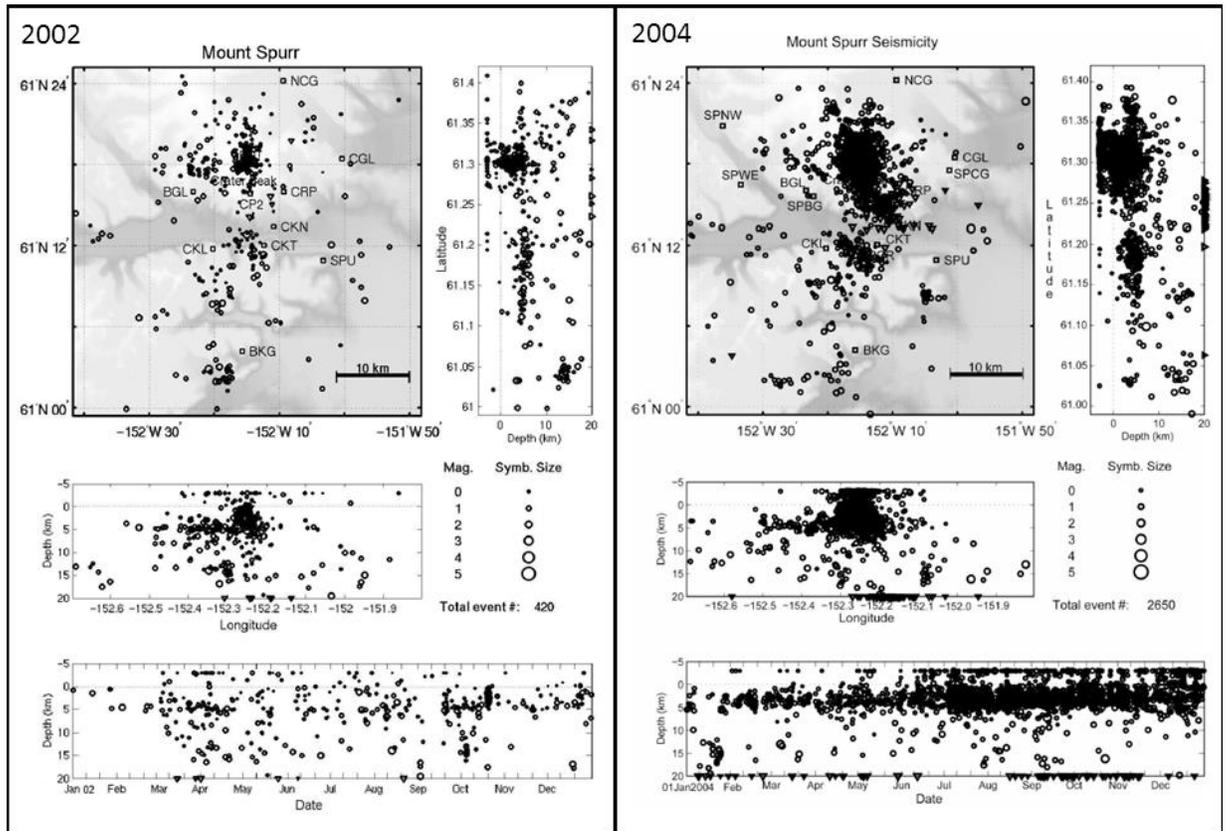
**Figure 4.** Shows the number of earthquakes within 20 km of Mount Spurr from 1991 to mid-2004. Note the small increase in earthquake activity in mid to late-2003 which predates the earthquake swarm in 2004 and the small increase in earthquake activity in early 2001, attributed to changing to the Earthworm acquisition system, which predates the 2002 swarm. A similar character can be found in early to mid-1991, predating the 1992 eruption. from (Power, 2004).

### ***3.4. Maximum Distance of Heat Transport***

In this section, I discuss the maximum distance that heat must travel through the volcanic edifice. It is this measure which will assist in determining the mechanism of heat transfer. This maximum depth is determined by the seismicity recorded at Mt. Spurr during the events as shown in Figure 5.

Figure 5 shows that the 2002 and 2004 hypocenters are generally located beneath the summit of Mt Spurr; their distribution suggests a concentration of hypocenters at a depth of approximately 5 km. Review of the seismic velocity model (Jolly *et al.*, 1994) indicates that

there is a concentration of seismic activity at the depth of 8.4 km beneath Mt. Spurr summit, associated with the shoaling of the brittle-ductile transition to a depth of 5 km below sea level directly beneath the volcanoes of the Cook Inlet. Along the Cook Inlet volcanic axis, however, an uncertainty of  $\pm 2.8$  km is associated with the 2002 and 2004 seismic swarm data (Dixon *et al.*, 2005; Dixon *et al.*, 2003) suggesting that the actual distribution of hypocenters may be spread between 5.6 km and 11.2 km depth beneath the Mt. Spurr summit.



**Figure 5.** (left) A plot of the seismicity of 2002, including the seismic swarm of 20 October 2002, showing a trend at an approximate depth of 5 km. from Figure A2 of (Dixon *et al.*, 2003). (right) A plot of the seismicity of 2004, including the increased seismicity of July 2004 and onwards. from Figure A2 of (Dixon *et al.*, 2005).

Assuming that 5 km depth below sea level represents the brittle-ductile transition (Jolly *et al.*, 1994) and the summit of Mount Spurr is approximately 3,374 m high (Coombs *et al.*, 2006), the

maximum distance heat must travel is then approximately 8 km. Using the maximum distance, and the duration of  $5.3 \times 10^7$  s calculated in the previous section, the maximum velocity of heat transfer is approximately  $1.5 \times 10^{-4}$  m/s.

### ***3.5. Potential Mechanisms for Heat Transport***

There are three possible mechanisms for heat transport within Mt. Spurr volcano: 1) magmatic degassing of volatiles associated with the observed increase in CO<sub>2</sub> flux between 2004 and 2006 (Doukas and McGee, 2007), 2) hydrothermal convection from the brittle-ductile transition to the volcanic surface of the summit of Mt. Spurr, near the base of the ice, and, 3) a combination of the enhanced CO<sub>2</sub> flux coupled with enhanced hydrothermal activity beneath the summit that results from the increased CO<sub>2</sub> flux between 2002 and 2004. The next three sections cover these mechanisms in more detail.

## **4. Melting Ice and Heating Meltwater via Magmatic Degassing of Volatiles**

This section begins by determining an amount of CO<sub>2</sub> necessary to melt the ice and heat the meltwater to 11 °C and comparing that to the observed emissions data. Furthermore, it explores the possibility of enhanced hydrothermal heat transfer resulting from an increase in CO<sub>2</sub> emissions.

### ***4.1. Minimum Mass and Mass Flux of CO<sub>2</sub> Derived from Cauldron Formation and Lake Heating***

In this section we determine the minimum mass and resulting mass flux of CO<sub>2</sub> necessary to melt the  $5.4 \times 10^6$  m<sup>3</sup> of ice and to heat the resulting meltwater to 11 °C. This is a minimum due to the assumption that all the available heat from degassed CO<sub>2</sub> went into melting the ice and

heating the meltwater to 11 °C, with no regard for heating the edifice on its rise to the surface. Forward Looking Infrared images (FLIR), taken by AVO, of the episodic debris flows sourced from the meltwater lake indicate a decrease in temperature from the debris flows in June and July 2004, to those in May 2005 (Coombs *et al.*, 2006). This suggests that the temperature of the meltwater may have been higher than the lake average of 11 °C that was recorded on 14 Apr 2006.

From Equation [2], the minimum thermal energy required to melt the  $5.4 \times 10^6 \text{ m}^3$  of ice and to raise its temperature to 11 °C, was calculated to be  $2.1 \times 10^{15} \text{ J}$ . The minimum mass of CO<sub>2</sub> necessary to obtain this amount of thermal energy can be solved for by using an energy balance equation. By rearranging the equation  $E = m c T$  and substituting the calculated thermal energy for  $E$ , the mass of CO<sub>2</sub> can then be solved for by:

$$m = E / (c_{CO_2} T) \quad [3]$$

where  $m$  is the mass of CO<sub>2</sub>,  $c_{CO_2}$  the specific heat capacity of CO<sub>2</sub>, and  $T$  is the temperature. Figure 6 shows how the mass of CO<sub>2</sub> necessary changes with the temperature of that CO<sub>2</sub> from 200 °C, which corresponds to the approximate hydrothermal temperature (Garchar *et al.*, 2012), and 1000 °C, which corresponds to the approximate temperature of andesitic magma. CO<sub>2</sub> released at the brittle-ductile transition would have a temperature of approximately 500 °C (Fournier, 1999).

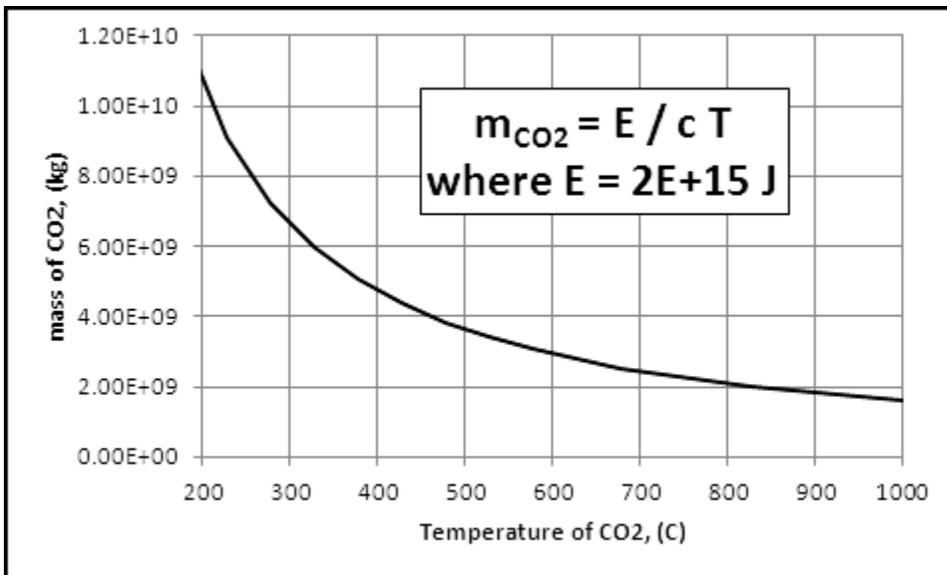
The minimum necessary mass of CO<sub>2</sub> at 1000 °C to melt the ice and heat the meltwater is found to be approximately  $1.7 \times 10^9 \text{ kg}$ , approximately  $1.7 \times 10^6 \text{ tonnes}$ ; CO<sub>2</sub> at a temperature of 200 °C would take approximately  $8.3 \times 10^9 \text{ kg}$ , or  $8.3 \times 10^6 \text{ tonnes}$ ; at 500 °C,  $3.6 \times 10^9 \text{ kg}$ . Whereas the melting event lasted  $5.5 \times 10^7 \text{ s}$ , we further determine the minimum mass flux of

CO<sub>2</sub> at 1000 °C to be 31 kg/s, or 2.7 x 10<sup>3</sup> t/d; CO<sub>2</sub> at 200 °C would require 150 kg/s, or 1.3 x 10<sup>4</sup> t/d; at 500 °C, 66 kg/s or 5.7 x 10<sup>4</sup> t/d.

These mass fluxes,  $Q$ , can alternatively be determined by:

$$Q = H / c_{CO_2} T \quad [4]$$

where  $H$  is the heat output determined in Section 3.2,  $c_{CO_2}$  is the heat capacity of CO<sub>2</sub>, and  $T$  is the temperature.



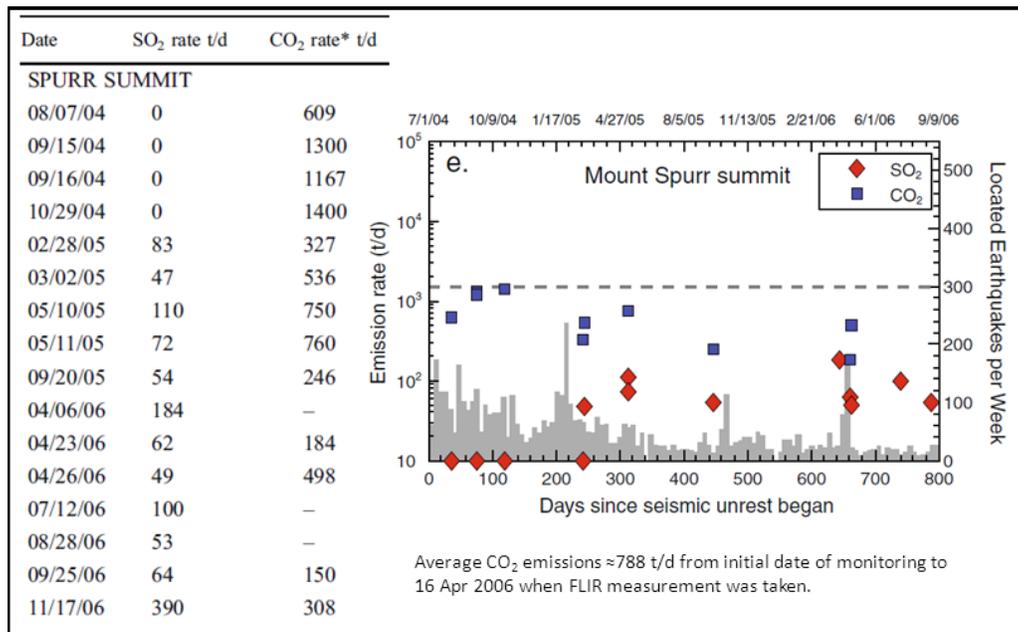
**Figure 6.** Shows the relationship between the temperature of a mass of CO<sub>2</sub> and the amount of mass of CO<sub>2</sub> needed to produce 2x10<sup>15</sup> J of thermal energy, using an expression for the specific heat of CO<sub>2</sub> as a function of temperature (Lemmon *et al.*, 2011).

#### 4.2. Heat Output of CO<sub>2</sub> Derived from Observed Emissions

Prior to the presumed cracking event evidenced by the increased seismicity in Oct 2002, Figure 4, background degassing may be attributed to a residual magma body. The cracking event would then increase the volatile emissions, as suggested in Figure 3.

Volatile emission from Mount Spurr was evidenced on 19 March 2004 by climbers in the area of the as of yet unformed ice cauldron, they stated the presence of steaming rocks below the summit as well as a strong odor of H<sub>2</sub>S, which was later confirmed on 11 July 2004 by a local Alaskan pilot noting a steam plume at ~1650 m and the odor of SO<sub>2</sub> (Coombs *et al.*, 2006).

Since its eruption in 1992, degassing of both SO<sub>2</sub> and CO<sub>2</sub> have continued at background levels ranging from a few tonnes per day of SO<sub>2</sub> and under 200 tonnes per day of CO<sub>2</sub> (Coombs *et al.*, 2006). Observed emissions during the non-eruptive event in this study were monitored at irregular intervals from 08 August 2004 (Doukas and McGee, 2007; Werner *et al.*, 2011) when emission levels significantly increased beyond background emissions, until they again subsided in November 2006. Tabulated results from (Doukas and McGee, 2007) were graphed by (Werner *et al.*, 2011) and are shown in Figure 7.



**Figure 7.** An excerpt from Table 1 of (Werner *et al.*, 2011) showing the observed degassing rates of CO<sub>2</sub> and SO<sub>2</sub> from Mount Spurr between the dates of 7 August 2004 and 17 November 2006 and the associated graph of that data. From the initial date of monitoring to 16 Apr 2006, the date of the AVO FLIR measurement noting the meltwater lake temperature at an average of 11 °C, the average CO<sub>2</sub> emissions were 788 t/d. modified from (Werner *et al.*, 2011).

The following calculations are conservative as they include the assumptions that: (1) the observed CO<sub>2</sub> consisted of the total degassed CO<sub>2</sub>, which is unlikely, and (2) all of the thermal energy from the CO<sub>2</sub> is 100% efficient in rising from the magma intrusion to the base of the ice with no pathway losses of thermal energy. Some of the losses incurred are due to the initial emissions of CO<sub>2</sub> transferring some of its thermal energy to the surrounding rock and hydrothermal fluid, further supporting that CO<sub>2</sub> degassing could not have been the sole mechanism of heat transport to the surface, or that CO<sub>2</sub> degassing was initially much higher than that recorded later in the sequence of volcanic unrest.

The average of the observed mass flow rate,  $Q$ , of CO<sub>2</sub> from 07 Aug 2004 to 14 Apr 2006 is calculated to be 788 tonnes per day, approximately 9 kg/s. Heat output,  $H$ , of the CO<sub>2</sub> emissions can be calculated by:

$$H = c_{CO_2} Q T \quad [5]$$

where  $c_{CO_2}$  is the specific heat capacity of CO<sub>2</sub>, and  $T$  corresponds to an assumed magmatic temperature of 500 °C. The resulting heat output of approximately 5 MW, is about 14% of the 38 MW necessary to form the ice cauldron and increase the meltwater temperature to 11 °C during its 21 month duration. Using a CO<sub>2</sub> temperature of 200 °C would result in a thermal output of approximately 2.2 MW, approximately 6% of that needed.

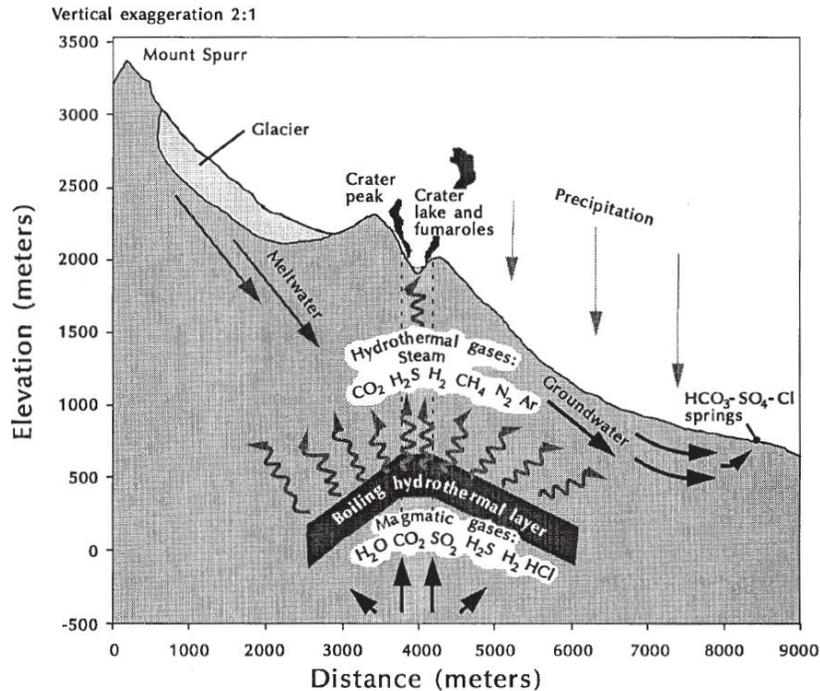
Scaling backward, if this same mass flux at 500 °C had begun a year earlier, effectively doubling the heat output, it would still only provide 10% of that necessary to melt the ice cauldron and heat the meltwater to 11 °C. It is only if we assume that these same emissions began in July 1990, which predates the previous eruption of 1992, that the heat output by CO<sub>2</sub> could account for all that is required. This result suggests that initially, CO<sub>2</sub> output must have

been greater than that observed between 2004 and 2006. It is also likely that the initial CO<sub>2</sub> output did not melt the ice directly, but rather warmed the underlying rock and hydrothermal fluid leading to enhanced hydrothermal heat output. Using the maximum recorded value of 1400 t/d on 29 Oct 2004 results in a value that is 25% for CO<sub>2</sub> at 500 °C, 11 % at 200 °C. This analysis suggests that CO<sub>2</sub> emissions cannot be the sole mechanism of heat transfer if degassing coincided with a seismic swarm around August 2004.

## **5. Characterization of a CO<sub>2</sub> Enhanced Hydrothermal System**

It has been shown in the previous section that degassed CO<sub>2</sub> alone is most probably not the sole provider of thermal energy to the surface of Mount Spurr.

Rather than assume that hydrothermal fluid must move from the top of the newly emplaced magma to the base of the forming ice cauldron, I assume that the base of the hydrothermal system lies much higher in the volcanic edifice. Using controlled-source audio-magnetotelluric, self-potential, and Hg and He soil-gas surveys from Turner and Wescott (1986), Motyka and Nye (1993) constructed a conceptual model of a hydrothermal system beneath Crater Peak, at a depth ranging between 2000 and 2500 m below the summit, and as shallow as 1000 m below the surface along the flank, Figure 8. Power *et al.* (1998) suggests that the hydrothermal system may extend to 4000 m beneath the summit, but here we use the more conservative value of 2000 m. Such a hydrothermal system may not be directly connected to a magmatic heat source at depth. Rather, a preexisting hydrothermal system may be rejuvenated by enhanced volatile flux stemming from a cracking event above a cooling magma at some unknown depth beneath the volcano or from the emplacement and cooling of a newly emplaced batch of magma (Motyka and Nye, 1993).



**Figure 8.** “Conceptual model of the Crater Peak hydrothermal system. Residual 1953 magma at unknown depth expels steam, CO<sub>2</sub>, SO<sub>2</sub>, HCl and other magmatic gases into overlying hydrothermal layer that was formed by post-1953 infiltration and accumulation of ground water. Steam and gases rich in CO<sub>2</sub> and H<sub>2</sub>S are driven off the boiling hydrothermal layer and feed fumaroles at Crater Peak summit crater and interact with shallow groundwaters to produce HCO<sub>3</sub>-SO<sub>4</sub>-Cl thermal springs at lower elevations.” from Figure 4 of (Motyka and Nye, 1993)

In this case, volatile release resulting from a cracking event at the brittle-ductile transition may enhance hydrothermal heat output beneath the ice cover.

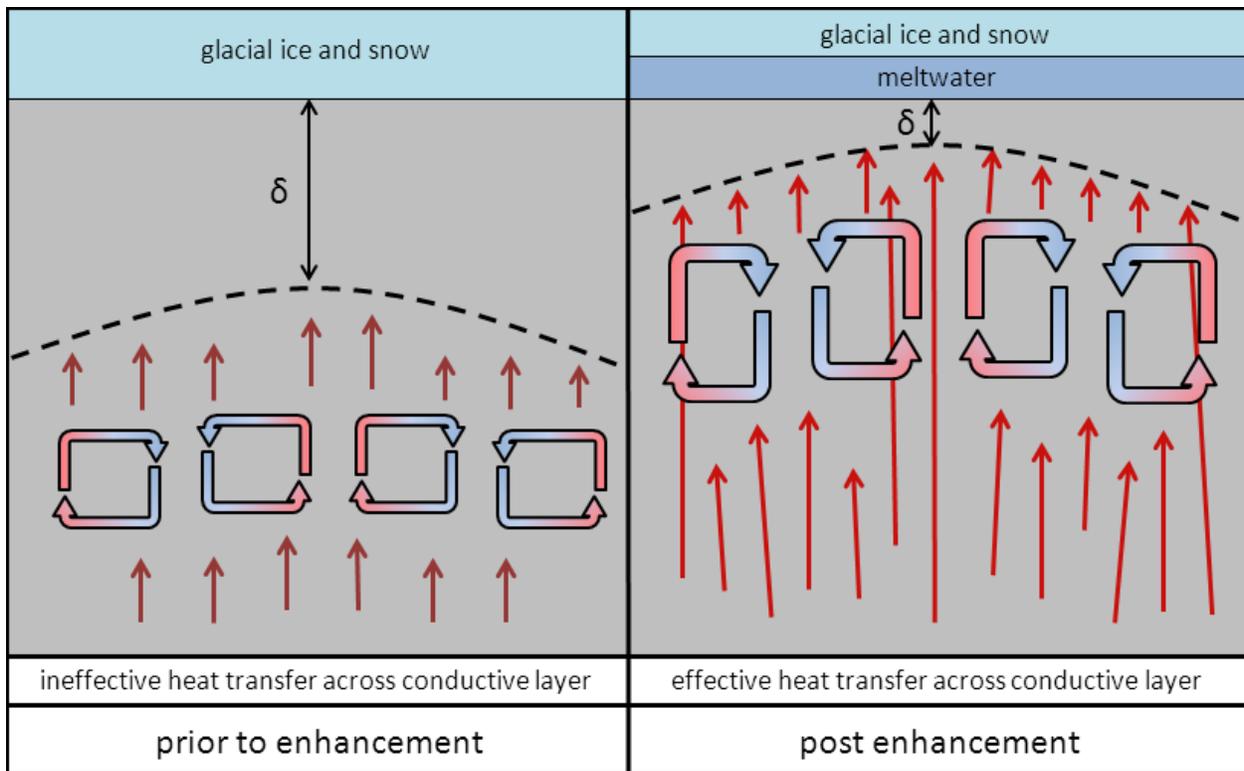
In this section we show how a CO<sub>2</sub>-enhanced hydrothermal system, circulating higher up in the volcanic edifice, could provide additional thermal energy to the ice at the summit of Mount Spurr.

### 5.1. Modeling the Conductive Layer

Consider a hydrothermal system existing in a quasi-steady state that was then perturbed by the degassing of volatiles from below (Power *et al.*, 2004), see Figure 9.

Prior to the addition of heat by newly released volatiles, hydrothermal heat was transferred across a conductive layer,  $\delta$ , such that hydrothermal heat transfer did not result in melting ice on the surface. As a result of increased  $\text{CO}_2$  flux from below, the newly invigorated system then provides more heat, thus diminishing the conductive layer allowing for a more rapid transport of heat into the ice. Alternatively, increased  $\text{CO}_2$  output may lead to fracturing and increased permeability of the hydrothermal system. The minimum, initial conductive layer thickness,  $\delta_i$ , is then shown to be equal to the ratio of the final Rayleigh number,  $Ra_f$ , to the initial Rayleigh number,  $Ra_i$ , multiplied by the final thickness of the conductive layer,  $\delta_f$ :

$$\delta_i = \delta_f \frac{Ra_f}{Ra_i} \quad [6]$$



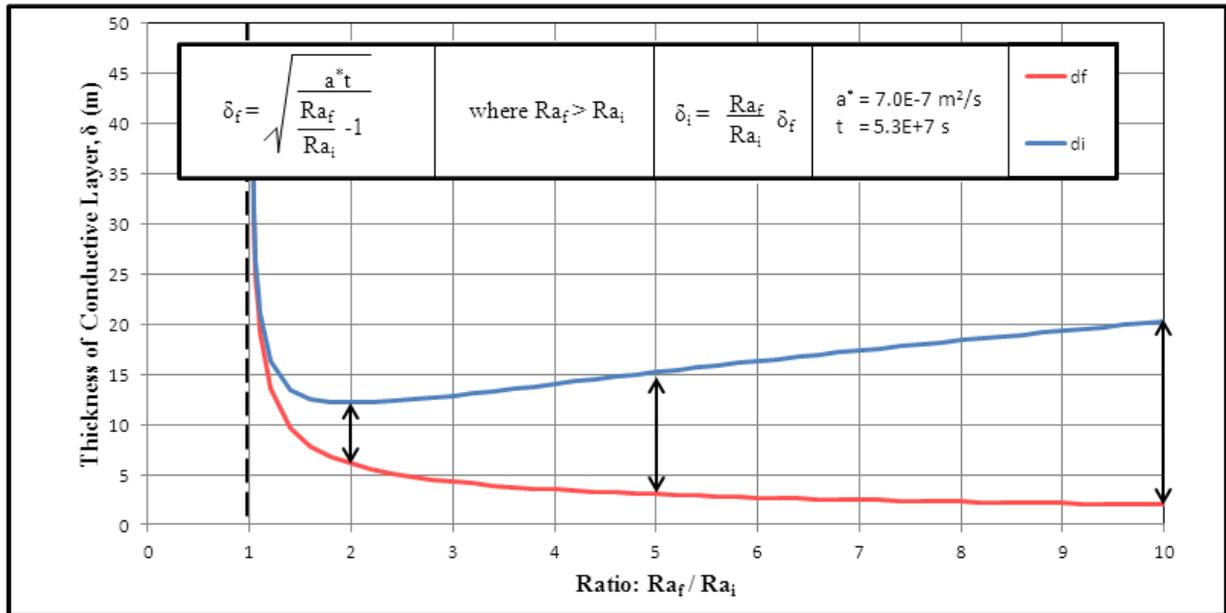
**Figure 9.** Left, the conductive layer  $\delta$  is sufficiently thick enough so as heat cannot effectively be conducted across the boundary. Right, increased heat input from below has invigorated the hydrothermal system, increasing its convective vigor and thus its characteristic Rayleigh number, such that the conductive layer has been sufficiently decreased so as to allow effective conductive heat transport.

Furthermore, it can be shown (see Appendix B) that for an increase in Rayleigh number, the final conductive layer thickness is related to an increase in Rayleigh number by the equation:

$$\delta_f = \sqrt{\frac{a^* t}{\left[\frac{Ra_f}{Ra_i} - 1\right]}} \quad [7]$$

where  $a^*$  is the effective thermal diffusivity of the system, and  $t$  is the  $5.3 \times 10^7$  s duration from the start of the thermal impulse to the time that impulse reaches the surface, calculated in Section 3.3. The underlying assumption in deriving Equation [7] is that the time,  $t$ , is the time needed to move the boundary layer from its initial to its final state.

Figure 10 shows that conductive layers with Rayleigh number ratios less than 1.5, the values of  $\delta_i$  and  $\delta_f$  are virtually indistinguishable, consequently we consider cases for Rayleigh number ratios between 2 and 10.



**Figure 10.** A plot of the initial conductive layer minimum and the final conductive layer maximum with respect to the ratio of an invigorated system's higher final Rayleigh number to that system's less invigorated initial Rayleigh number.

Assuming that the initial Rayleigh number is associated with a hydrothermal fluid of approximately 200 °C (Garchar *et al.*, 2012; Motyka and Nye, 1993) an increase in  $Ra$  could be associated with an increase in hydrothermal temperature associated with enhanced CO<sub>2</sub> release. Because both thermal expansion coefficient and viscosity are strong functions of temperature, only a modest increase in hydrothermal temperature would result in doubling  $Ra$ . Mechanical fracture of the edifice could also significantly increase permeability (Bredehoeft and Ingebritsen, 1990). The constraining factors of Equation [6] are the effective diffusivity, which does not vary significantly over the temperature range of the hydrothermal system, and the duration of the thermal state change,  $t$ , from increased subsurface heating to surface melt.

Given the  $5.3 \times 10^7$  s duration of the thermal impulse event,  $t$ , calculated from Section 3.3, and by the effective thermal diffusivity,  $a^*$ , the final thickness of the conductive layer,  $\delta_f$ , of a doubled Rayleigh number would then be approximately 6 m and the initial thickness of the conductive layer 12 m. Increasing the temperature to 400 °C, for example, would also lead to an increase in  $\alpha$  and a decrease in  $\nu$ , resulting in an increase in  $Ra$  of at least a factor of five. This would result in  $\delta_f = 3$  m and  $\delta_i = 15$  m, respectively. The additional release of CO<sub>2</sub> could also lead to cracking and enhanced permeability of the hydrothermal system (Bredehoeft and Ingebritsen, 1990; Ingebritsen and Manning, 2010; Koulakov *et al.*, 2013). A tenfold increase in Rayleigh number results in  $\delta_f = 2$  m and  $\delta_i = 20$  m.

The planform area,  $A$ , of the hydrothermal system can be approximated by:

$$A = \frac{H \delta}{\lambda \Delta T} \quad [8]$$

where  $H$  is the 38 MW minimum heat output of the ice melt and meltwater heating event;  $\delta$ , the thickness of the final conductive boundary layer;  $\lambda$ , the thermal conductivity of andesite; and  $\Delta T$ , the temperature difference between the top and bottom of the hydrothermal system, which yields

a minimum area of approximately  $4 \times 10^5 \text{ m}^2$ . By using the relationship of an area of a circle to that circle's radius, we obtain a radius of approximately 357 m. Neal *et al.* (2007) estimate the radius to be 300 m, which is in good agreement with this result.

One can also estimate the radius of the surface expression of the hydrothermal system by investigating the area of fumarolic activity from the GIS map of Mt. Spurr (Coombs *et al.*, 2006) shown in Figure 11. By finding the centroid of the fumarolic activity, a best fit circle can then be estimated and a radius measured, which in this instance is approximately 410 m, an area of  $5.3 \times 10^5 \text{ m}^2$ , a 14% difference in radius resulting in a 28% difference in area with that area calculated in Equation [8].

## 5.2. Permeability of the Volcanic Edifice

In a vigorously convecting hydrothermal system, the relationship between the thickness of the conductive boundary layer,  $\delta$ , velocity of fluid transport,  $u$ , and the effective thermal diffusivity,  $a^*$ , can be scaled by the relationship (Bejan, 1995):

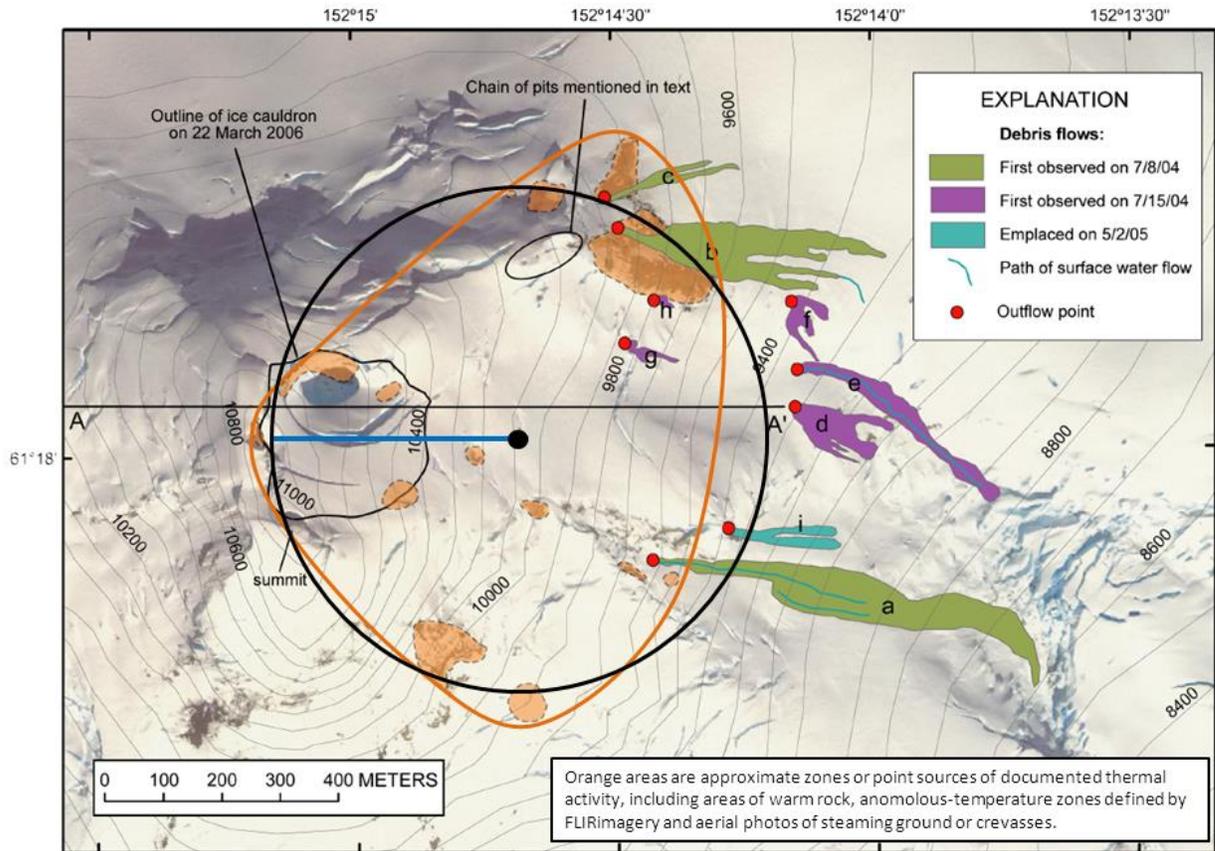
$$\delta = \frac{a^*}{u} \quad [9]$$

Permeability,  $k$ , can be calculated by solving Equation [9] for velocity,  $u$ , substituting into Equation [13], then rearranging to solve for  $k$ :

$$k = \frac{\nu a^*}{\alpha g \Delta T \delta} \quad [10]$$

where  $\nu$  is the kinematic viscosity of the fluid;  $a^*$  is the effective thermal diffusivity where  $a^* = \lambda / (\rho_w c_w)$ ,  $\lambda$  is the thermal conductivity of andesite,  $\rho_w$  is the density of water, and  $c_w$  is the heat capacity of water;  $\alpha$  is the thermal expansion coefficient of the hydrothermal fluid;  $g$  is acceleration due to gravity;  $k$  is the permeability;  $\Delta T$  is the change in temperature of the hydrothermal system of 200 °C; and  $\delta_i$  is the initial thickness of the conductive layer of 12 m as

calculated in Equation [7]. The mean permeability is shown to be approximately  $\sim 10^{-14}$ , which is more expected and more acceptably within the range of  $\sim 10^{-11}$  to  $\sim 10^{-15}$  m<sup>2</sup> for fractured igneous rock (Freeze and Cherry, 1979).



**Figure 11.** The orange outline roughly delineates the area of fumarolic activity, and potentially the area of the hydrothermal system’s planform surface expression. The black dot marks the centroid of the irregular planform expression, and the black circle an estimated circle of equal area to the irregularly shaped planform area. The blue line marks the radius of that circle. Modified from (Coombs *et al.*, 2006) .

### 5.3. Additional Investigation of CO<sub>2</sub> Release

As a magma body cools, CO<sub>2</sub> and other volatiles are degassed and travel upward through the permeable medium, transferring heat to the rock and hydrothermal fluids. If the hydrothermal

fluid is under-saturated, the volatiles will dissolve into the fluid and reach thermal equilibrium with the rock and fluid surroundings. Once saturation occurs, a continuing flux of volatiles will pass through the circulating hydrothermal fluid, transfer some heat to the fluid and adjacent rock, and escape to the surface. One effect of the volatile transfer is to heat the hydrothermal system, resulting in an increased Rayleigh number and a thinning of the conductive boundary layer as described in Section 5.1. Here, I estimate the amount of CO<sub>2</sub> needed to heat the hydrothermal system and adjacent rock within the Mt. Spurr volcanic edifice.

The heat energy,  $E$ , contained within a mass,  $m$ , of CO<sub>2</sub> at an assumed magmatic temperature  $T_m = 1000$  °C, can be found by a rearrangement of Equation [3]:

$$E = m c_{co2} T_m \quad [11]$$

where  $c_{co2}$  is the heat capacity of CO<sub>2</sub> appropriate for the temperature. The mass,  $m$ , needed to raise the temperature of the rock and fluid in the hydrothermal system by an amount  $\Delta T$  is then given by:

$$m = V \Delta T [ \varphi \rho_w c_w + (1-\varphi) \rho_a c_a ] / c_{co2} T_m \quad [12]$$

where  $V$  is the volume to be heated and  $\varphi$  is the porosity. This expression assumes that the hydrothermal fluid remains in the liquid phase, and that the CO<sub>2</sub> flux does not heat the rock beneath the hydrothermal system. Assuming a constant porosity of 0.10 and a cylindrical volume 2000 m high and a basal area of  $4 \times 10^5$  m<sup>2</sup> as calculated in Equation [8] and  $\Delta T = 100$  °C, the minimum required mass  $m$  of CO<sub>2</sub> is  $2.5 \times 10^{11}$  kg is required to heat the hydrothermal system as described to deliver  $1.2 \times 10^{18}$  J of thermal energy.

## 6. Discussion

### *6.1. Uncertainties in the Analysis*

In this paper, I have used a variety of data sets associated with the non-eruptive volcanic event at Mt. Spurr Alaska between 2002 and 2006 to assess the roles of CO<sub>2</sub> flux and enhanced hydrothermal activity as mechanisms for driving the observed formation of the ice cauldron and heating of the melt water. Although the data provide a detailed timeline of events beginning with increased seismic activity in 2002 and culminating with the cessation of the event in 2006, a number of uncertainties prohibit a unique assessment of the driving mechanisms.

The first of these uncertainties would be that we assume the ice melted and heated the lake to only 11 °C, but it may have been greater, and then cooled to 11 °C. Once the cauldron had been formed, there was continued fumarolic activity and likely heating of the surrounding rock, this potential additional heat lost to the environment was not accounted for. Hence the amount of heat needed to describe the thermal event, whether by volatiles alone or by a combination of volatile release and hydrothermal activity is somewhat uncertain. Geochemical arguments suggest that the base of the hydrothermal system may have begun at 200 °C, but the final temperature of the system is an unknown, and therefore the amount of increased heat to be provided by CO<sub>2</sub> is also an unknown. It is not fully clear as to whether the outflow of the debris flows exactly matches the volume lost in lake level, possibly there was sub-glacial ice melted as meltwater transferred heat to the underside of glaciers as it traveled downhill.

There is little constraint on the physical dimensions and makeup of the hydrothermal system. The actual thickness, depth, and planform area of the hydrothermal system is not measurably quantified and the chemical composition of the hydrothermal fluid is unknown. As stated

previously, there is no direct temperature data for the hydrothermal system. It is these factors we use to determine such values as Rayleigh number and permeability. However, the calculations seem reasonable within the constraints of the available data.

Although there has been a suggested range of permeability for a volcanic regime, there is no actual quantifying data to suggest the actual permeability within the edifice of Mt. Spurr. Permeability exerts the primary control on convective vigor, and it seems likely that changes in permeability resulting from pressurized volatiles enhanced the permeability of the pre-existing hydrothermal system.

As a result of these uncertainties, I used lower estimates for the amount of thermal energy released in the event and this lower estimate is used to estimate both the required CO<sub>2</sub> flux and the hydrothermal vigor. Despite these uncertainties, however, I think a number of important conclusions can be reached concerning this event.

## ***6.2. Implications of the Results***

Despite the uncertainties the in analyses performed here, one can make inferences regarding the potential for predicting magmatic unrest events at Mt. Spurr. Based on the conceptual model used here, the occurrence of an unrest event is linked to exsolution of magmatic volatiles that pond beneath the brittle-ductile transition and advectively transport thermal and mechanical energy to the surface when that impermeable barrier is broken. If one knew the composition of the magmatic fluids, the strength of the brittle-ductile transition, and the mechanisms that lead to rupture, a time frame could be determined as to the potential for subsequent events of unrest. Such data are not available for Mt. Spurr, and may be difficult to know in advance at any volcano; however, increased seismicity concentrated at a depth near the brittle-ductile transition

likely signal breach of that impermeable barrier and the initiation of a volcanic event. For the unrest event at Mt. Spurr between 2002 and 2006, there was a period of more than a year between the initiation of seismicity in October 2002 and the surface expression of ice melt and CO<sub>2</sub> flux to the atmosphere that began in June/July 2004. The delay time between rupture of the brittle-ductile transition and release of volatiles most likely corresponds to (a) the time for the pressure perturbation to propagate from the brittle-ductile transition to the base of the hydrothermal system, and (b) the time for CO<sub>2</sub> to migrate through the volcanic edifice to the surface. The time scale  $\tau_p$  for diffusion of the pressure pulse that occurs as a result of breaking the brittle ductile seal and changing the pressure from lithostatic to hydrostatic can be estimated from the expression:

$$\tau_p \sim L^2/a_p \quad [13]$$

where  $L$  is the approximate  $6 \times 10^3$  m distance between the brittle ductile transition and the base of the hydrothermal system and  $a_p = k c_l^2 / \nu \phi$  is the coefficient of pressure diffusivity (Phillips, 1991). In this expression,  $k$  is the permeability of  $\sim 10^{-14}$  as calculated in Equation [10],  $c_l \sim 1.4 \times 10^3$  m/s is the speed of sound in water,  $\nu$  is the kinematic viscosity of the fluid as defined in Appendix A, and  $\phi$  is the assumed 0.1 porosity of the host medium. The pressure diffusivity is then  $\sim 0.65$  m<sup>2</sup>/s, which yields a pressure diffusion time scale of  $\sim 5.5 \times 10^7$  s, for a travel distance of 6 km. The permeability at depth is likely less than that estimated for the hydrothermal system in the upper part Mt. Spurr; however, this would be offset by a decrease in porosity as well as a decrease in fluid viscosity that would accompany an increase in temperature at depth.

The pressure pulse along with CO<sub>2</sub> transport at a temperature of approximately 500 °C, the approximate temperature of the brittle-ductile transition, likely contribute together to enhancing

the Rayleigh number, and hence heat transport, of the hydrothermal system circulating within the volcanic edifice. It appears from our calculations that it is enhanced heat transport from the hydrothermal system that is mainly responsible for the ice melting event at Mt. Spurr summit rather than directed heating by the observed CO<sub>2</sub> flux. Hence understanding the time delay between cracking the permeability seal that can be seen from seismicity data and the response of the hydrothermal system to this event can assist in planning for evacuation and/or diversion of people and cargo from the area.

## **7. Conclusions**

The results of the analyses presented here indicate that the observed amount of CO<sub>2</sub> flux cannot provide the heat necessary to form the ice cauldron and heat the resulting meltwater, therefore, CO<sub>2</sub> emission cannot be the sole mechanism of heat transport.

A pre-existing hydrothermal system enhanced by CO<sub>2</sub> emission does have the capability to transport the required amount of heat in the required amount of time, provided that CO<sub>2</sub> can increase the Rayleigh number of the hydrothermal system by a factor of ten. This invigoration more likely results from increased permeability caused by the episodic release of pressurized CO<sub>2</sub> than by an increase in  $\Delta T$  across the hydrothermal system. Based on the heat needed from hydrothermal activity, the permeability in the upper 2 kilometers of Mt Spurr is  $\sim 10^{-14} \text{ m}^2$ .

The increase in seismic activity leading up to and including the event of unrest is more likely caused by pressurized volatiles cracking less permeable strata and making their escape, leading to an increase of observed volatile flux and subsequent heating, rather than the emplacement of a new magma body.

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## Appendix A: Symbol Definitions

Physical Meaning	Symbol	Value	Units
<b>Latin Symbols</b>			
Coefficient of pressure diffusivity	$A$	$\sim 2$	m/s
Effective diffusivity	$a^*$	$7 \times 10^{-7}$	$\text{m}^2/\text{s}$
Area	$A$		$\text{m}^2$
Speed of sound in water	$c_l$	$1.4 \times 10^3$	m/s
Specific heat capacity of ice	$c_i$	$2.11 \times 10^3$	J/kg °C
Specific heat capacity of andesite	$c_a$	$1.4 \times 10^3$	J/kg °C
Specific heat capacity of CO <sub>2</sub> 200 to 1000 °C	$c_{CO_2}$	$1.2 \times 10^3$	J/kg °C
Specific heat capacity of water	$c_w$	$4.2 \times 10^3$	J/kg °C
Thermal Energy	$E$		J
Acceleration due to gravity	$G$	9.8	$\text{m}/\text{s}^2$
Thickness of a hydrothermal system	$H$		m
Heat output	$H$		W
Permeability	$K$		$\text{m}^2$
Latent heat of fusion, H <sub>2</sub> O	$L_{wi}$	$3.4 \times 10^5$	J/kg
Length from impermeable barrier to hydrothermal sys.	$L$		m
Mass	$M$		kg
Pressure	$P$		Pa or bar
Mass flux	$Q$		kg/s or t/d
Rayleigh number	$Ra$		dimensionless
Rayleigh number, final	$Ra_f$		dimensionless
Rayleigh number, initial	$Ra_i$		dimensionless
A change in temperature	$\Delta T$		°C
Temperature, subscripts $f$ is final, $i$ is initial	$T$		°C
Temperature of fusion, H <sub>2</sub> O	$T_{fus}$	0	°C
Initial temperature of magma	$T_m$	1000	°C
Duration of event	$T$		s
Velocity of heat transport	$U$		m/s
Volume of melted ice, (Coombs <i>et al.</i> , 2006)	$V_i$	$\sim 5.4 \times 10^6$	$\text{m}^3$
<b>Greek Symbols</b>			
Thermal expansion coefficient, hydrothermal fluid	$\alpha$	$1.0 \times 10^{-3}$	1/°C
Thickness of thermal conduction layer, final	$\delta_f$		m
Thickness of thermal conduction layer, initial	$\delta_i$		m
Thermal conductivity, andesite	$\Lambda$	2.9	W/m °C
Kinematic viscosity of hydrothermal fluid	$\nu$	$3.0 \times 10^{-7}$	$\text{m}^2/\text{s}$
Density of Andesite	$\rho_a$	$2.7 \times 10^3$	$\text{kg}/\text{m}^3$
Density of H <sub>2</sub> O, scale analysis value	$\rho_w$	$1.0 \times 10^3$	$\text{kg}/\text{m}^3$
Time scale of diffusion	$\tau_p$		s
Porosity	$\Phi$		dimensionless

## Appendix B: A Derivation of a Boundary Layer Relationship to a Perturbation in Rayleigh Number

A steady-state hydrothermal system reacts to a perturbation from below. In this case, magma is emplaced below a hydrothermal system; outgassed volatiles from the emplaced magma provide additional heat to the base of a hydrothermal system, altering that hydrothermal system's behavior.

A hydrothermal system state can be described by its Rayleigh number,  $Ra$ :

$$Ra = \frac{\alpha g k T h}{\nu \alpha^*} \quad [B1]$$

which indicates the vigor with which it convects. When the factors that determine the Rayleigh change, that change is reflected in the Rayleigh number and may be expressed as a ratio of the final Rayleigh number to the initial Rayleigh number.

$$\frac{Ra_f}{Ra_i} \quad [B2]$$

Any heat output by a volcanic hydrothermal system must then travel through a thermal boundary layer before being sensed at the surface. A hydrothermal system without proper vigor may be unable to transport heat at a rate sufficient enough to be sensed at the surface, such as in the case of melting glacial ice and snow on the volcanic surface.

However, changes in those factors which increase the Rayleigh number will increase the vigor with which the hydrothermal system convects, and in some cases be raised enough to transport heat across that thermal boundary layer to the surface. This thermal boundary layer is denoted as  $\delta$ , which is related to the Rayleigh number by the scaling relationship:

$$\delta = \frac{h}{Ra} \quad [B3]$$

where it is shown that the boundary layer,  $\delta$ , has an inverse relationship with  $Ra$ .

$$\delta \propto \frac{1}{Ra} \quad [B4]$$

An increase in Rayleigh number,  $Ra$ , expressed as a ratio of the final Rayleigh number,  $Ra_f$ , to the initial Rayleigh number,  $Ra_i$ , is then the inverse ratio of that which the final boundary layer,  $\delta_f$ , is to the initial boundary layer,  $\delta_i$ .

$$\frac{Ra_f}{Ra_i} = \frac{\delta_i}{\delta_f} \rightarrow \frac{Ra_f}{Ra_i} \delta_f = \delta_i \quad [B5]$$

The time lapse,  $t$ , is determined by the time change from the initiation of the thermal impulse, taken in this case to be the date of the initial seismic swarm, and the date which the thermal impulse was felt on the surface, denoted by the melting of ice.

Since we are interested in the change in boundary layer thickness,  $\Delta\delta$ , as it diminishes from its larger initial distance,  $\delta_i$ , to its shorter final distance,  $\delta_f$ , I assume that the time  $t$  needed to move the boundary a distance  $\Delta\delta$  is given by

$$t = \frac{\delta_i - \delta_f}{u_f} \quad [B6]$$

The final velocity,  $u_f$ , is related to the final boundary layer,  $\delta_f$ , and the effective diffusivity of the system,  $a^*$ , through scale analysis:

$$u_f = \frac{a^*}{\delta_f} \quad [B7]$$

Substituting [B7] into [B6] yields:

$$t = \frac{\delta_f}{a^*} (\delta_i - \delta_f) \quad [B8]$$

Multiply both sides by  $a^*$ :

$$a^* t = \delta_f (\delta_i - \delta_f) \quad [B9]$$

Substitute [B5] into [B9] for  $\delta_i$ :

$$a^* t = \delta_f \left( \frac{Ra_f}{Ra_i} \delta_f - \delta_f \right) \quad [B10]$$

Factor out  $\delta_f$  from the right side of the equation:

$$a^*t = \delta_f^2 \left( \frac{Ra_f}{Ra_i} - 1 \right) \quad [\text{B11}]$$

Divide both sides by:  $\left( \frac{Ra_f}{Ra_i} - 1 \right)$

$$\frac{a^*t}{\left( \frac{Ra_f}{Ra_i} - 1 \right)} = \delta_f^2 \quad [\text{B12}]$$

Take the square root of both sides:

$$\sqrt{\frac{a^*t}{\left( \frac{Ra_f}{Ra_i} - 1 \right)}} = \delta_f \quad \blacksquare \quad [\text{B13}]$$