

**Monitoring Pesticides in the Groundwater and Submarine  
Groundwater Discharge of the Eastern Shore of Virginia.**

by

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MONITORING PESTICIDES IN THE GROUNDWATER AND THE SUBMARINE  
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(ABSTRACT)

The purpose of the research was to determine if pesticides were being transported from the place of application by the shallow groundwater and discharged into the Chesapeake Bay and Atlantic Ocean, and quantify the pesticides if they were transported. One reference (undeveloped) and four agricultural sites were tested over a 11 month period from April 1992 to February 1993. Over 500 groundwater samples were analyzed from both shallow wells and seepage meters placed in the Chesapeake Bay and Magothy Bay. The samples were analyzed in accordance to EPA Method 525.1 by solid phase extraction with octadecyl bonded disks followed by gas chromatography. The samples were examined for 5 of the most commonly used pesticides: atrazine, alachlor, carbofuran, cyanazine, and metolachlor. Pesticides were detected in only 16 samples. All the detections were at low concentrations, with only one being over 1  $\mu\text{g/L}$ . The study concluded that if pesticides were being transported by the groundwater, they were below a  $\mu\text{g/L}$  (ppb).

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## I. Introduction

Since prehistoric times humans have cultivated crops for a steady food source. With the advent of man-made compounds and intensive farming techniques, crop yields have steadily improved. One of the reasons for the improvement was the widespread use of pesticides. Pesticides are divided into two main categories; insecticides and herbicides. Insecticides are used to kill a wide variety of insects and nematodes. Herbicides are used against grasses, molds, and fungi that attack or compete with the crop.

Gradually, people realized that pesticides could effect humans as well. The discovery of DDT's bioaccumulation and long half-life in the environment led to a greater public concern and awareness. Testing by the EPA showed that two of the pesticides in this study, alachlor and atrazine, are possible carcinogens. The maximum contaminate levels (MCL) in drinking water set by the EPA were 3  $\mu\text{g}/\text{L}$  for atrazine and 2  $\mu\text{g}/\text{L}$  for alachlor. The long term affect of many pesticides on humans is not well known.

Studies have confirmed that some groundwater sources have been contaminated by synthetic organic compounds, including pesticides (Domagaski and Dubrovsky, 1992; Holden, et al., 1992; Krill and Sonzogni, 1986; Leonard, et al., 1988; Pionke and Glotfelty, 1989).

Recently, researchers have demonstrated that submarine

groundwater discharge (SGWD), which is the discharge of groundwater below the surface of a body of water, such as an ocean, accounts for a much larger fraction of the hydrological cycle than previously thought (Simmons, et al., 1992). This raised the question of whether pesticides were being transported by SGWD into bays and oceans. In response, this study was undertaken with two main objectives:

1. To determine if pesticides were being transported by the groundwater from the fields they were applied and into the Chesapeake Bay and Magothy Bay.

2. To quantify the amount pesticide transport, if any occurred.

The study also analyzed soil samples, and an independent laboratory, Ohmicron Corporation, analyzed duplicates of most of the samples. The soil samples were included to determine if there was a correlation between the soil and groundwater concentrations of pesticides, while the Ohmicron samples were used help confirm if any pesticides were in the water samples.

## **II. Literature Review**

### **A. Introduction**

The study tested the groundwater and SGWD of the Eastern Shore of Virginia for five pesticides; atrazine, alachlor, carbofuran, cyanazine, and metolachlor. The pesticides were selected by the following criteria: they were commonly used on the eastern shore of Virginia (Hohlt, 1993); they were highly susceptible to leaching into the groundwater, and they had been detected in other groundwater studies in the United States.

### **B. Submarine Groundwater Discharge**

In general, unconfined coastal aquifers are characterized by flow in the direction of adjacent surface waters. These flow patterns are in response to elevated hydraulic heads within upland regions. The advective movement of groundwater into aquatic systems is termed submarine groundwater discharge (SGWD). As fresh groundwater moves into nearshore coastal environments, flow paths are sharply deflected upwards by salinity density gradients. Consequently, most unconfined groundwater discharge occurs immediately adjacent to the land-water interface and represents a mixture of fresh groundwater and interstitial seawater (Bear, 1979).

A consideration of SGWD as a contaminant transport mechanism is relatively recent. Traditional methods to estimate solute flux across the sediment-water interface only

considered diffusion and assumed advective transport mechanisms to be negligible. However, evidence now suggests that advective transport mechanisms of solutes are significant in lake, estuarine, and marine sediments (Lee, 1977; Lee, et al., 1980; Sayles and Jenkins, 1982; Zimmermann, et al., 1985; Cornett, et al., 1989; Simmons, 1989; Reay and Gallagher, 1991; Simmons, et al., 1991). Groundwater discharge has been implicated in nitrogen enrichment of coastal systems (Sewell, 1982; Johannes and Hearn, 1985; Lee and Olsen, 1985; Valiela, et al., 1990). Although significant effort has been invested in studying surface and groundwater quality in the Chesapeake Bay watershed, it has only been recently that research has focused on the significance of groundwater discharge to the Bay system (Simmons, 1989; Libelo, et al., 1991). Research has also shown that transport of fertilizer-related nutrients by SGWD significantly influence water column nutrient concentrations in Virginia's coastal waters (Reay, et al., 1992).

### **C. History of the Pesticides**

Each pesticide's history and properties are given in the following sections (also see Table 1).

#### **1. Alachlor**

Monsanto patented alachlor (2-chloro-2',6'-diethyl-N-(methoxymethyl) acetamide) in 1967. It has an acute lethal dose in 50% of the test animals (LD<sub>50</sub>) of 1200 mg/kg of body

weight (Windhole, 1983). Alachlor is classified as a herbicide, and used to control grasses and broadleaf weeds in corn, soybeans, and peanuts (Holden, et al., 1992). Its solubility in water is 203 mg/L (Gustafson, 1989).

## 2. Atrazine

Atrazine (2-chloro-4-ethylamino-6-isopropylamino-1,3,5-triazine) was patented by Giegy in 1960. Its LD<sub>50</sub> is 1750 mg/kg (Windhole, 1983). The water solubility of atrazine is listed as 52 mg/L (Gustafson, 1989). Atrazine is applied as a pre- and post-emergence herbicide. It is one of the most commonly used pesticides in the United States, being applied to corn and soybean fields to control grasses. Heatwole, citing Pait, estimated that approximately 0.65 million Kg of atrazine were applied in the Chesapeake Bay drainage area in 1988 (Heatwole, et al., 1992).

## 3. Carbofuran

Carbofuran (2,3-dihydro-2,2-dimethyl-7-benzofuranol methylcarbamate) is a insecticide/nematicide patented by Bayer in 1964. It is used on a wide variety of crops to control insect infestations. Carbofuran was the only highly toxic pesticide in the study with an LD<sub>50</sub> of 3 mg/kg (Windhole, 1983). It has a water solubility of 700 mg/L (Gustafson, 1989).

## 4. Cyanazine

Degussa patented cyanazine (2-[[4-chloro-6-(ethylamino)-

Table 1  
Studied Pesticides' Properties

Pesticide	LD <sub>50</sub> , mg/kg	Solubility, mg/L	K <sub>ow</sub>
Alachlor	1200	203	2.78
Atrazine	1750	52	2.45
Carbofuran	3	700	2.44
Cyanazine	182	171	2.24
Metolachlor	2780	530	2.81

\* Data from: Domagalski and Dubrovsky, 1992; Gustafson, 1989; Windhole, 1983

S-triazin-2-yl]amino]-2-methylpropanenitrile) in 1968. The listed LD<sub>50</sub> for cyanazine is 182 mg/kg (Windhole, 1983). The solubility in water of cyanazine is 171 mg/L (Gustafson, 1989). Cyanazine is a herbicide, and is used to kill grasses in corn, soybean, and sorghum (Squillace and Thurman, 1992).

#### 5. Metolachlor

Metolachlor (2-chloro-N-(2-ethyl-6-methylphenyl)-N-(2-methoxy-1-methylethyl)acetamide) was patented in 1973 by Ciba-Giegy. It was the least toxic of the pesticides studied with a LD<sub>50</sub> of 2780 mg/kg (Windhole, 1983). Metolachlor is classified as a herbicide, and mainly used to control grasses in corn and soybeans (Bouchard, et al., 1982). Its water solubility is 530 mg/L (Gustafson, 1989).

#### C. Pesticide Transport and Breakdown

All five of the pesticides in the study have been detected in the groundwater of the United States (Domagaski and Dubrovsky, 1992; Holden, et al., 1992; Krill and Sonzogni, 1986; Leonard, et al., 1988; Pionke and Glotfelty, 1989). Atrazine was the most commonly found pesticide in all the cited studies. This was unexpected, since most pesticide transport models predict that carbofuran, with its higher solubility and lower K<sub>oc</sub> would be found in higher concentrations than atrazine (Donigian and Carsel, 1987; Jury, et al., 1987; Shirmohammadi, et al., 1989).



## 1. Pesticides detected in groundwater studies

The following groundwater studies were conducted with one or more of the pesticides in this project. It should be noted that all the studies were looking at high risk areas with shallow water tables and very permeable soils.

In a study of artificially recharged groundwater in Nebraska, Exner (1990) found that the shallow, unconfined aquifer was contaminated with atrazine, alachlor, metolachlor, and cyanazine. Atrazine was found in almost all the samples from 0.3 to 8.8 ppb, while the other 3 pesticides were found in about half the samples at concentrations from 0.1 to 0.9 ppb. Atrazine was also found in the deeper regional aquifer, at concentrations from 0.1 to 2.5 ppb.

Also in Nebraska, a study of 41 wells in a shallow, river influenced aquifer found atrazine in all 108 samples (Wehtje, et al., 1983). The concentrations ranged from 0.01 to 8.29  $\mu\text{g/L}$ , with an average of 0.38  $\mu\text{g/L}$ . Nitrate and atrazine appear to be linked with the highest concentrations of nitrate and atrazine occurring in the same wells.

A third Nebraska study with over 2200 well samples found 13.5% of the samples contained atrazine. The contamination was at very low concentrations with a median of 0.38  $\mu\text{g/L}$ . Alachlor, cyanazine, and metolachlor were also detected in 0.77, 0.57, and 0.50% of the samples, respectively. Like atrazine, the median concentration were very low with

cyanazine being the highest at 0.10  $\mu\text{g/L}$  (Spalding, et al., 1989).

Testing the discharge of Big Spring, a large groundwater spring in Iowa, Libra, et al. (1987) said, "Atrazine is the most widely used herbicide in the basin, and it is the dominant pesticide found in the groundwater." Atrazine was detected at an average concentration of 0.2  $\mu\text{g/L}$  in the base flow of the spring. During discharge events, atrazine, alachlor, cyanazine, and metolachlor were detected at concentrations of 5.1, 0.6, 1.2, and 0.6  $\mu\text{g/L}$ , respectively. Alachlor and cyanazine were also intermittently found in the base flow.

A Wisconsin survey of 349 wells found all five pesticides. Metolachlor was detected in 27% of the samples. Atrazine, alachlor, cyanazine, and carbofuran were also found in 14, 12, 7, and 3% of the samples, respectively (Holden, 1986).

A later Wisconsin study of 358 wells found pesticides in only 7% of the samples. The most commonly detected pesticide was atrazine, which was found in 3% of the samples. Alachlor, metolachlor, and carbofuran were all detected once in separate samples (Krill and Sonzogni, 1986).

However, an Arkansas study of 119 wells found only 1 sample with pesticides detected. Alachlor, cyanazine, and metolachlor were detected at 5.5, 5.8, and 6.9  $\mu\text{g/L}$ . Since

the pesticides were not detected in previous or subsequent samples, the sample was concluded to be a spill or laboratory error (Cavalier, et al., 1989)

In a Kansas study of 103 wells, atrazine was detected in 4 wells, and alachlor was detected in 1 well. The highest concentrations were 40  $\mu\text{g/L}$  for atrazine and 1.8  $\mu\text{g/L}$  for alachlor (Steichen, et al., 1988).

Pionke, et al. (1989) found atrazine in 70% of the wells tested in a Pennsylvania study. The concentrations ranged from 0.013 to 1.11  $\mu\text{g/L}$ . He also found the following correlation between pesticide detection and nitrate and chloride concentrations.

The probability of atrazine being found above the minimum detection limit was high when  $\text{NO}_3\text{-N}$  and Cl concentrations were above 4.0 and 3.0 mg/L, respectively. Also, generally the higher these concentrations, the higher the atrazine concentration.

Holden, et al. (1992) also found a link between the nitrate level and the probability of finding a pesticide in the sample. In the national alachlor well water survey, sponsored by Monsanto, over 1400 wells were sampled for pesticides. Atrazine, alachlor, cyanazine, and metolachlor were detected in 12, 0.8, 0.3, and 1% of the samples, respectively. Almost all the pesticide concentrations were at low levels with only 0.22% of the concentrations being over 1  $\mu\text{g/L}$ .

An upstate New York study of 73 wells and 129 samples found pesticides in 6 samples. Atrazine was detected twice, at 1 and 3  $\mu\text{g/L}$ . Also, a carbofuran metabolite, 3-hydroxy carbofuran, was detected in 2 samples at 2 and 3  $\mu\text{g/L}$  (Walker and Porter, 1990). In the project, metabolites were not considered, only the actual pesticides.

A shallow groundwater study in the Coastal Plains geological region of Georgia, atrazine movement through the soil was tested. Atrazine was detected in the groundwater two months after application, with the highest concentration, 52  $\mu\text{g/L}$ , occurring 150 days after application. Smith, et al. (1990) stated that "Up to 5% of the applied atrazine was calculated to be present within the saturated zone 187 days after application."

In another study of pesticide movement in the Coastal Plains soil in Georgia, atrazine and other pesticides were applied to test plots for a three year period. Atrazine was detected in the second and third year's groundwater samples at depths up to 2.4 meters below the ground's surface (Leonard, et al., 1988). This indicated that atrazine had a months long lag in transport from the surface to the groundwater, and even though its reported half life was as low as 20 days, detectable amounts remained in the soil for over a year.

## 2. Pesticides detected in SGWD

Since contamination of SGWD was a new field of study,

there were few references about it. However, in a study of Big Spring in Iowa, it was estimated that the baseflow concentration of atrazine was 0.2  $\mu\text{g/L}$  in the groundwater seepage (Libra, et al., 1987). Also, in a study of the pesticides in the Cedar River in Minnesota and Iowa, Squillace and Thurman (1992) said,

A minimal estimate of atrazine transported by the groundwater flow component of the river discharge past the Bertram sampling site is 800 kg/year (April 1984 through March 1985), which is ~6% of the total atrazine transported.

### 3. Half lives of the pesticides

The half life of a pesticide in the soil or groundwater depended on many factors including temperature, organic content, soil moisture, clay content, and depth. Increasing any of the factors except depth decreased the half life of the pesticide. Increasing depth increased the half life of the pesticide (Bouchard, et al., 1982; Weber and Peter, 1982). For atrazine the half life was listed as between 20 and 140 days (Loague, et al., 1990; Thurman, et al., 1992). Alachlor's half life was reported in a range from 14 to 50 days (Gustafson, 1989; Spalding, et al., 1989; Thurman, et al., 1992). The range of half lives listed for carbofuran was 30 to 45 days (Spalding, et al., 1979; Shirmohammadi, et al., 1989). Cyanazine had a half life range from 14 to 85 days (Gustafson, 1989; Thurman, et al., 1992). The range for metolachlor's half life was 20 to 180 days (Spalding, et al.,

Table 2  
Pesticide Half Live Ranges

Pesticide	Range of Half Lives, days
Alachlor	12-50
Atrazine	20-140
Carbofuran	30-45
Cyanazine	14-85
Metolachlor	20-180

\* Data from : Loague, et al., 1989; Thurman, et al., 1992; Gustafson, 1989; Spalding, et al., 1979; Spalding, et al., 1989; Shirmohammadi, et al., 1989; Bouchard, et al., 1982.

1979; Bouchard, *et al.*, 1982) (See Table 2).

#### **D. Pesticide Detection Methods**

##### **1. Liquid-liquid extraction**

Liquid-liquid extraction (LLE), was commonly used to separate analytes with polarities different from their original solvent into a new solvent. For water samples a nonpolar solvent like methylene chloride or hexene was added to the sample. The added solvent could not be miscible with water. This caused two separate phases to form, and nonpolar compounds would migrate into the added solvent (Blumberg, 1988). Liquid-liquid extraction was generally done as follows. The sample was poured into a separatory flask and a small amount of the nonpolar solvent was added (approximately 50 mL for a 1 L sample). The flask was shaken vigorously allow the analytes to contact the solvent. The flask was then set down and time was given for the solvents to separate. The nonpolar solvent was decanted into another container. This process was repeated twice, and all of the nonpolar solvent was collected, then evaporated down to the required volume.

While LLE works well at extracting most nonpolar compounds from water, it uses a relatively large amount of solvents. This was a concern because most nonpolar solvents are either known (benzene) or suspected (methylene chloride) carcinogens. Occasionally, an emulsion would form at the interface between the two solvents, making the removal of the

second solvent difficult or impossible. Also, the shaking procedure often caused lost samples due to spills and dropping of the flask.

## 2. Solid phase extraction

In Solid Phase Extraction (SPE) the sample is drawn through a porous matrix by a vacuum. The matrix was made of tetrafluoro-polyethene (TFPE) fibril holding either octyl or octadecyl bonded silica. Nonpolar compounds were attracted and adsorbed to the hydrocarbon chains on the silica in the matrix. After the water sample was passed through the matrix, the nonpolar solvent was drawn through the matrix. The compounds were desorbed from the hydrocarbon chains and dissolved into the solvent. For a 1 L sample, approximately 30 mL of solvent was required, while LLE would need about 5 or 6 times more solvent for the same sample.

### a. Cartridges

With Method 525.1, "Determination of Organic Compounds in Drinking Water by Liquid-Solid Extraction and Capillary Column Gas Chromatography/Mass Spectrometry", the EPA switched from LLE to SPE for most pesticides and poly-aromatic hydrocarbons (Eichelberger, *et al.*, 1991). The original technique used small plastic cartridges, similar to the body of a syringe, to hold the TFPE/bonded silica matrix. The cartridges only held approximately 10 mL of the sample, so multiple transfers of the sample or a cartridge reservoir was required. The



plastics in the cartridge could interfere or contaminate the sample being extracted (Junk, et al., 1988). Also, the small surface area of the cartridges lead to frequent clogging and long extraction times, 2-3 hours at the recommended flow rate (Kraut-Vass and Thoma, 1991).

#### b. Disks

An alternative to the cartridges, Empore<sup>tm</sup> extraction disks were designed to decrease the extraction time and the probability of clogging. The Empore<sup>tm</sup> extraction disks were 47 mm diameter and 0.5 mm thick. They were 90% octadecyl-bonded silica (octyl-bonded was also available) and 10% PFTE by weight (Figure 1). Hagen said, "The SPE disk can be considered to be a large-diameter, short-length extraction column." (Hagen, et al., 1990) At the recommended flow rate, a 1 L sample was extracted in about 30 minutes. Although the overall flow rate increased, the specific discharge (flow/area) in the disk was about the same as it was in the cartridge. In a comparison of the cartridges vs. the disks, Kraut-Vass and Thoma (1991) said, "Despite the short contact time, the general performance of the disk compared well with that of the extraction cartridge." Finally, since the disk was made entirely of PFTE and bonded silica the risk of contamination or interference with the sample was reduced.

#### E. Limits of Detection

There was a wide range in the limit of detection (LOD) of

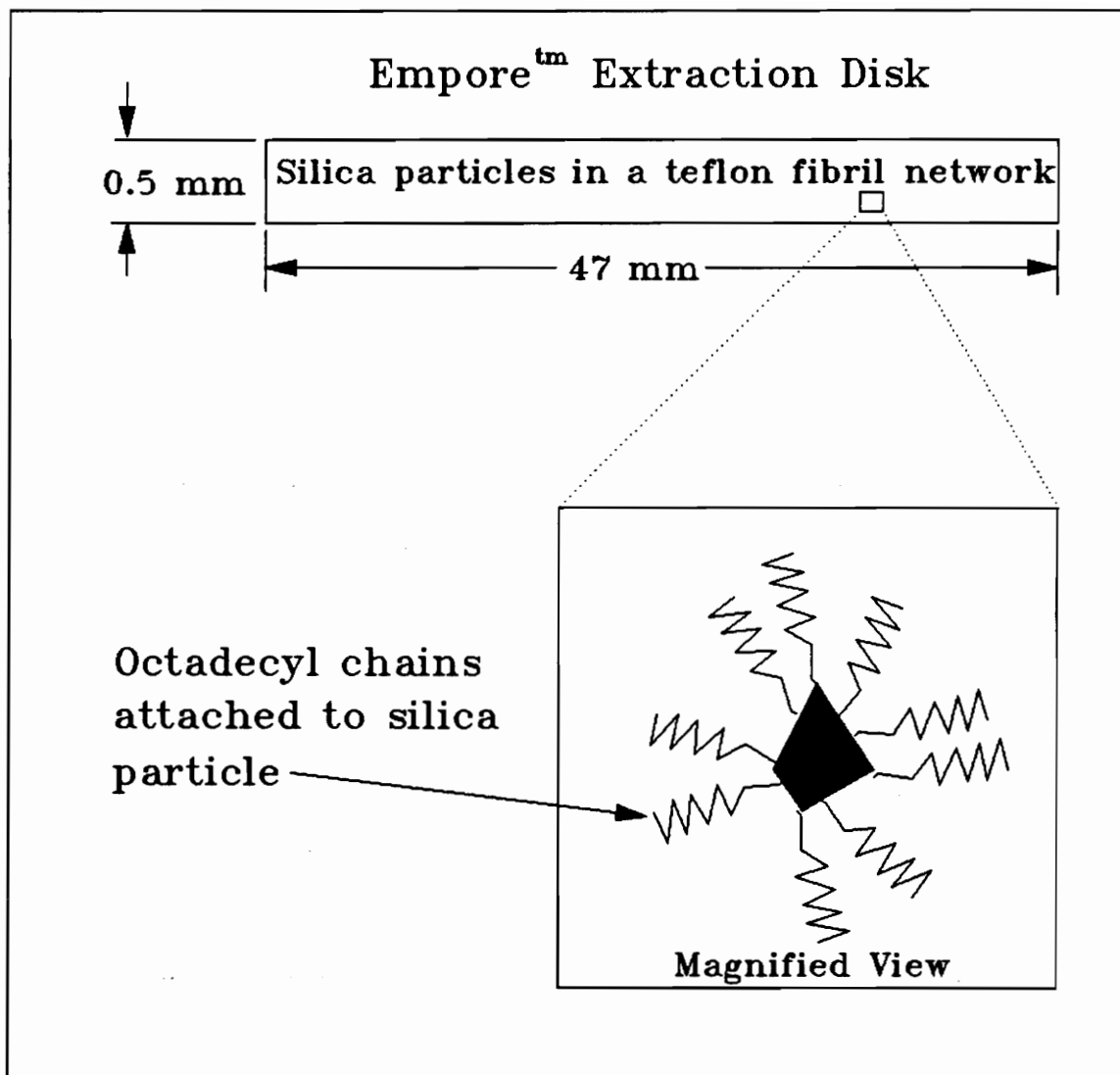


Figure 1  
Diagram of Solid Phase Extraction Disk

the pesticides reported in the literature. The lowest reported LODs were: 0.01  $\mu\text{g/L}$  for atrazine (Spalding, et al., 1989; Wehtji, et al. 1983), 0.03  $\mu\text{g/L}$  for alachlor, and metolachlor (Holden, et al., 1992), 0.05  $\mu\text{g/L}$  for cyanazine (Exner, 1990), and 0.05  $\mu\text{g/L}$  for carbofuran (Legrand, et al., 1991). The highest LODs reported were: 1.2  $\mu\text{g/L}$  for atrazine (Steichen, et al., 1988), 1  $\mu\text{g/L}$  for alachlor, metolachlor, and cyanazine (Krill and Sonzogni, 1986; Walker and Porter, 1990), and 5  $\mu\text{g/L}$  for carbofuran (Walker and Porter, 1990) (See Table 3). Interestingly all the highest LOD's were in studies using LLE.

#### **F. Efficiencies of Recovery**

Since SPE with the Empore<sup>tm</sup> disks is newly approved by the EPA, few literature sources are available. However three sources were found:

In a test of 43 organic compounds, the percent recovery varied from 18% to 519% with an average of 103% (Kraut-Vass and Thoma, 1991). In a test of phthalates in tap water, Hagen, et al. (1990) found percent recoveries from 82 to 117%. Finally, EPA Method 525.1 listed the recovery of 2-chlorobiphenyl at 78% at 0.2  $\mu\text{g/L}$  (Eichelberger, et al., 1991).

For the study 4-chlorobiphenyl was used as a surrogate at 10  $\mu\text{g/L}$  with an average recovery of 74%, which was within the range found in the literature, and only 4% different from the

listed value for 2-chlorobiphenyl by the EPA.

Table 3  
Range of LOD's in the Literature

Pesticide	LOD range, in $\mu\text{g/L}$
Alachlor	0.03-1
Atrazine	0.01-1.2
Carbofuran	0.05-5
Cyanazine	0.05-1
Metolachlor	0.03-1

\* Data from: Spalding, et al., 1989; Wehtji, et al., 1983; Holden, et al., 1992; Exner, 1990; Legrand, et al., 1991; Krill and Sonzogi, 1986; Walker and Porter, 1990.

### III. Methods & Materials

#### A. Research Chemicals

##### 1. Pesticides, Chemical Standards, and Solvents

All pesticides and chemical standards were bought from Chemical Services, Incorporated, West Chester, PA. The pesticides used in the study were: atrazine (2-chloro-4-ethylamino-6-isopropylamino-1,3,5-triazine), CAS # 1912-24-9, alachlor (2-chloro-2',6'-diethyl-N-(methoxymethyl) acetamide), trade name Lasso, CAS # 15972-60-8, carbofuran (2,3-dihydro-2,2-dimethyl-7-benzofuranol methylcarbamate), trade name Furadan, CAS # 1563-66-2, cyanazine (2-[[4-chloro-6-(ethylamino)-S-triazin-2-yl]amino]-2-methylpropanenitrile), trade name Bladex, CAS # 21725-46-2, and metolachlor (2-chloro-N-(2-ethyl-6-methylphenyl)-N-(2-methoxy-1-methylethyl)acetamide), trade name Dual, CAS # 51218-45-2.

The internal standards were 4-chlorobiphenyl, and the two deuterated compounds acenaphthene-d<sub>10</sub>, and phenanthrene-d<sub>10</sub>, CAS # 1517-22-2.

The main solvent used for extraction and for the preparation of most of the chemical standards was pesticide and Optima<sup>™</sup> grade methylene chloride (dichloromethane) CAS # 75-09-2. Pesticide grade methanol CAS # 67-56-1, and acetone CAS # 67-64-1 were also used. Anhydrous Sodium Sulfate CAS # 7757-82-6 was used as a drying agent. All of the solvents and the sodium sulfate were bought from Fisher Chemical

Corporation (Raleigh, NC).

## 2. Stock Solution Preparation

All pesticide and chemical stock standards were made in 2 mL volumetric flasks at a concentration of 5 mg/mL, and then stored in 4 mL amber vials with teflon lined lids at 4° C. The alachlor stock standard was made in acetone, but all the other standard solutions were prepared in methylene chloride. Solutions of lower concentrations were made by injecting the necessary number of  $\mu\text{L}$  of stock standard into a 2 mL volumetric flask and then diluting to 2 mL with methylene chloride. Calibration standards were made for quantitative analysis of samples; these solutions were prepared at 0.01, 0.05, 0.1, 0.25, 0.5, 1, 2, 5, and 10 mg/L. Spiking solution, for adding to the matrix/spike duplicates and reagent blanks, were prepared from the stock solutions at a concentration of 50  $\mu\text{g/mL}$ . Varying volumes of this solution were added to the spiked samples to create the proper concentration.

### B. Laboratory Glassware Procedures

#### 1. Glassware

For extraction, a three piece, all glass, vacuum filter was used (Kontos, Vinland, NJ). The extract was stored in 40 mL amber, borosilicate glass vials with teflon lined tops (Scientific Specialties, Randallstown, NJ), until blown down, when they were stored in 4 mL amber, borosilicate glass vials with teflon lined tops, (Scientific Specialties, Randallstown,

NJ).

## 2. Cleaning Procedures

The 1 liter sampling jars were washed with soap and water, then rinsed three times with distilled, deionized water. All extraction equipment was washed and rinsed three times with distilled, deionized water between samples. The 40 and 4 mL vials were rinsed three times with methylene chloride before each use.

### C. Field Sampling Procedures

#### 1. Site Selection

Five sites were sampled on Virginia's coastal plains, with three on the eastern shore of Virginia. Four of the sites are working farms with the fifth being an undeveloped marsh. The sites were chosen because the watershed characteristics were conducive to possible groundwater contamination, and the sites were already being used by another Virginia Tech researcher for nutrient transport studies (Dr. George Simmons Jr., Biology Department). A map of all the sites is included (Figure 2). The sites were sampled every 4 to 6 weeks except for the marsh which was sampled every 2 or 3 months. The sites are described below:

#### *Agricultural site 1*

This site was located on the Phillips Creek which drains into the Hog Island Bay in Northampton County. The surficial soils have been classified as Bojac sandy loam which



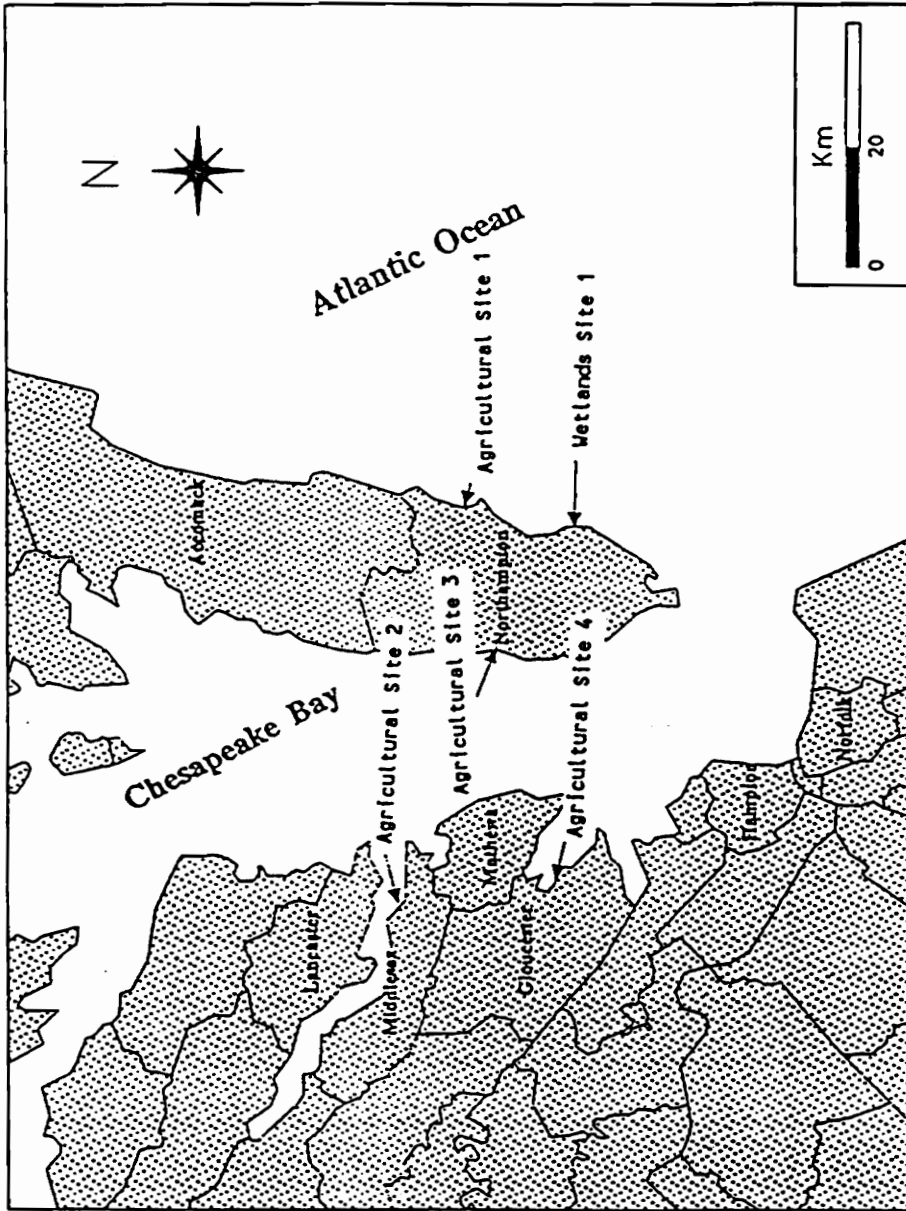


Figure 2  
Map of All Sampling Sites

demonstrates very rapid permeabilities (5.0 to 50.0 cm/hr). The water table was located approximately 1.0 meters below the soils surface. Surficial nearshore sediments were sandy and display vertical hydraulic conductivities of 14.0 cm/hr. Mean tidal range was 0.6 meters with summer salinities on the order of 30 ppt. Sampled well locations were shown in Figure 3.

#### *Agricultural site 2*

This site was located on the southern shore of the Rappahannock River in Middlesex County. The spring 1991 crop was corn with atrazine and metolachlor applied for weed control. The winter 1991 crop consisted of winter wheat with soybeans planted in spring 1992. The surface soils have been classified as Slagle silt loam which demonstrates moderate permeabilities (1.5 to 5.1 cm/hr) (Newhouse et al., 1985a). The water table was located approximately 4.0 meters below the soils surface. Surficial near shore sediments were sandy underlain by a thin clay lens. Vertical hydraulic conductivities were on the order of 0.6 cm/hr. Mean tidal range was approximately 0.5 meters with summer salinities on the order of 13 ppt. The approximate location of the sampled wells was shown in Figure 4.

#### *Agricultural site 3*

This site was located on Cherrystone Inlet in Northampton County. Winter 1991 crop consisted of winter wheat with soybeans planted in spring 1992. In the

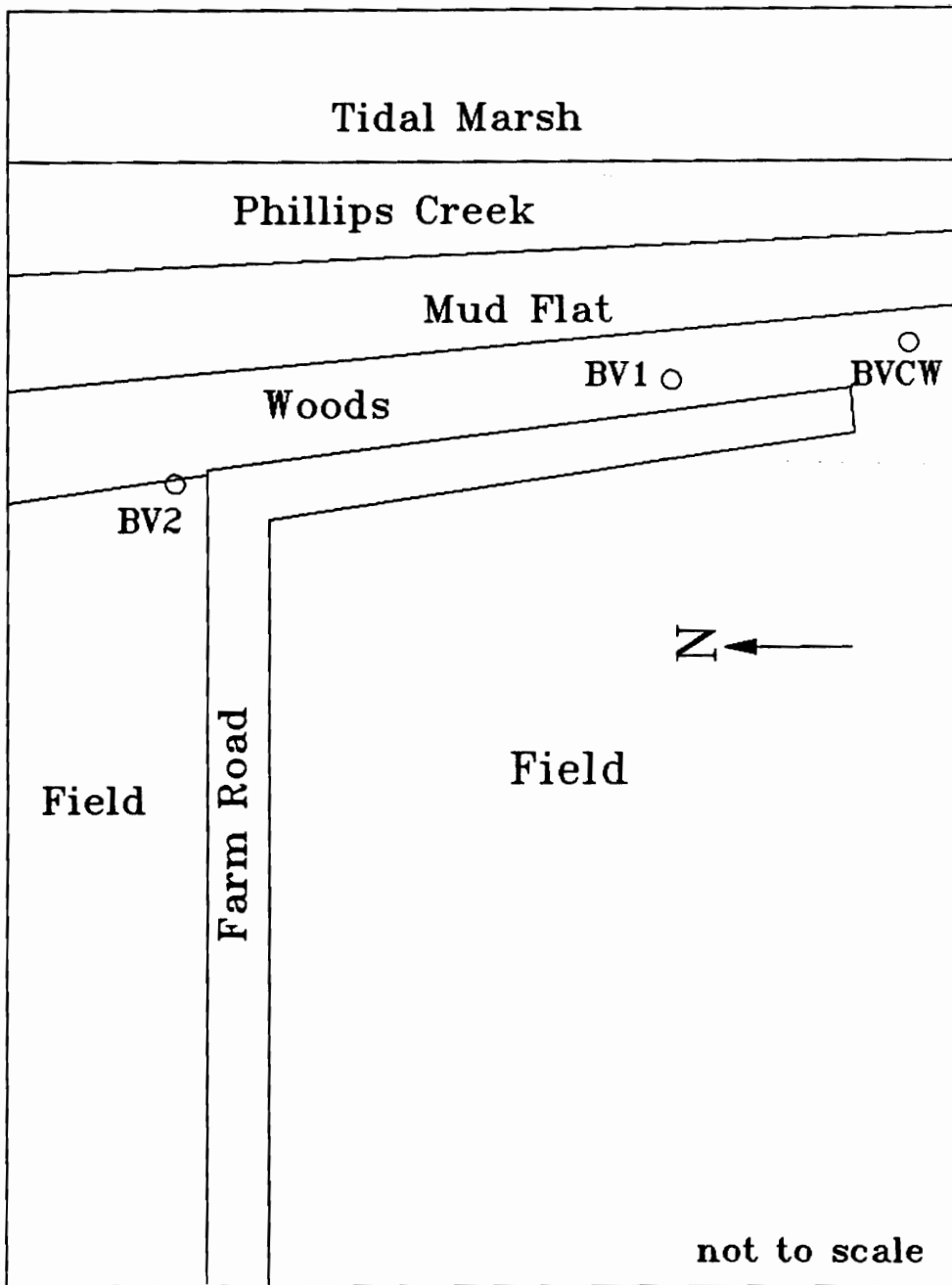


Figure 3  
Wells at Agricultural Site 1

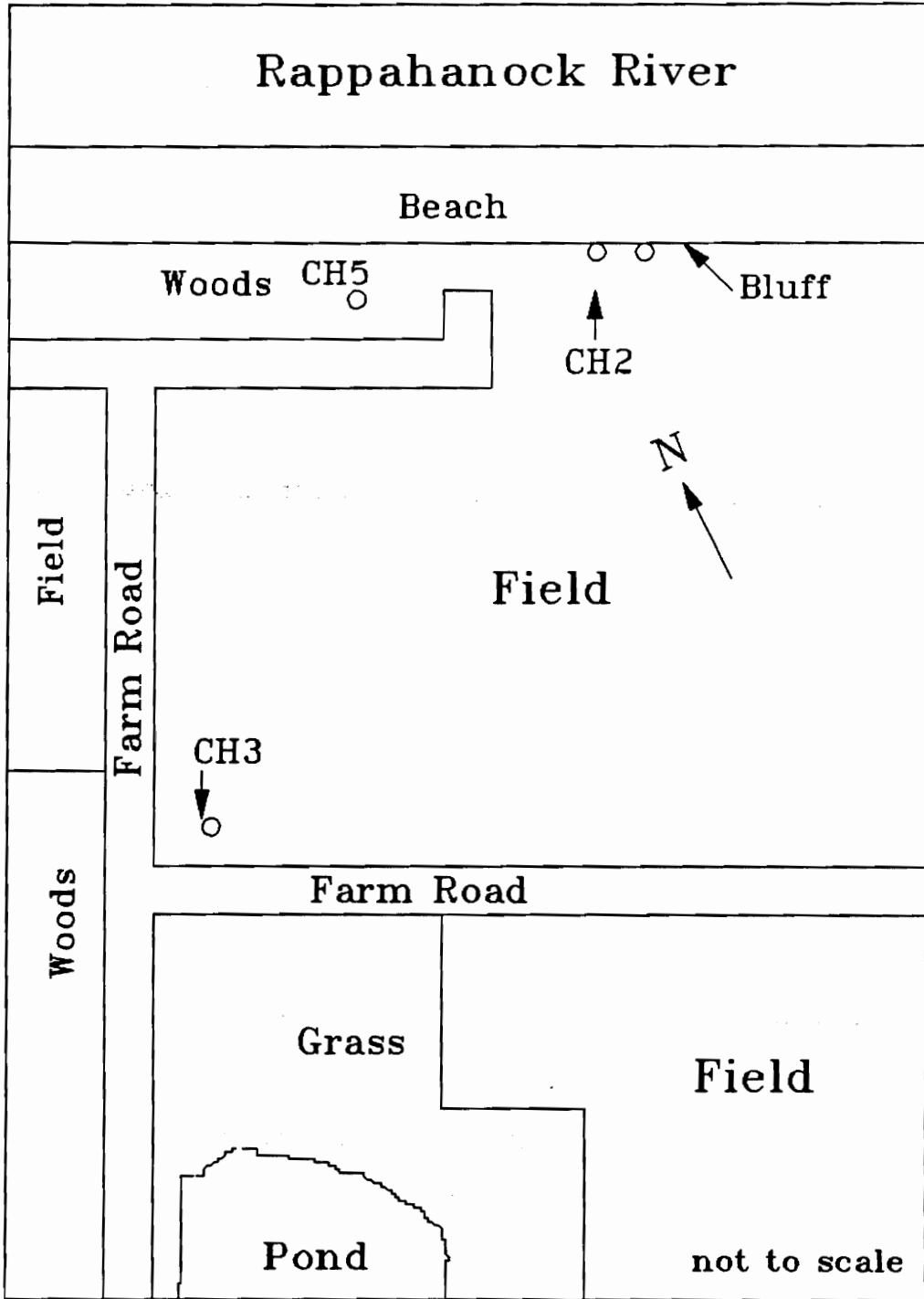


Figure 4  
Wells at Agricultural Site 2

fall of 1992 barley was planted. None of the pesticides monitored by the study were applied in 1991 or 1992. The surficial soils have been classified as Bojac fine and Munden sandy loams which demonstrate moderately rapid (5 to 15 cm/hr) and rapid (15 to 120 cm/hr) permeabilities (Cobb and Smith, 1989). The water table was located approximately 2 meters below the soils surface. Surficial nearshore sediments were sandy with vertical hydraulic conductivities on the order of 16.1 cm/hr. Mean tidal range was 0.7 meters with summer salinities on the order of 20 ppt. The approximate location of the sampled wells was included in Figure 5.

#### *Agricultural site 4*

This site was located on the southern shore of the Piankatank River in Gloucester County. The surficial soils have been classified as a mix of Psammets-Hamludults complex and Ochlockonee Variant complex which demonstrate permeabilities varying from moderately slow (0.5 to 1.5 cm/hr) to rapid (15.2 to 50.8 cm/hr), (Newhouse et al., 1985b). The water table was located approximately 1.0 meters below the soils surface. Surficial nearshore sediments were sandy with vertical hydraulic conductivities on the order of 0.9 cm/hr. Mean tidal range was 0.4 meters with summer salinities on the order of 11 ppt. Figure 6 contained the approximate well locations.

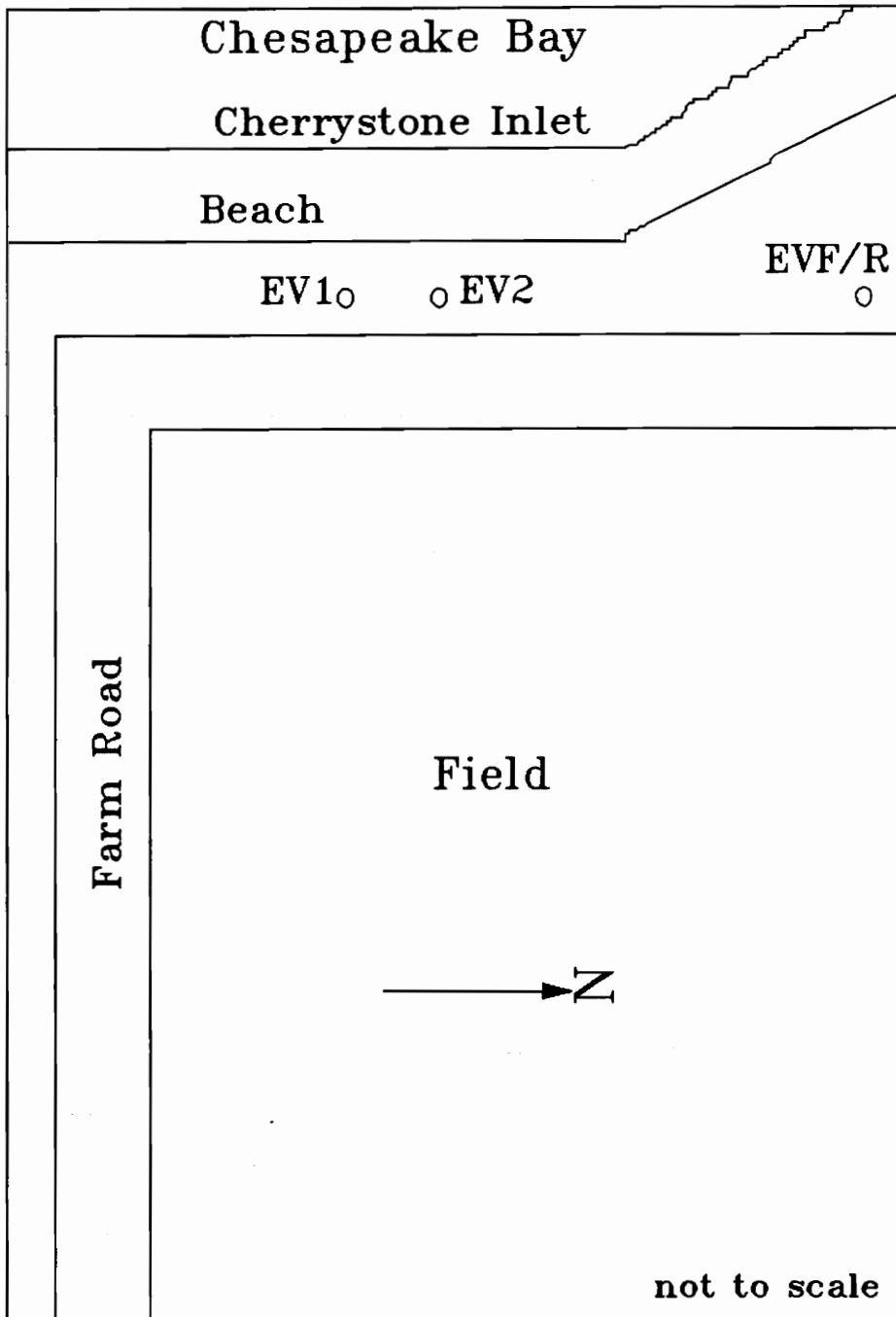


Figure 5  
Wells at Agricultural Site 3

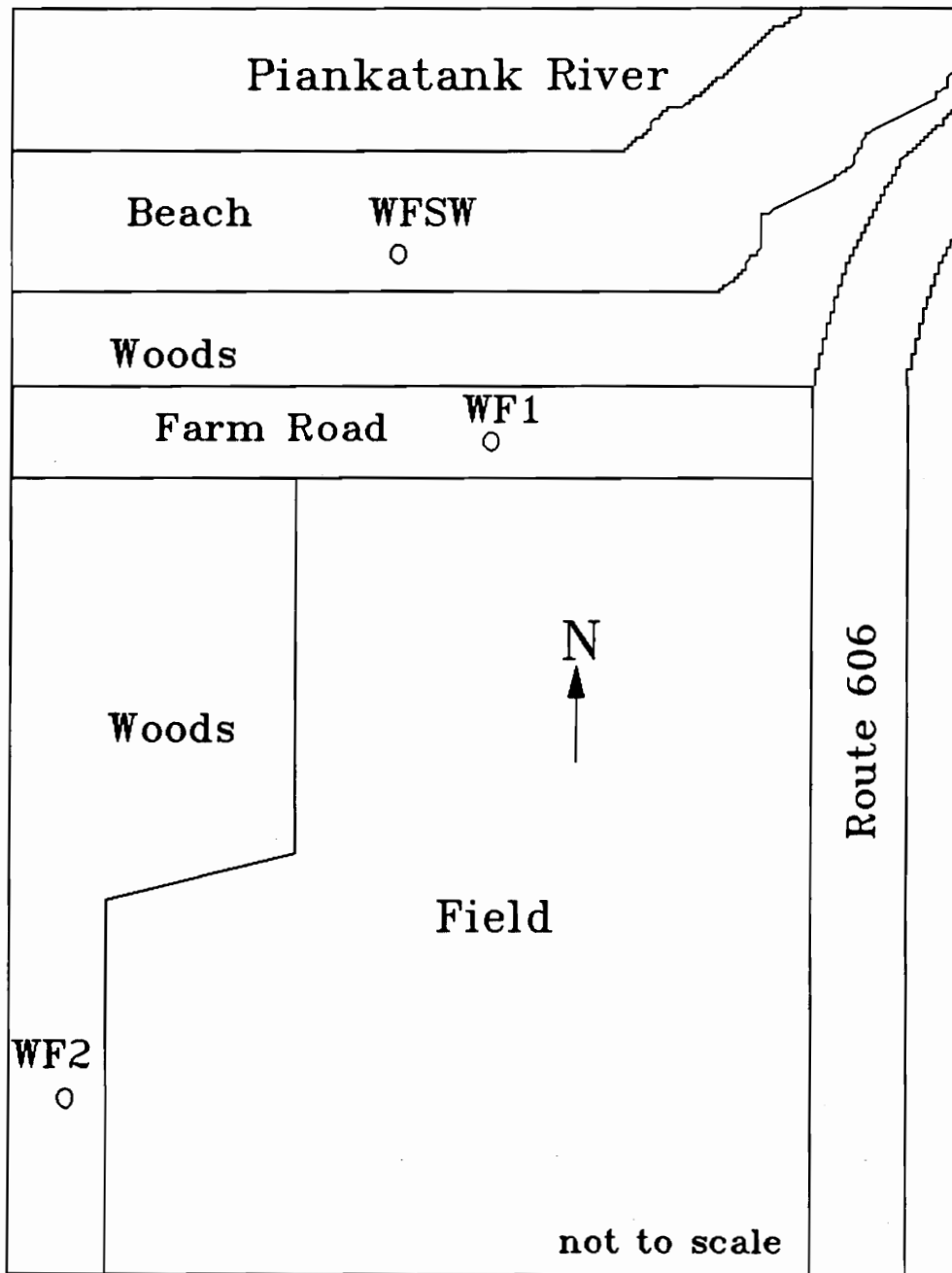


Figure 6  
Wells at Agricultural Site 4

### *Wetlands site 1*

This site was located on Magothy Bay in Norhtampton County. The site was adjacent to 300 meters of salt marsh followed by 300 meters of mesic forest before reaching an upland agricultural field. Surficial soils have been classified as Chincoteague silt loam. Mean tidal range was 0.9 meters with summer salinities on the order of 32 ppt. This site served as a natural background site, and had no agricultural activity. Figure 7 showed the estimated locations of the sampled wells.

### 2. Sampling Equipment

All wells were water jet drilled by or under the directions of Dr. William Reay. All the wells, except the middle well at wetlands site 1, penetrated only the top, unconfined aquifer. All wells extended approximately 1 meter below the water table. The bottom meter of the wells was screened to allow groundwater entrance. The wells were packed with pea gravel for approximately 1.5 meters from the bottom of the well. The middle well at wetlands site 1 tapped the second, confined aquifer. Agricultural site 2 had the deepest wells, approximately 6 meters, and site 4 had the shallowest at about 1.3 meters. The wells were lined with 5 cm diameter polyvinylchloride (PVC) casing and well screen.

The wells were pumped at 1 liter per minute with a Geotech Geopump 2 battery powered peristaltic pump with a 50



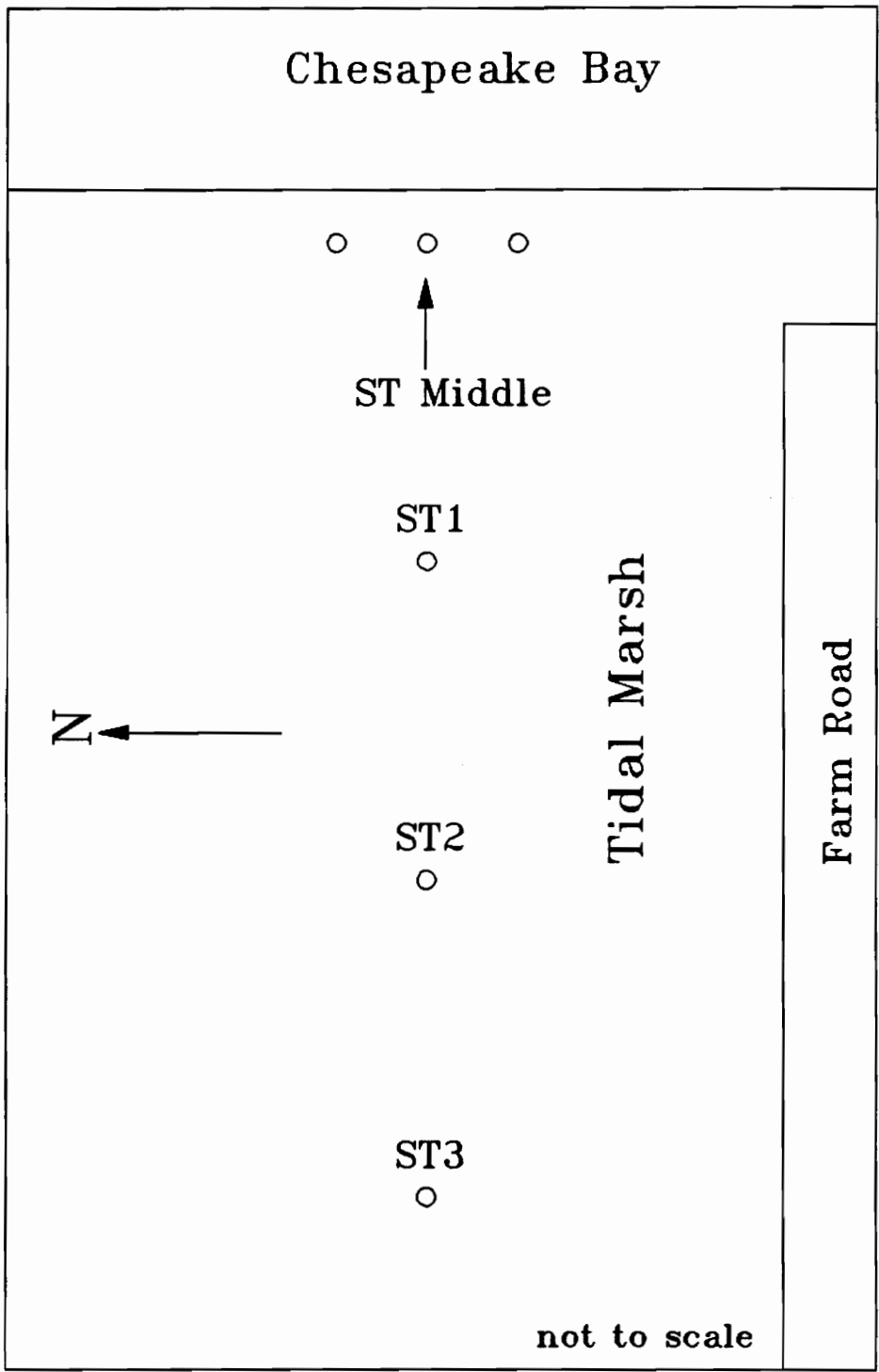


Figure 7  
Wells at Wetland Site 1

cm piece of three-sixteenths inch inside diameter Nalgene™ tubing in the pump head. To lift the water from the well to the pump, three-sixteenths inch inside diameter teflon tubing was used. The samples were collected directly from the end of the pump and collected in 1 liter, amber borosilicate glass jars with teflon lined lids.

Seepage meters were used to collect the ground water seeping into the Chesapeake Bay. Seepage meters were one ended cylinders .75 meters in diameter and 20 cm deep. The top, enclosed end of the cylinder, had a hole drilled into it for attachment of a teflon bag to collect the seepage. The seepage meters were made of Nalgene® and plexiglass with a silicone sealant. The seepage bags were teflon gas sampling bags having a 3.75 liter volume (Fisher, Raleigh, NC). Seepage bags were connected to the seepage meters by a nalgene tube and a rubber stopper.

### 3. Sampling Procedure and Preservation

#### a. Well Samples

The well water samples were taken after approximately 5 well volumes had been pumped out of the well. The sample was taken directly from the end of the pump. Three wells at each site were sampled each 4 to 6 weeks. Originally, duplicate samples were taken from two wells and quadruplicate samples from the third well. Starting in September, 1992, two wells had quadruplicate samples taken, and the remaining well had

duplicate samples taken. Occasionally, one of the wells would not recharge enough for the full amount of samples to be taken from the site. The samples were stored on ice in coolers until returned to the laboratories at Virginia Tech, where they were stored in refrigerators at 4° C until extraction. Samples for the nutrient study and Ohmicron's immunoassay procedure were taken at the same time as the study's samples.

#### b. Seepage Meter Samples

At low tide, four seepage meters were placed far enough offshore that they were completely under water. The meters were imbedded into the sediments with the closed end up to within an inch of the top of the meter (Figure 8). This was done to minimize the amount of sea water trapped in the headspace between the sediment and the top of the meter. At high tide, the bags were removed and the volume of the seepage water was measured. This water was discarded, for it was mostly sea water that had been trapped in the head space of the meter. Then the collection bags were replaced. At the next low tide the collection bags were removed and the seepage meter samples were taken directly from the teflon sampling bag. The bags were 3.75 L, but they were rarely full. All the water in the collection bags at low tide was retained for analysis. See Table 4 for a complete listing of actual volumes for seepage meter samples. The seepage meter samples were also stored on ice in coolers until returned to the

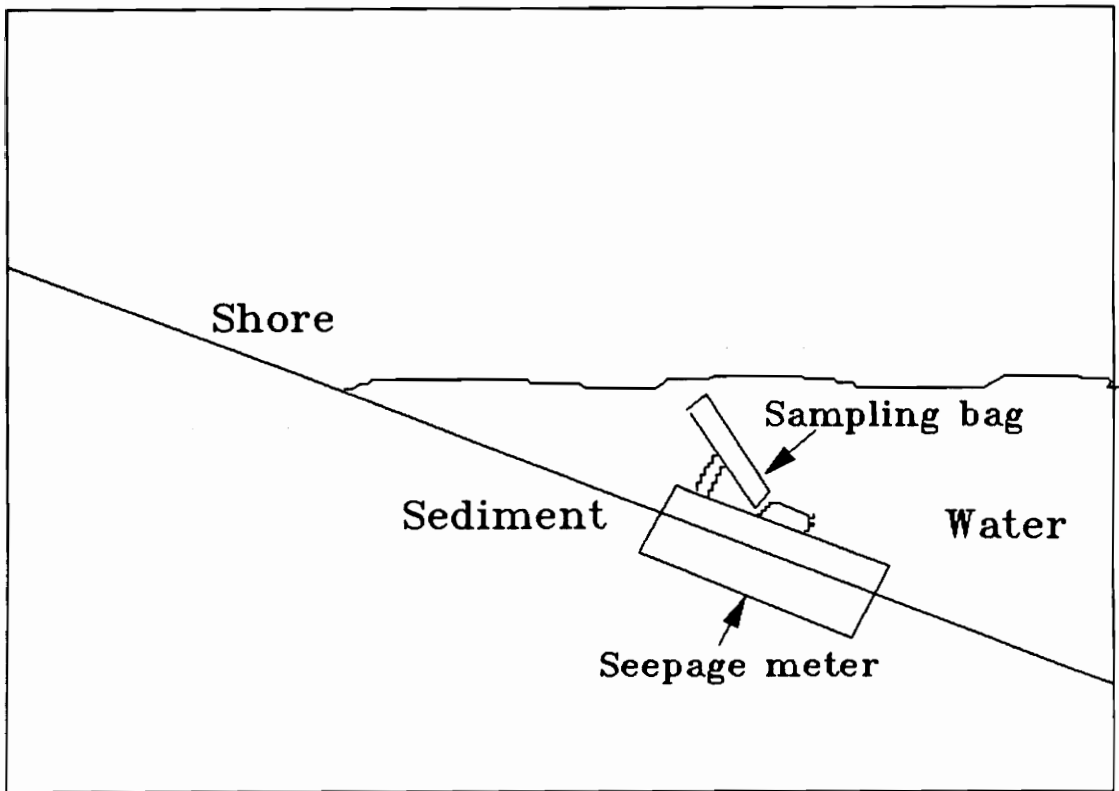


Figure 8  
Typical Seepage Meter Placement

**Table 4**  
**Seepage Meter Sample Volumes (mL)**

Site	SM <sup>a</sup>	Month								
		May	June	July	Aug	Oct	Nov	Dec	Jan	Feb
EV <sup>b</sup>	1	700	1300	700	1300	2200	200			200
	2	1200	900	600	1000	2300	500	1300		150
	3		1300		1300	1300	600			
	4	500	700		1100	2300	1300			
CH <sup>c</sup>	1			1300	1300	300		CLOSED for Hunting Season		290
	2		1300	1300		700			480	
	3	1300	1100		1300	1300			1280	
	4			1300		300			2280	
BV <sup>d</sup>	1	1300	2900		400		700			3600
	2	1300		800	100	1300				580
	3	2300	1100		300		3500			4300
	4	1300	1300		700	2300	2300			4700
WF <sup>e</sup>	1					1500			1100	880
	2				500	900		1300	1700	1250
	3		1300		1300	1300			1700	1480
	4		1300		800				3500	1580
ST <sup>f</sup>	1			1300						
	2			1300						
	3			1000						
	4			1300						
	5			1300						
	6			1300						

- a SM was Seepage Meter number
- b EV was Agricultural Site 3
- c CH was Agricultural Site 2
- d BV was Agricultural Site 1
- e WF was Agricultural Site 4
- f ST was Wetlands Site #1

*goosegrass*

laboratories at Virginia Tech. At Virginia Tech, they were stored at 4°C until extracted. Samples for Dr. Simmons' nutrient study and Ohmicron's testing were taken directly from the collection bags.

#### **D. Laboratory Analyses**

##### **1. Extraction**

All samples were spiked with surrogates of deuterated acenaphthene and phenanthrene. Chrysene-d<sub>10</sub> was tried in May but discontinued due to its much longer detention time than the pesticides of interest. Starting in September, 1992, 4-chlorobiphenyl was added to all the samples as a surrogate for detection on the electron capture detector (ECD). The deuterated surrogates were added at 5 µg/L for the April samples and 2 µg/L for all others, while the 4-chlorobiphenyl was added at 10 µg/L.

Solid phase extraction was done with 47 mm Empore<sup>tm</sup> extraction disks (Varian, Harbor City, CA) in an all glass filtering apparatus (Kontos, Vineland, NJ). The apparatus consists of a funnel; a vacuum armed, fritted glass filter, and a 2 L flask. The Empore disk was placed between the funnel and the fritted glass filter. Due to the turbidity of some of the samples, all the extractions used a 1 micrometer non-bonded glass fiber filter (Fisher Scientific, Raleigh, NC) seated on top of the Empore disk to trap the particulates and prevented them from clogging the Empore disk (Figure 9).

After assembly, the Empore disk was preconditioned by drawing 5 mL of methylene chloride through the disk, leaving a meniscus of liquid on the top. Five mL of methanol was then drawn through, again leaving a little liquid on top of the filter. Finally, 5 mL of reagent water was drawn through, leaving a layer of water on top. The disk was not allowed to dry, as this causes the performance to become erratic and unreproducible.

After preconditioning, the 1 liter sample was drawn through the apparatus. Flow rate varied depending on the turbidity of the sample, but averaged 30 mL per minute. When the entire sample had passed through the disk (approximately 30 to 60 minutes), the disk was allowed to dry for 5 minutes with the vacuum attached to reduce the amount of water trapped in the disk. The 2 L flask was replaced by a smaller flask containing a 40 mL test tube. Three 5 mL volumes of methylene chloride were passed through the disk into the test tube, with the disk allowed to dry between each volume. The first two volumes were also used to rinse the sample bottle before being poured into the apparatus. The final rinse was pipetted around the rim of the funnel of the apparatus. The 15 mL of methylene chloride eluant were passed through a funnel with a few grams of anhydrous sodium sulfate to absorb and remove any residual water. The eluant was evaporated down to approximately 1 mL under a cool nitrogen gas stream, then

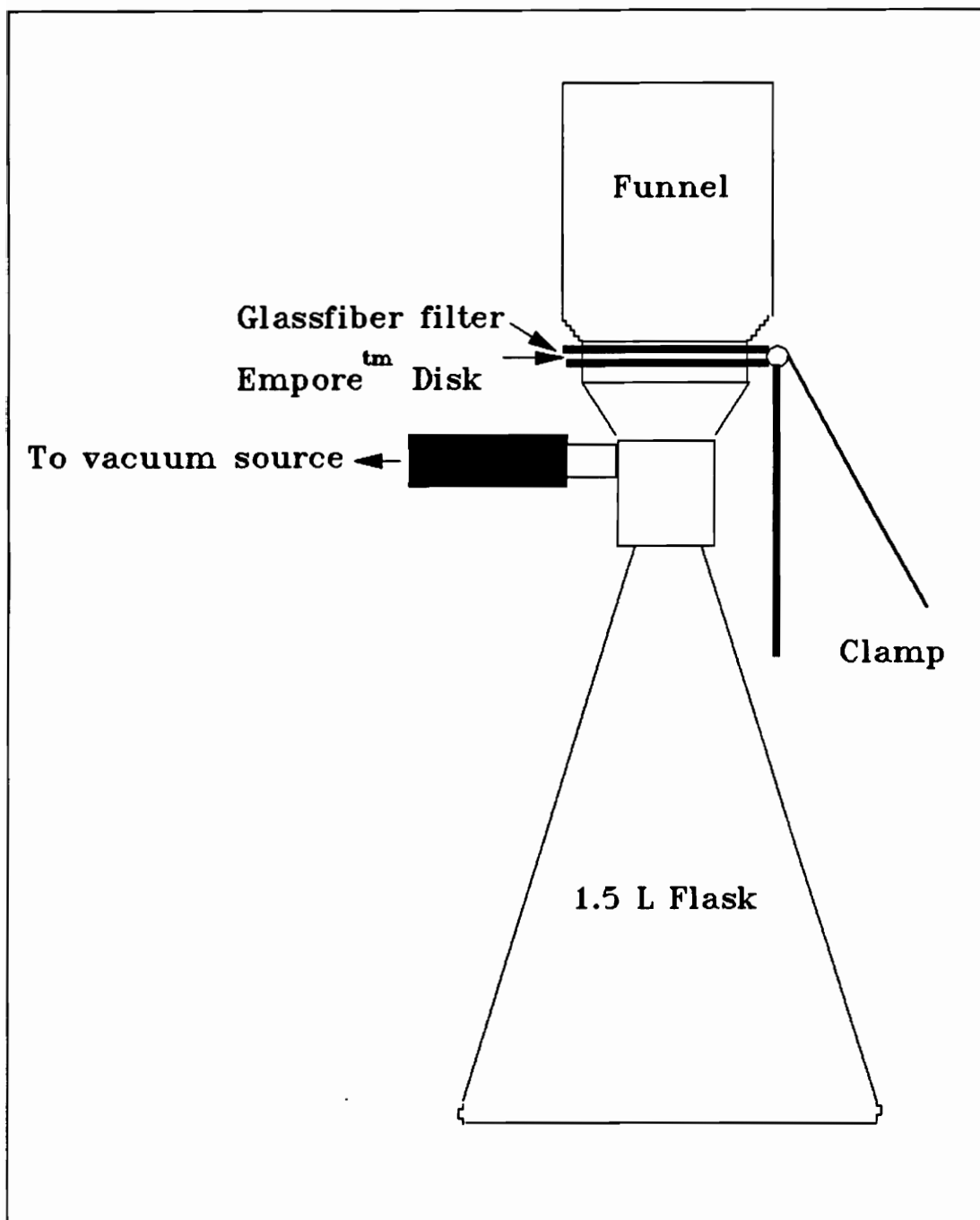


Figure 9  
Extraction Apparatus



stored at 4° C until analyzed on the GC.

## 2. Gas Chromatography

All samples were injected on a Hewlett-Packard Gas Chromatograph model # 5890, with a Hewlett-Packard model # 3396 series II Integrator. Both a Flame Ionization Detector (FID) and a Electron Capture Detector (ECD), with Ni<sup>63</sup>, were used. Selected samples were injected on a Hewlett-Packard Gas Chromatograph model # 5890 series II; connected to a Hewlett-Packard Mass Spectrum Detector (MS) model # 5970 with a Hewlett-Packard data station model # 5970. The details of each instrument were described as follows.

### a. GC/FID and GC/ECD

Originally, the FID was used with a 30 m x 0.25 mm x 0.25 μm DB-5 capillary column. The FID proved to be more sensitive than the GC/MS, so all samples from May and June 1992 were analyzed using the FID. However, since all the pesticides except carbofuran were chlorinated, the ECD was more sensitive than the FID for most of the pesticides. All samples from June 1992 to February 1993 were injected on the ECD. The ECD provided better sensitivity, but metolachlor (which was monitored after the July 1992 samples) and cyanazine co-eluted on the DB-5 column. To resolve the compounds, a different column was employed, the 30m x 0.25mm x 0.25 μm DB-210 (J&W Scientific, Rancho Cordova, CA). Although it did separate metolachlor and cyanazine, the DB-210 caused the carbofuran

and metolachlor peaks to co-elute, so the DB-5 column was also used to analyze the samples. From October, 1992 on, the ECD was used with both the DB-5 and DB-210 columns (Table 5).

The temperature program for the FID analysis was to start and hold for one minute at 45°C, then 42.5°C/min to 130°C, then 7°C/min to 180°C, and finally 12°C/min to 280°C with a final time of 3 minutes at 280°C. The injection port was at 250°C and the detector was at 300°C. The helium linear velocity was 35 cm/sec or 2 mL/min. Approximately 3 µL of each sample were injected.

For the ECD with the DB-5 column, the temperature program used was: start and hold for 1 minute at 75°C, then ramp at 10°C/min to 275°C and hold for two minutes. The injector port was at 250°C and the ECD was at 300°C. With the DB-210 column, the program was: start and hold for 1 minute at 140°C, then ramp at 5°C/min to 240°C, with a final time of 2.5 minutes at 240°C. The injector port and the detector were at 250°C. The helium linear velocity was 35 cm/sec and the nitrogen make up gas flow was 70 mL/min for both columns.

For the GC/ECD, calibration curves were made by multiple injections (4 or 5) of the 0.1, 0.5, 0.75, 1, 2, and 5 mg/L calibration standards. A calibration curve was calculated using least squares regression. A single calibration standard, usually the 2 mg/L, was injected at the beginning of each work day and after every eight hours of analysis. If the

calibration standard was not within 1 standard deviation of the predicted value based on the calibration curve, then no samples were analyzed until the instrument was demonstrated to be operating properly. The chromatograph of a typical standard was included as Figure 10. Samples in which pesticides were detected were also analyzed by the GC/MS.

On January 5, 1992 an autosampler was installed on the GC/ECD. Some of the December samples and all of the January and February samples analyzed on the GC/ECD used the autosampler for injection.

b. GC/MS

A DB-5 30 m x 0.25 mm capillary column with a film thickness of 0.25  $\mu\text{m}$  (J&W Scientific, Rancho Cordova, CA) was used for the pesticide analysis. A multi-ramp temperature program as specified in EPA Method 525.1 was used for separation of the compounds. Approximately 3  $\mu\text{L}$  of a sample was injected each analysis.

The temperature program was as follows: the initial oven temperature was 45°C held for one minute followed by a ramp of 28.3°C / minute to 130°C immediately followed by a ramp of 12°C / minute to 180°C which was in turn followed by a 7°C / minute ramp to a temperature of 240°C and finally a 12°C ramp to 280°C. The injection temperature was 220°C and the transfer line was held at 250°C. The carrier gas was helium with a linear velocity of 33 cm/sec.

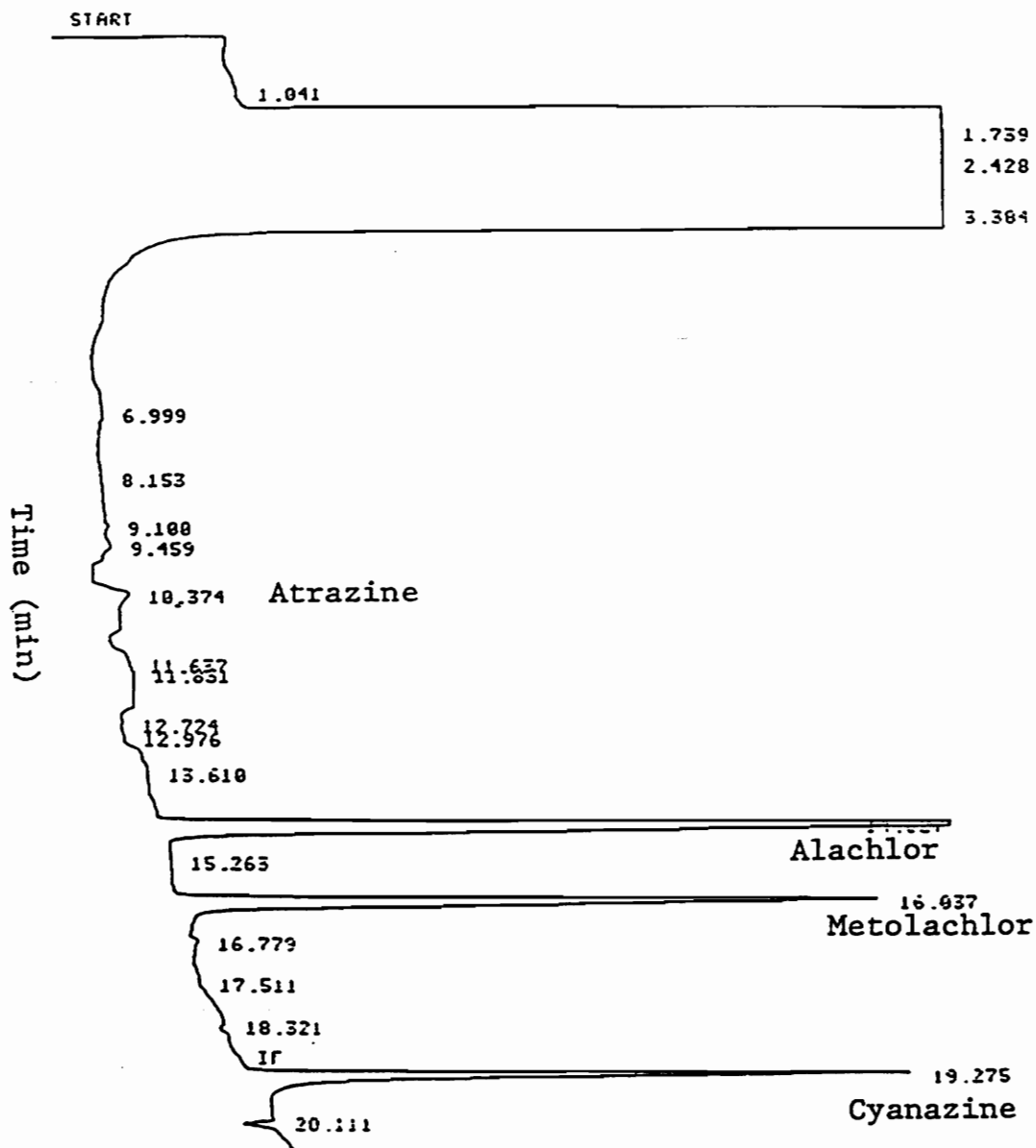


Figure 10

Chromatogram of a 0.5 g/L Calibration Standard Using the GC/ECD

The Mass Spectrometer (MS) was a quadropole model with 70 electron volt electron impact ionization. A mass range from 50 to 450 m/z was scanned. The scanning rate was 1.1 scans per second. Compounds were identified by retention time and comparison to the online library of spectra in the data station (Wiley-NBS database). For the GC/MS the autotune program was run at the start of every day's run and after eight hours work. The autotune used perfluorotributylamine (PFTBA) as a test substance. The autotune was checked by injecting decafluorotriphenyl phosphine (DFTPP) and checking the relative strength of its ions against the guidelines stated in EPA Method 525.1. A pesticide calibration standard was also injected to evaluate and compare retention times, mass spectra, and response factor (Figure 11). The retention times of the surrogates and pesticides were listed in Table 5.

The pesticide concentration in an individual sample was calculated from the response factors generated from the internal standards (acenaphthene-d<sub>10</sub>, and phenanthrene-d<sub>10</sub>). The quantification ions specified in EPA Method 525.1 were used to generate response factors and to calculate pesticide concentrations. These response factors were calculated by injecting standard pesticide solutions at 0.75, 1, 2, and 5 mg/L with a constant internal standards concentration of 2 mg/L. The standard response factor equation was used, which is as follows:

**Table 5**  
**Pesticide Retention Times using the GC/ECD**

DB-5	Retention time (min)	DB-210	Retention time (min)
4-Chloro-biphenyl	12.00	4-Chloro-biphenyl	5.80
Carbofuran	14.77	Carbofuran	15.36
Atrazine	14.88	Atrazine	9.56
Alachlor	16.73	Alachlor	13.74
Metolachlor	17.50	Metolachlor	15.22
Cyanazine	17.53	Cyanazine	18.45

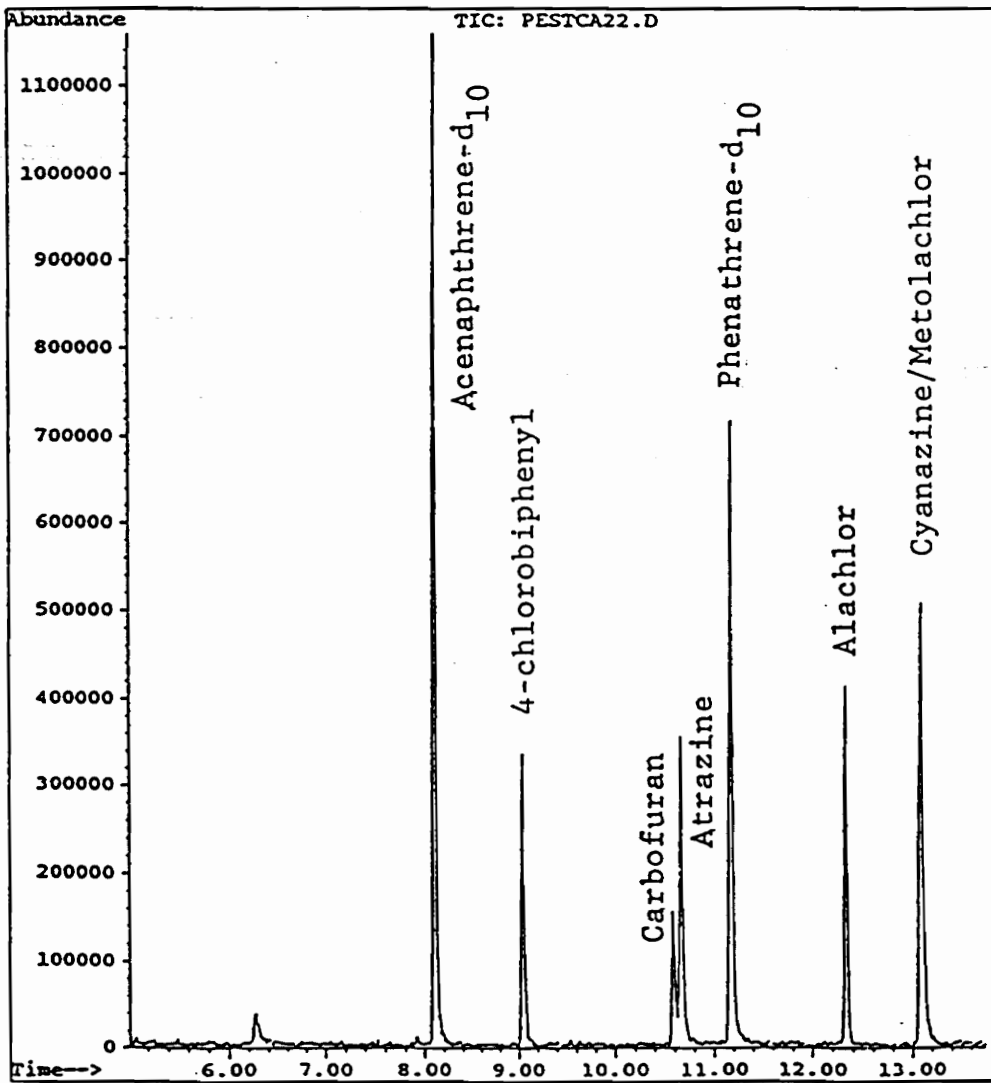


Figure 11

Chromatogram of a 5 µg/L Calibration Standard Using the GC/MS

$$RF = \frac{A_x C_{is}}{A_{is} C_x}$$

RF = Response Factor

$A_x$  = Area of the quantitation ion for the compound being measured

$A_{is}$  = Area of the quantitation ion for the specific internal standard

$C_x$  = Concentration of the compound being measured (mg/L)

$C_{is}$  = Concentration of the specific internal standard (mg/L)

Measured concentration were calculated in micrograms per liter (ppb).

$$\text{Concentration(ppb)} = \frac{A_x * Q_{is}}{A_{is} * RF * V}$$

$A_x$  = Area of characteristic ion for compound being measured

$Q_{is}$  = Total quantity of internal standard added to water sample (micrograms)

V = Volume of original water sample.

The GC/MS had poor sensitivity for the five pesticides and was



only used to test for possible confirmation of pesticides detected by the GC/FID and GC/ECD.

## **E. Quality Assurance and Quality Control**

### **1. Instrument's Performance**

An independent analytical chemist not associated with the project prepared 6 check samples in July 1992 that were analyzed to determine any experimental bias. The samples were extracted and injected on the GC/MS. The regression values were generally close to the actual concentration. Carbofuran's values contained the largest errors. The values are listed in Table 6.

### **2. Method Detection Limits**

All the calibration standards were analyzed, with the instrument Lowest Detectable Quantity (LDQ) for each pesticide recorded. Table 7 listed the LDQ's for the pesticides on each instrument. The instrument Limit of Detection (LOD) was calculated after making the calibration curve for each instrument. The LOD's were determined by multiplying the standard deviation of the a calibration standard on the instrument by the t-test value at a 99% confidence level plus the background noise of the GC. The Method Detection Limits (MDLs) were calculated from the spiked matrix samples the same way as LODs, except the standard deviation of the spiked samples was used (See tables 8&9). In the project, the LODs for the pesticides was

**Table 6**  
**Pesticide Concentrations in the Check Standards**

Pesticide	Concentration, $\mu\text{g/L}$ (ppb)	
	Actual	Measured
Alachlor	1.0	ND
	1.0	1.2
	2.5	3.2
	2.5	2.3
	6.0	6.0
	6.0	6.0
Atrazine	1.0	0.5
	1.0	1.5
	2.5	2.8
	2.5	2.2
	6.0	4.5
	6.0	5.5
Carbofuran	1.0	0.5
	1.0	1.3
	2.5	2.1
	2.5	1.8
	6.0	4.0
	6.0	3.0

Table 7  
Instrument Lowest Detectable Quantities of Pesticides  
(in mg/L)

	GC/MS	GC/FID	GC/ECD
Atrazine	1	1	1
Alachlor	0.75	0.75	0.01
Carbofuran	2	1	1
Cyanazine	2	1	0.01
Metolachlor	0.5	0.5	0.01

\* Note that the LDQ did not include the extraction concentration factor of 1000, which would make the sample concentration in  $\mu\text{g/L}$

**Table 8**  
**Instrumental Limits of Detection of the Pesticides**  
**(mg/L)**

Pesticide	GC/MS	GC/FID	GC/ECD
Atrazine	1.09	3.5	0.59
Alachlor	2.5	3.9	0.099
Carbofuran	1.0	3.8	2.6
Cyanazine	1.9	5.9	0.025
Metolachlor	0.3	N/A*	0.061

- \* The GC/FID was not used to determine metolachlor
- \* Note that the LDQ did not include the extraction concentration factor of 1000, which would make the sample concentration in  $\mu\text{g/L}$

Table 9  
Method Detection Limits<sup>a</sup> of the Pesticides  
( $\mu\text{g/L}$ )

Pesticide	GC/MS <sup>b</sup>	GC/ECD <sup>c</sup>	Immunoassay (MDC) <sup>d</sup>
Atrazine	0.64	0.36	0.05
Alachlor	0.88	0.33	0.04
Carbofuran	1.6	1.5	0.06
Cyanazine	1.9	0.32	0.04
Metolachlor	N/A*	0.50	0.06

\* GC/MS and GC/FID were not used to determine metolachlor

a The MDLs were determined by multiplying the standard deviation of an added concentration of a pesticide (see below) by the t-test value at the 99% confidence level.

b The added concentration used for the GC/MS was 2 mg/L, except for cyanazine. 5 mg/L was used for it.

c The added concentrations used for the GC/ECD were 0.25 mg/L for alachlor and cyanazine, 0.5 mg/L for atrazine and metolachlor, and 2 mg/L for carbofuran.

d Data from OHMICRON Corporation

within the range of the LODs found in the literature, except for the LODs using the GC/FID. However the GC/FID was only used for two months, May and June 1992.

Ohmicron's immunoassay procedure listed a Minimum Detectable Concentration (MDC), which is a measure of the minimum pesticide concentration necessary to yield a response in the immunoassay. Since the immunoassay method measures pesticides directly in the water sample without prior concentration, the MDCs are comparable to the solid phase extraction/GC MDLs.

### 3. Duplicates

Duplicate samples were taken from all wells each trip. This was done to assess the variability of the sampling analysis techniques, and for safety in case a sample was lost accidentally. Also, one well from each site from April to September, and two wells from each site after September had quadruplicate samples taken for spikes.

### 4. Blanks

For every group of twenty samples, one lab blank, one equipment blank, and one travel blank were analyzed. All blanks were made with Milli-Q reagent water. No blank sample ever had a detectable amount of a pesticide. The lab blank was to ensure that the analysis itself was not adding contaminants to the samples. The equipment blank was to ensure that the pump and tubing were not contaminating the

sample. Finally, the travel blank ensured that the samples were not contaminated during transportation.

#### 5. Fortified Samples (Spiked Samples)

A spike was a sample that had pesticides added before it was extracted. For every twenty samples a lab fortified spike was analyzed. It was made of Milli-Q reagent water with internal standards and the pesticides. Additionally, matrix spikes were made from quadruplicate field samples. Two matrix spikes were made from each site's samples from April to September, 1992 and four from each site's samples after September. The internal standards were at 5  $\mu\text{g/L}$  and the pesticides were added at 5 and 20  $\mu\text{g/L}$  for the April samples. After April, the internal standards were added at 2  $\mu\text{g/L}$  while the pesticides were added at 0.25, 0.5, 1, 2, and 5  $\mu\text{g/L}$ . The pesticides were added at the same concentration in a sample, i. e. 0.25  $\mu\text{g/L}$ , but at different concentrations for different samples. Graphical correlation of the spikes is included in Appendix G.

#### 6. Recoveries of Surrogates

Initially phenanthrene- $\text{d}_{10}$  and acenaphthrene- $\text{d}_{10}$  were used as surrogates at 5  $\mu\text{g/L}$  in April 1992, and 2  $\mu\text{g/L}$  afterwards. However, they did not produce good peaks on the GC/ECD. Starting in August 1992, 4-chlorobiphenyl was added to all samples at 10  $\mu\text{g/L}$ . The recovery rate for it is listed for every sample from August to February 1993 in Appendix A. The

overall recovery rate for 4-chlorobiphenyl was 74%. By month, the recovery rate was 39%, 62%, 68%, 65%, 120%, and 90% for August, October, November, December, January, and February respectively. A standard curve was developed based upon multiple injections of 4-chlorobiphenyl standard solutions at four different concentrations. A curve was developed using least squares regression. Percent recovery was calculated with this curve by dividing the calculated 4-chlorobiphenyl concentration by the added concentration, 10  $\mu\text{g/L}$ , assuming 100% recovery.

#### **G. Statistical Analyses**

All statistical analysis was done using Borland's Quattro Pro version 4.0. The coefficients of the regressions performed were included in Appendix C.



## IV. Results and Discussion

### A. Seepage Rates

The average seepage rate for all the samples was 0.63 L/(m<sup>2</sup>\*hr). Individual seepage rates for each site were shown in Figure 12. Note that the seepage rates listed were the gross movement of water through the sediment. From salinity studies, only 35% of the water was estimated to be fresh water, with the other 65% being seawater that had migrated into the sediment (Simmons, et al., 1992).

The seepage rates varied with each sample, but the average discharges over the study period were 0.96, 0.74, 0.59, and 0.40 L/(m<sup>2</sup>\*hr) for Agricultural Sites 1, 2, 3, and 4, respectively. The two sites with the highest discharge, agricultural sites 1 and 2, also had soils with the most clay and silt. Both site's soil contained layers or pockets of sandier soils in between the clayey soils. This might have caused the seepage to be funneled into the sandier areas of soil, which would have increased the discharge rate.

The lowest or close to the lowest discharges for all sites were in November and December of 1992. The highest discharge was in February 1993 for two sites, 1 and 4, July 1992 for agricultural site 2, and October 1992 for site 3. The discharge appeared to follow the seasonal rainfall variations, with high flows in late spring of 1992 and late winter of 1993, and low discharges in the summer and late fall

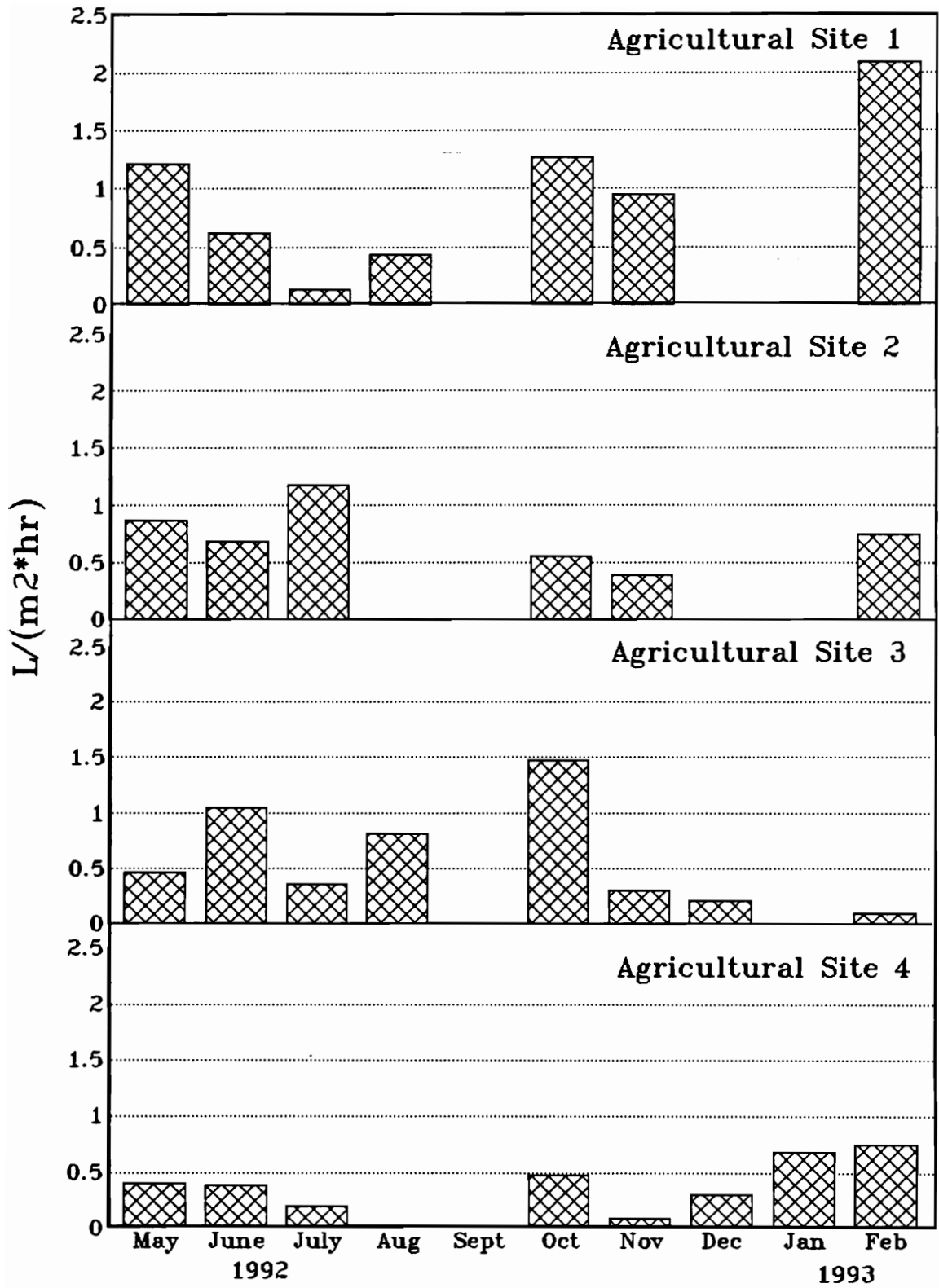


Figure 12  
Seepage Rates by Site

of 1992. It rained the day before and the first day of the October 1992 sampling trip, which probably increased the discharges.

Although all the seepage rates seemed small, the total area which groundwater seepage occurred is millions of square meters. It has been estimated that the total groundwater inflow into the Chesapeake Bay was approximately equal to a major tributary, such as the James River, and 30% of the nitrogen flux in the Chesapeake Bay occurred by the submarine groundwater discharge's nutrient content (Reay and Simmons, 1992).

#### **B. Nutrient Concentrations/Fluxes**

Since the study was done in cooperation with Dr. Simmons' groundwater and seepage water nutrient flux study, samples for the nutrient study were taken at the same time as the samples from the pesticide study. From the nutrient sample analyses, nutrients, especially nitrate, are present at high levels in the agricultural sites' groundwater, and seepage water samples often had high levels of nutrient transport. Several researchers have linked high nitrate concentrations with an increase probability of pesticide contamination (Spalding, *et al.*, 1979; Spalding, *et al.*, 1989; Domagalski and Dubrovsky, 1992; and Leonard, *et al.*, 1988). The entire nutrient sample data was listed in Appendix D.

##### **1. Upland well nutrient concentration**

A typical chromatogram of an upland well sample is included as Figure 13. The average nitrate concentration for all the upland well nutrient samples was 14.25 mg/L as N. The average nitrate concentration in the groundwater over the period of the study was 15.04, 14.32, 15.18, and 12.37 mg/L as N for agricultural sites 1, 2, 3, and 4 respectively (Figure 14). The EPA's maximum contaminate level (MCL) for nitrate is 10 mg/L as N. Therefore, the average nitrate concentration in the groundwater over the study period at all the agricultural sites was over the EPA's MCL. This indicated that the fertilizers applied to the fields were percolating into the groundwater. The survey from agricultural site 2 showed that fertilizer was applied in the fall of 1991 and spring of 1992. However, the nitrate concentrations are fairly steady in the samples varying from 13.09 to 16.60 mg/L as N. The survey from agricultural site 3 listed fertilizer applied on October 24, 1992 and March 10, 1993. The March application was after the study period ended, but the October 1992 sample was at the same level as the August 1992 sample. Apparently the nitrate traveled slowly down into the aquifer at a relatively constant rate.

Three other nutrients were studied in the groundwater samples: ammonia, nitrite, and phosphate. The ammonia concentrations were always at least two orders of magnitude below the nitrate levels. The ammonia concentrations ranged

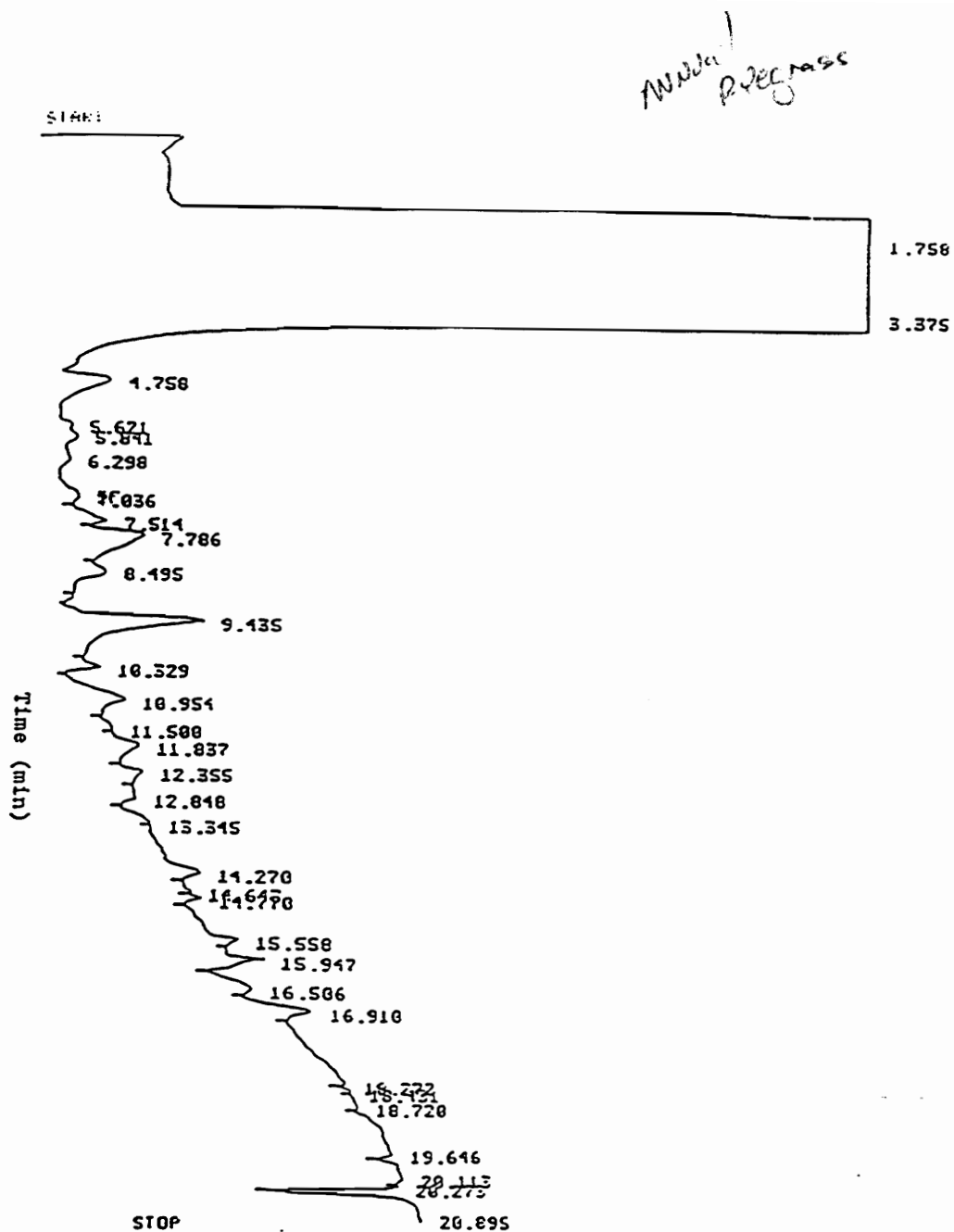


Figure 13

Typical Upland Well Chromatogram Using the GC/ECD (attn 2)  
 This chromatogram from WF2 #2 (No pesticides detected)

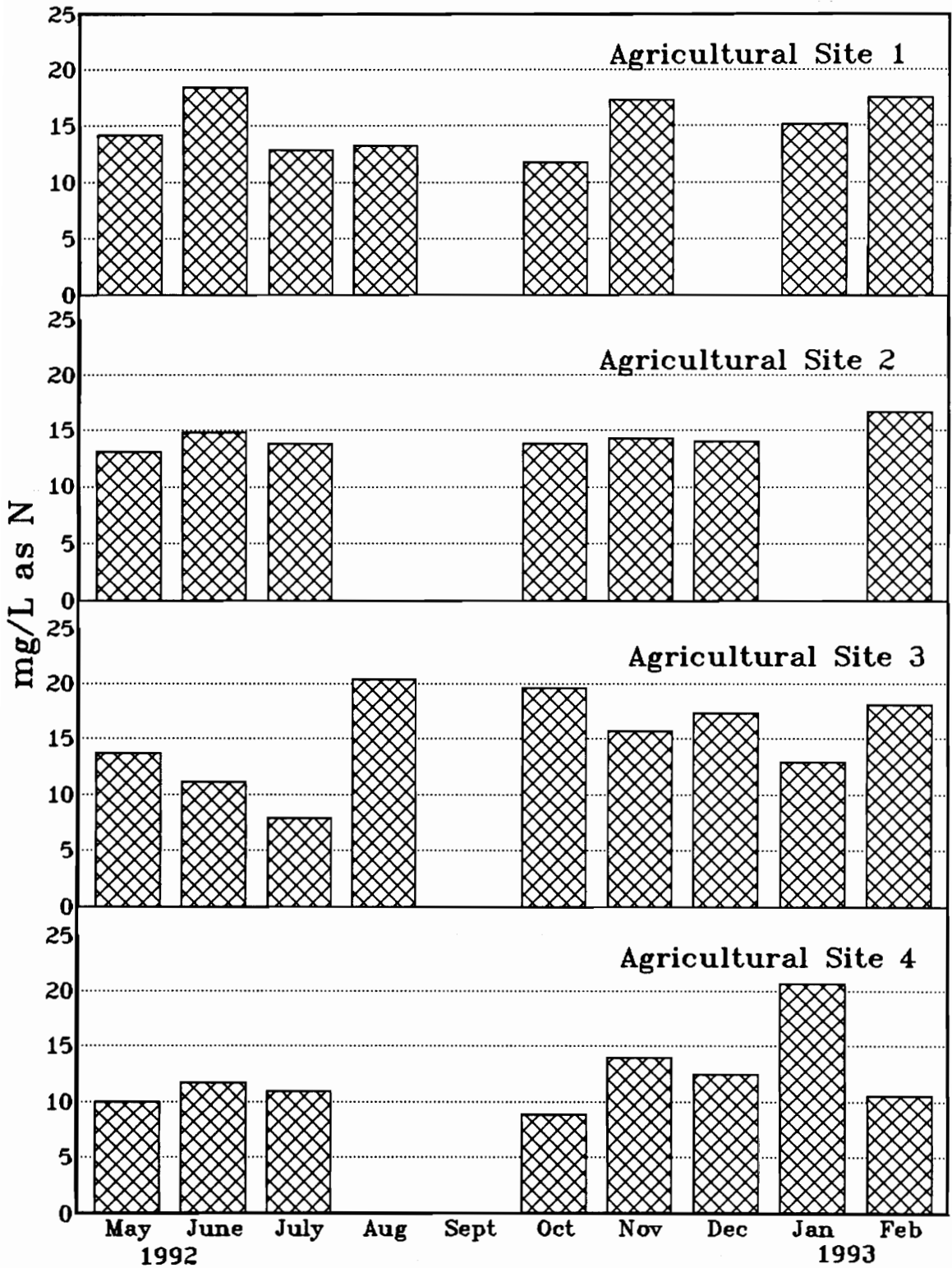


Figure 14  
Upland Well Nitrate Concentration

from 0.01 to 0.14 mg/L as N, with the highest ammonia concentration only 1.0% of the average nitrate level. The nitrite concentrations were similarly low compared to the nitrate ones. Nitrite values varied from below detection to 0.12 mg/L as N, which was again less than 1% of the average nitrate concentration. Average phosphate concentration for all the samples was 0.017 mg/L. The phosphate concentrations ranged from <0.001 to 0.093 mg/L. By site, the average phosphate concentrations were 0.011, 0.007, 0.032, and 0.011 mg/L for agricultural sites 1, 2, 3, and 4 respectively.

## 2. Seepage meter nitrate concentrations

The nitrate concentration in the seepage water was measured directly from the sample, Therefore, the concentrations were diluted by the seawater which entered the seepage meters. The seepage meter samples averaged 35% freshwater, but varied from <0.1 to >99% freshwater. The average nitrate concentration for all the seepage meter samples was 0.73 mg/L as N. The average nitrate concentrations at each site were: 1.50 mg/L as N at site 1, 0.41 mg/L as N at site 2, 0.59 mg/L as N at site 3, and 0.43 mg/L as N at agricultural site 4. Sites 1 and 3, showed larger than average concentrations in October and November 1992. The October 1992 sample at site 3, measured at 3.09 mg/L as N, was an order of magnitude greater than an other sample taken at the site. The highest nitrate concentration

was in a sample from agricultural site 1, with a 4.05 mg/L as N concentration in November 1992. Except site 3, all the sites also a high nitrate concentration in the February 1993 seepage meter samples (Figure 15).

The high nitrate concentration and nitrate flux in the October 1992 submarine groundwater discharge sample from site 3 correlated well with the upland groundwater sample. All three were the highest taken at the site during the study, which indicated that the nitrate was being transported from the field into Chesapeake Bay. However, a comparison of the all the upland well vs. the seepage meter nitrate concentrations had a poor correlation with an  $R^2 = 0.188$  (Figure 16).

### 3. Sediment nutrient fluxes

The nutrient fluxes were calculated by comparing the nutrient's concentration in the ambient seawater to its concentration in the sampling bag. The seepage meter was assumed to contain 6 L of seawater in its headspace when inserted into the sediment. The volume of the seepage bag plus 6 L was multiplied by the nitrate concentration in the bag to yield the mass of nitrate in the sample. The mass present at the installation of the seepage meter was calculated as 6 L times the ambient nutrient concentration. The difference, the sample mass minus the ambient mass, was divided by the product of the area of the seepage meter, 0.25



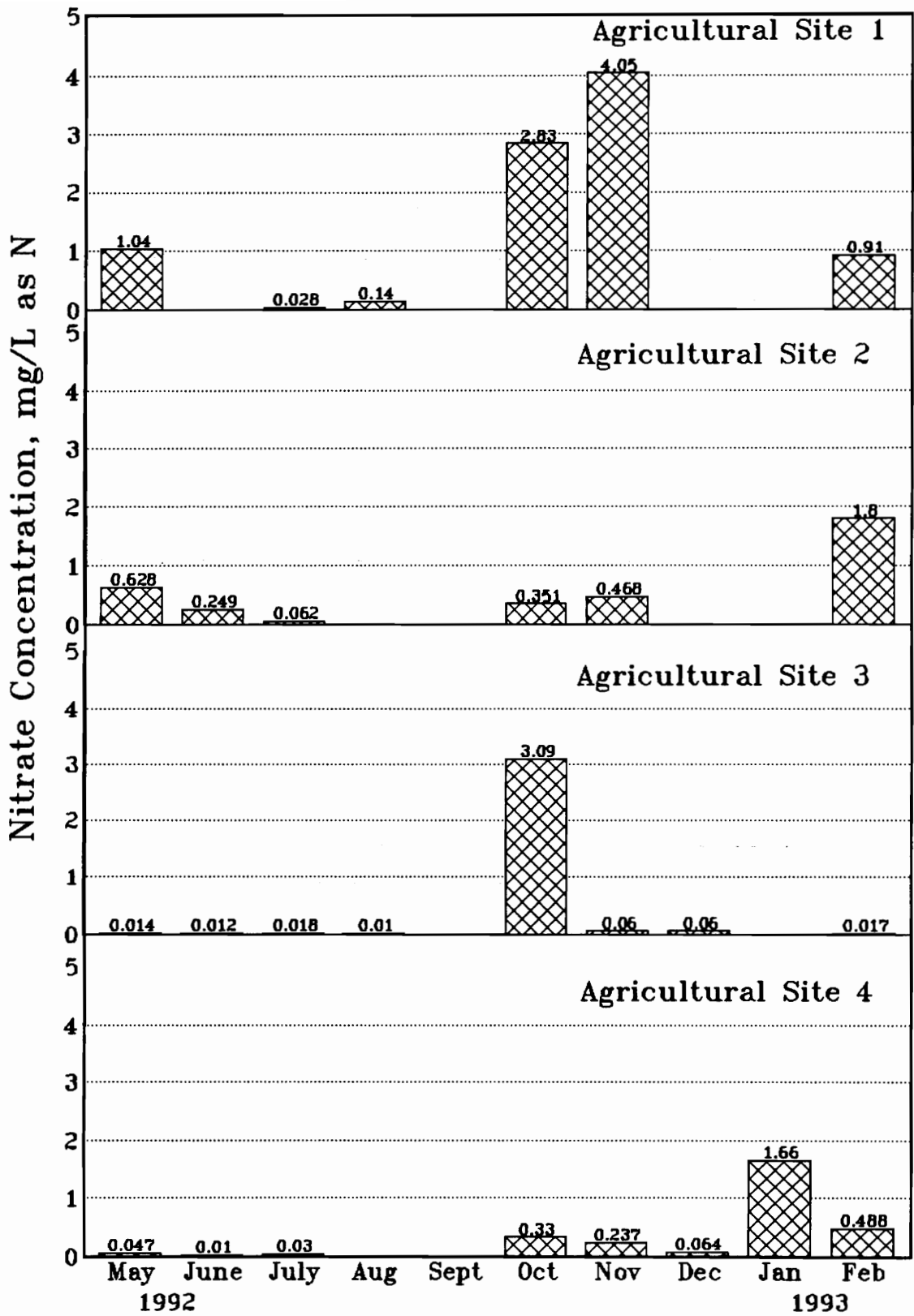


Figure 15  
Seepage Meter Nitrate Concentration

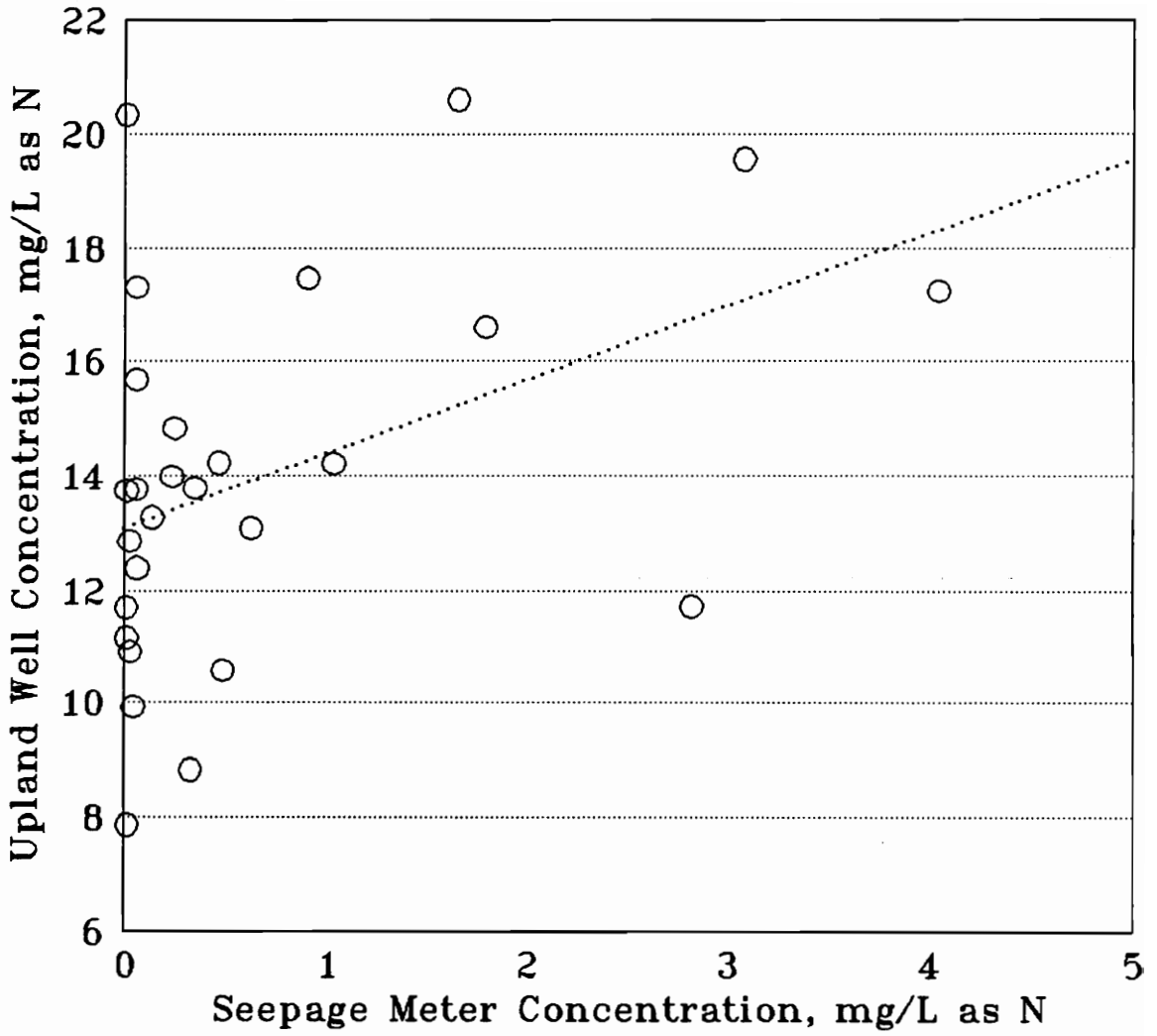


Figure 16  
 Seepage Meter Nitrate Concentration  
 vs.  
 Upland Well Nitrate Concentration

m<sup>2</sup>, and multiplied by the time between installation and removal of the bag to determine the flux.

The seepage meter nitrate fluxes varied widely from 20.15 mg/(m<sup>2</sup>\*hr), November 1992 at agricultural site 1, to several negative values, most at site 3 (Figure 17). The October 1992 samples from all the sites showed a high nitrate flux in the submarine groundwater discharge.

By plotting the sediment nitrate flux vs. the freshwater discharge rate and using linear regression to find the best fit line, the slope can be calculated. The slope of the regression line equals the nitrate concentration of the freshwater entering the seepage meter, which is the SGWD (Figure 18). Assuming the nitrate is conservative, the SGWD nitrate concentration is the same as the upland well nitrate concentration. The accuracy of the graphs was directly related to their coefficient of determination, R<sup>2</sup>. The graph of agricultural site 3 had the highest R<sup>2</sup> at 0.96, and its slope, 15.5 mg/L, was the closest to the actual average upland well concentration of 15.18 mg/L. Site 3 had the worst R<sup>2</sup> value at 0.24, and its slope was only a fourth of the actual upland well concentration (Table 10).

The sediment contained many types of bacteria and algae, which could consume and produce ammonia, nitrate, and phosphate. The average ammonia concentration in the seepage meter samples was 0.27 mg/L as N, which was approximately an

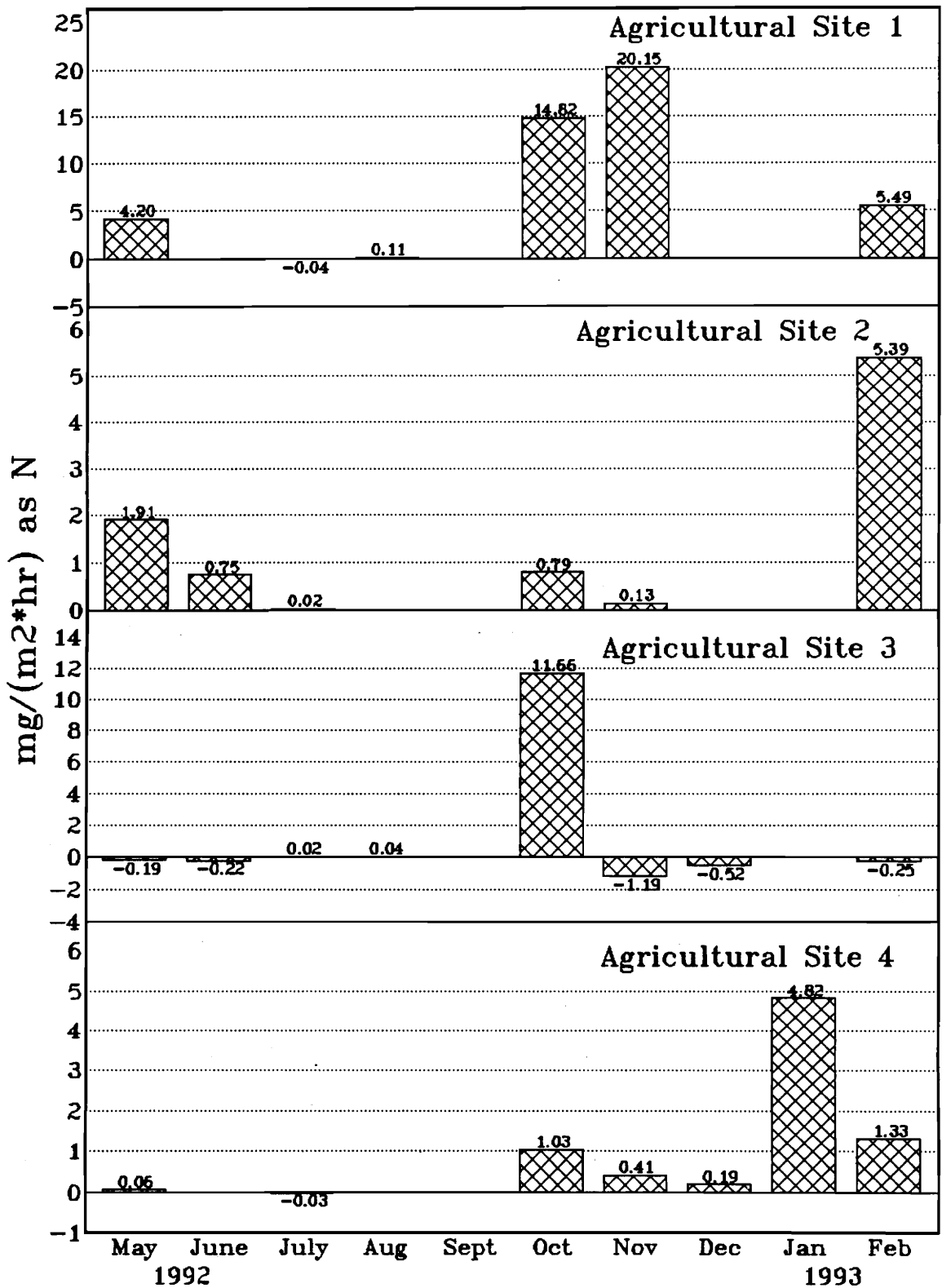


Figure 17  
 Sediment Nitrate Flux  
 Note the Y-axis are not to the same scale

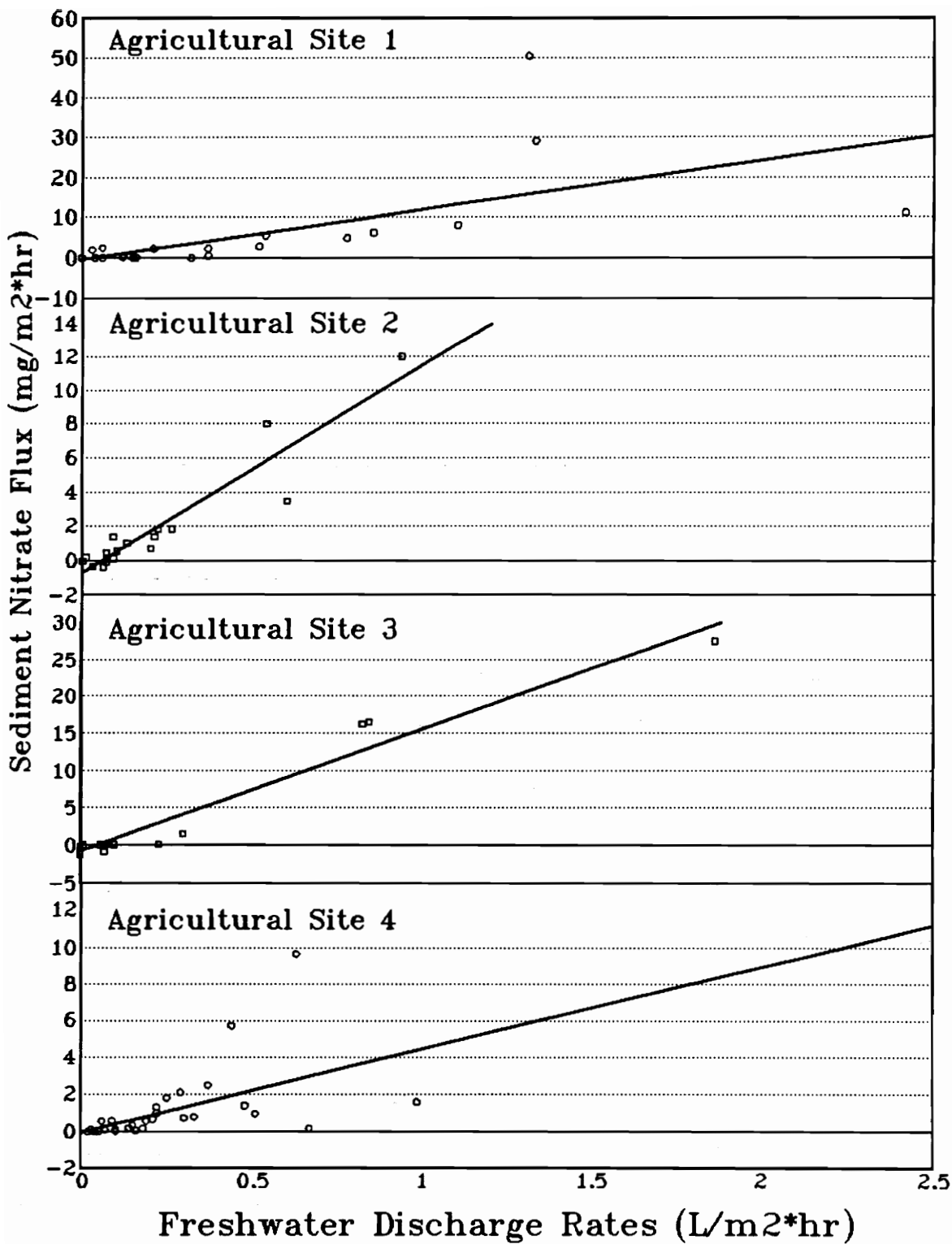


Figure 18  
Sediment Nitrate Flux vs. Freshwater Discharge Rate

Table 10

Groundwater Nitrate Concentrations

Site #	R <sup>2</sup>	Regression Slope, mg/L	Ave. Upland Nitrate concentration, mg/L
1	0.40	12.0	15.04
2	0.89	11.5	14.32
3	0.96	15.5	15.18
4	0.24	4.3	12.37

order of magnitude higher than the ammonia concentrations in the upland groundwater samples. Agricultural site 4 had the highest average ammonia concentration in the seepage meter samples at 0.85 mg/L as N, and it also had the lowest nitrate concentrations in seepage meter samples. Dr. Simmons showed that nitrification and denitrification occurred in some of the sediments at the study sites (Simmons, et al., 1992), which caused reduced or negative values in the seepage nutrient flux.

### **C. Pesticides**

For a pesticide to be reported as positively detected, the pesticide must have been detected above the instrumental LOD at the correct detention time using the GC/ECD on both the DB-5 and the DB-210 columns, and/or positively identified by GC/MS analysis. Due to the high LOD on the GC/MS, only two samples were positively identified as having a pesticide by GC/MS analysis. The two samples were the August CH3 #1 sample at site 2 for metolachlor, and the November BVCW #2 at agricultural site 1, also for metolachlor.

Of the over 500 samples analyzed on the GC, only 20 could be considered to have positive detection of a pesticide. Of these, 9 were quantified, with the other 11 being below the MDL (See Table 11).

The most common pesticide detected was cyanazine with 8. Metolachlor was next with 6; atrazine was detected 4 times,

**Table 11**  
**Summary of Pesticide Detections in Groundwater by GC**  
**Analysis**

Month	Virginia Tech ID	Site #	Pesticide Conc., µg/L				
			Atr	Ala	Carb	Cyan	Met
June	WF1 #1 <sup>a</sup>	4		0.66			
July	EV2 #1	3				0.16*	
	EVF/R #1	3				0.53	
	BV1 #1	1	0.59			0.16*	
	WF1 #1	4	0.63				
	WF2 #1	4	0.60				
	CH2 #1	2	1.52			0.17*	
Aug	BV2 #2	1					0.07*
	WFSW #1	4					0.07*
	CH3 #1	2					0.33
Oct	CHSM 1-4 combined	2					0.04*
Nov	BVCW #2	1					b
	CH2	2				0.02*	
	CH5 #1	2				0.55	
	EV2 #1	3				0.03*	
Dec	WFSW	4				0.02*	
Feb	BV1 #1	1		0.05*			0.04*
	CHSM 2	2					0.32

\* Below Method Detection Limit

a alachlor was also detected in a duplicate at 0.57 µg/L

b metolachlor was identified using the GC/MS, below the LOD



and alachlor was found twice. Carbofuran was not detected in any sample. This could be because it was not applied during the period of the study (not all of the surveys have been returned), or because of its high LOD on all the analytical instruments.

1. Upland well concentrations

Agricultural site 2 had the highest number of samples with pesticides detected with 7. Agricultural sites 1, and 4 had pesticides detected in 5 samples, and site 3 had pesticides detected in 3 samples.

Although agricultural site 1 did not return the survey, it had the highest number of pesticides detected in the soil samples. Alachlor was identified in 4 of the samples, June, July, October, and December 1992 at 145, 484, 162, and 19.8  $\mu\text{g}/\text{kg}$  respectively (Schicho, 1993). The site also had 4 pesticides detected in the groundwater samples. Metolachlor was detected, below the MDL, in the August and November 1992 samples. In the July 1992 BV1 #1 sample, Atrazine was detected at 0.59  $\mu\text{g}/\text{L}$ , and cyanazine was detected below the MDL.

According to the returned survey, agricultural site 2 had atrazine and metolachlor applied in the summer of 1991. Atrazine was detected in the groundwater in July, 1992, at 1.52  $\mu\text{g}/\text{L}$ , but not in any of the soil samples taken by Schicho (1993). He did find metolachlor in the June and October 1992,

and February 1993 soil samples, and metolachlor was detected in the groundwater in August 1992 and February 1993, below the MDL. Also, cyanazine was detected in the groundwater at 0.17  $\mu\text{g/L}$  in July 1992, even though it was not listed as being applied. Agricultural site 2 also had anoxic groundwater for much of the study, which could explain why, even a year after application the two pesticides, atrazine and metolachlor, were still detected. The half-life for atrazine has been estimated between 20 and 140 days, and metolachlor's half-life was estimated at 20 to 90 days (Leonard, et al., 1988; Spalding, et al., 1989; Thurman, et al., 1992), although Bouchard found metolachlor still detectable after 2 years in some soil samples (Bouchard, et al., 1982).

Agricultural site 3's survey stated that they did not apply any of the pesticides which were monitored in the study. However, cyanazine was detected in the groundwater in 3 samples, two in July and one in November of 1992. The July 1992 EVF/R #1 concentration was 0.53  $\mu\text{g/L}$ , but the other two concentrations were below the MDL. Two soil samples from the site contained metolachlor, June at 26.9  $\mu\text{g/kg}$  and December at 7.4  $\mu\text{g/kg}$ , but the concentrations were closed to the LOD for the pesticide.

The survey from agricultural site 4 was not returned. In the groundwater, metolachlor was detected, below the MDL, in the August 1992. Also, alachlor was detected twice in June

1992, at 0.57  $\mu\text{g/L}$  in WF1 #2 and 0.66  $\mu\text{g/L}$  in WF2 #1, and atrazine was detected in WF1 #1 and WF2 #1 at 0.63 and 0.60  $\mu\text{g/L}$ , respectively, in July 1992. Metolachlor was detected in the December and January soils samples at 7.1 and 8.7  $\mu\text{g/kg}$  respectively.

No pesticides were detected in any of the samples for wetlands site 1.

Since cyanazine was not reported as being applied to any of the sites, but was detected in 5 samples, the possibility of false positive detection was a real concern. Cyanazine was eluted at the same time as metolachlor with the DB-5 column. While the DB-210 column eluted cyanazine at a much later time than metolachlor (19.2 vs. 15.9 minutes), it was still possible that the cyanazine "detections" could have been metolachlor with a contaminate on the DB-210 column at the right time, or a contaminate with the same retention time as cyanazine.

## 2. Seepage meter concentrations

The seepage water samples had only 2 sample with a pesticide detected. Metolachlor was detected, below the MDL, in the October CHSM1-4 combined and the February CHSM 2 samples at site 2. All the other seepage meter samples did not have any pesticides above the LOD. Most of the seepage meter samples were very dirty after extraction, with dozens of GC peaks (Figure 18). The presence of so many compounds could

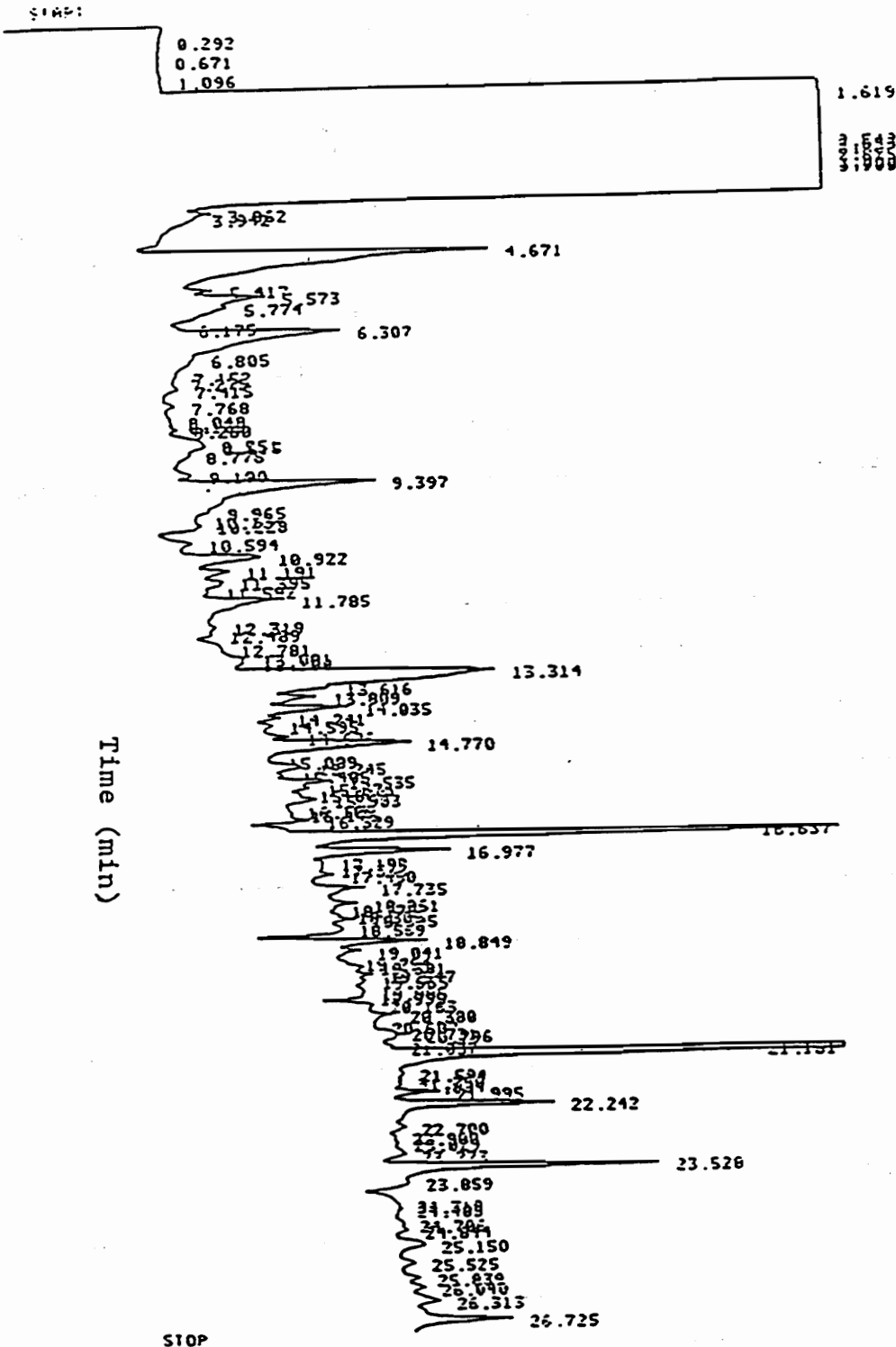


Figure 19

Typical Seepage Meter Chromatogram Using the GC/ECD (atta 2)

This was BVSM 1&3 Combined (No pesticides detected)

have masked the any of the pesticides present. Due to the low number of pesticides detected in the cleaner groundwater samples, however, few, if any, detectable pesticides are expected in the seepage. Also, the soil data supports the theory that most of the pesticides were bound to the soil, rather than migrating into the groundwater. Finally, the seepage samples were only about 35% groundwater, with the rest being infiltrated seawater. This 3:1 dilution ratio may have caused any pesticides present in the seepage water to be below the LOD.

The Ohmicron results listed 20 pesticides detected in the seepage meter samples. However, Ohmicron listed three seepage meter samples with three separate pesticides, atrazine, alachlor, and cyanazine, detected in each sample, and three seepage meter samples each with two pesticides, atrazine and alachlor, detected. All the seepage meter samples with multiple pesticides detected were in August 1992. The only pesticide detected by Ohmicron in all the other seepage meter samples was atrazine.

### 3. - Ohmicron Concentrations

Ohmicron Corporation reported 63 detections of the five pesticides in the study over their LOD in the samples they analyzed (See Table 12). The highest pesticide concentration was 4.46  $\mu\text{g/L}$  in the November 1992 BVCW sample. The only other pesticide detected over 1  $\mu\text{g/L}$  was the January 1993 BVCW

Table 12  
Pesticide Detections by Ohmicron Corporation

Site	Number of times the pesticide was detected				
	Atr	Ala	Carb	Cyan	Metol
Ag. site 1	5	23	0	0	2
Ag. site 2	8	2	1	2	1
Ag. site 3	0	1	0	0	0
Ag. site 4	6	3	0	2	0
Wet. site 1	0	0	0	0	0

sample at 1.22  $\mu\text{g}/\text{L}$ . The samples from agricultural site 1 had 30 of the detections, 23 of them for alachlor, which was also found in 4 soil samples. Site 2 samples were next with 14 detections, then site 4 samples with 11, and finally site 3 samples had only 1 pesticide detected. The complete results were included in Appendix F.

Although none of the blank samples had any pesticides detected using gas chromatography, Ohmicron data stated that two of the equipment blanks and two of the travel blanks contained atrazine. The concentrations reported, 0.05, 0.06, 0.07, and 0.07  $\mu\text{g}/\text{L}$ , were well under the LOD for the GC analysis. The equipment blanks could have been contaminated by a pesticide adsorbed to the tubing used to pump the samples out of the wells. The travel blank detections were unexpected. Since the same milli-Q water was used for all the blanks, the laboratory blanks should have had the contaminate too, but no detections were reported for the laboratory blanks.

Most of Ohmicron's reported pesticide detections were below the LOD of the GC analysis. The detections could have been either a low concentration of the pesticides in the groundwater, which has been reported by several researchers (Holden, *et al.*, 1992; Spalding, *et al.*, 1989; Steichen, *et al.*, 1988), or false positives, which have been reported for alachlor and metolachlor using the immunoassay procedure

(Baker, et al., 1993). In the study, the 1 and 2  $\mu\text{g/L}$  laboratory and matrix spikes from October 1992 through February 1993 did not have metolachlor added, yet Ohmicron listed 18 of the spiked samples as having metolachlor present. The GC analysis did not detect any metolachlor in the 1 and 2  $\mu\text{g/L}$  spikes. The immunoassay procedure was significantly faster and easier to perform than the solid phase extraction, but it appeared to general more false positives.

#### 4. Soil sample concentrations

Soil samples were taken at the sites at the same time as the groundwater samples. A master's candidate analyzed them using a soxhlex extraction process with identification on the GC/MS. Mr. Schicho found pesticides in about a fifth of all the soil samples taken. However, half, 14 of 28, of the field soil samples contained a detectable amount of a pesticide, while none of the sediment samples had a detectable pesticide. The largest concentration was 484  $\mu\text{g/kg}$  of Alachlor, in the July 1992 field sample from agricultural site 1 (Schicho, 1993). Complete soil sample data was included in Appendix E.

The field soil samples had a much higher concentration of pesticides than the water samples. The average concentration of all the pesticides detected in the soil samples was approximately 82  $\mu\text{g/kg}$  (Schicho, 1993), The groundwater samples were at much lower concentrations. Only one of the



detections was over 1 ppb, which was July CH2 #1 at 1.5 ppb, with the second highest being 0.63 ppb, July WF2 #1. This suggests that the pesticides in the study were being mostly adsorbed to the soils, and not transported by the groundwater at a concentration above 1  $\mu\text{g/L}$ .

## 5. Spiking Results

For all spikes, the pesticide recovered with the greatest precision was alachlor, and carbofuran had the most variability. All the spikes' measured values had a large variability, but the values were generally within 50% of the added concentration, except for carbofuran's measured concentration, which was often several times the added value. Since carbofuran was the only non-chlorinated pesticide in the study, it had much poorer resolution on the GC/ECD than the other pesticides. Consequently, carbofuran was not detected in many of the spiked samples. Spiked sample results were also discussed in Appendix G.

### D. Contaminates

The most common contaminate found was di-ethyl phthalate which was found in most samples, with several other phthalates also observed in many samples. In the summer and early fall of 1992, many samples contained N,N-diethyl toluamide (DEET), which was the active ingredient of the insect repellent Off!™. This was surprising because the insect repellent was never sprayed in the area where the samples were taken, although it

was used at the eastern shore laboratory, and equipment handling was minimized. Petroleum products were detected in several of the samples from agricultural site 4 in the summer of 1992, but later samples contained no detectable oil products. Also, most of the seepage meter samples and several of the groundwater samples contained hexanedioic acid dioctyl ester. A doctoral candidate, who was working with algae, said that hexanedioic acid was a byproduct of some types of algae (Rashash, 1992).

## V. Conclusions

The purpose of the study was to quantify pesticides in the groundwater of the Chesapeake Bay area and determine if pesticides were being transported by the groundwater into the Chesapeake Bay. Solid phase liquid extraction followed by gas chromatography was used to analysis the water samples, and an independent company, Ohmicron Corporation, also analyzed samples using an immunoassay analysis. Pesticides were analyzed in groundwater, SGWD, and soil samples. The study also examined groundwater transport of nutrients from the sites into the Chesapeake Bay.

From the results of this study, the following conclusions were derived:

1. Alachlor, atrazine, carbofuran, cyanazine, and metolachlor were not detected in the groundwater in most (96%) of the samples analyzed by gas chromatography.

2. In all the samples where a pesticide was detected, the concentration of the pesticide was below 1  $\mu\text{g}/\text{l}$ , except for one sample.

3. Half of the field soil samples contained a detectable pesticide with an average concentration of 82  $\mu\text{g}/\text{kg}$ . It appeared that most of the pesticides were adsorbed to the soils, with a small amount dissolved into the groundwater.

5. The submarine groundwater discharge of the studied

pesticides into the Chesapeake Bay was too dilute to find with the method used.

6. The Ohmicron immunoassay analysis had a lower LOD for the pesticides than solid phase liquid extraction, but appeared to generate more false positive detections.

7. Nutrients, especially nitrate, were being transported by the groundwater into the Chesapeake Bay.

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## Appendix A

### List of All Water Samples

Every sample extracted is listed in tables 1A through 10A. The tables corresponded to the month the samples were taken, with Table 1A being April, 1992 and Table 10A being February, 1993. Numbers with an asterix (\*) are below the MDL for the pesticide.

#### Abbreviations

BV, BR: Agricultural site 1

CH: Agricultural site 2

EV: Agricultural site 3

WF, AH: Agricultural site 4

ST: Wetlands site 1

EB: Equipment blank

TB: Travel blank

LB: Laboratory blank

LFB: Laboratory Fortified blank (pesticides added)

(%) Res: The percent response of 4-chlorobiphenyl in the sample after injection on the GC, compared to an injection of an equal amount of the standard of 4-chlorobiphenyl on the GC.

\*\* : The 4-chlorobiphenyl peak was masked, or not present.

ND: Not Detected

N/A: Not applicable, the pesticide was not added to the sample.

Table 1A  
April Samples

Name	Type	Comments	Volume	Measured Conc. (µg/L)			
				Atr	Ala	Carb	Cyan
EV7 #1	well	Spiked @ 20 ppb	1 L	1.24	1.77	1.88	0.41
EV7 #2	well	Spiked @ 5 ppb	1 L	0.17*	0.25*	0.34*	ND
EV7 #3	well		1 L	ND	ND	ND	ND
EV7 Field #4	well		1 L	ND	ND	ND	ND
EV1 #1	well		1 L	ND	ND	ND	ND
EV1 #2	well	duplicate of EV1 #1	1 L	ND	ND	ND	ND
ESGS1 #1	well		1 L	ND	ND	ND	ND
ESGS1 #2	well	duplicate of ESGS1	1 L	ND	ND	ND	ND
ESGS2 #1	well		1 L	ND	ND	ND	ND
ESGS2 #2	well	duplicate of ESGS2	1 L	ND	ND	ND	ND
WF #1	well	Spiked @ 5 ppb	1 L	0.27*	0.33	0.56*	ND
WF #2	well	Spiked @ 20 ppb	1 L	1.58	2.01	2.40	1.03
WF #3	well	duplicate of WF #2	1 L	ND	ND	ND	ND
WF #4	well	duplicate of WF #2	1 L	ND	ND	ND	ND
WUP #1	well		1 L	ND	ND	ND	ND
WUP #2	well	duplicate of WUP #1	1 L	ND	ND	ND	ND
CH2 #1	well		1 L	ND	ND	ND	ND
CH2 #2	well	duplicate of CH2 #1	1 L	ND	ND	ND	ND
CH2 #3	well	duplicate of CH2 #1	1 L	ND	ND	ND	ND
CH2 #4	well		1 L	ND	ND	ND	ND

Table 1A continued

Name	Type	Comments	Volume	Measured Conc. ( $\mu\text{g/L}$ )				
				Atr	Ala	Carb	Cyan	
CH5 #1	well		1 L	ND	ND	ND	ND	ND
CH5 #2	well	duplicate of CH5 #1	1 L	ND	ND	ND	ND	ND
AH1 #1	well		1 L	ND	ND	ND	ND	ND
AH1 #2	well	duplicate of AH1 #1	1 L	ND	ND	ND	ND	ND
AH1 #3	well	Spiked @ 5 ppb	1 L	ND	0.25*	0.07*	1.03	
AH1 #4	well	Spiked @ 20 ppb	1 L	1.65	1.91	2.59	1.21	
AH Beach #1	well		1 L	ND	ND	ND	ND	ND
AH Beach #2	well	duplicate of AH Beach #1	1 L	ND	ND	ND	ND	ND
AH Beach #3	well	Spiked @ 5 ppb	1 L	0.06*	0.52	0.09*	ND	
AH Beach #4	well	Spiked @ 20 ppb	1 L	0.43	1.15	1.66	ND	
LH Tap #1	well		1 L	ND	ND	ND	ND	ND
BR9 #1	well		1 L	ND	ND	ND	ND	ND
BR9 #2	well	duplicate of BR9 #1	1 L	ND	ND	ND	ND	ND
BR15 #1	well		1 L	ND	ND	ND	ND	ND
BR15 #2	well	duplicate of BR15 #2	1 L	ND	ND	ND	ND	ND
BR15 #3	well	Spiked @ 5 ppb	1 L	0.07*	0.44	0.41*	ND	
BR15 #4	well	Spiked @ 20 ppb	1 L	1.08	1.90	1.88	ND	
LB #1	blank	Reagent water	1 L	ND	ND	ND	ND	ND
TB #1	blank	Travel blank	1 L	ND	ND	ND	ND	ND
TB #2	blank	duplicate of TB #1	1 L	ND	ND	ND	ND	ND



Table 1A Continued

Name	Type	Comments	Volume	Measured Conc. (µg/L)			
				Atr	Ala	Carb	Cyan
EB #1	blank	equipment blank	1 L	ND	ND	ND	ND
EB #2	blank	duplicate of EB #1	1 L	ND	ND	ND	ND

Table 2A  
May Samples

Name	Type	Comments	Volume	Measured Conc. (µg/L)			
				Atr	Ala	Carb	Cyan
WF #1	well		1 L	ND	ND	ND	ND
WF #2	well	Spiked @ 2 ppb	1 L	0.32*	0.61	0.70*	ND
WF #3	well	duplicate of WF #1	1 L	ND	ND	ND	ND
WFSW	well		1 L	ND	ND	ND	ND
WFSM 1-4	SM		1 L	ND	ND	ND	ND
WF Amb To	sea		1 L	ND	ND	ND	ND
CH New #1	well		1 L	ND	ND	ND	ND
CH New #2	well	Spiked @ 2 ppb	1 L	0.77	1.18	1.67	ND
CH New #3	well	Spiked @ 10 ppb	1 L	3.54	4.57	7.69	0.92
CH2	well		1 L	ND	ND	ND	ND
CH5	well		1 L	ND	ND	ND	ND
CHSM 3	SM		1 L	ND	ND	ND	ND
CHSM 4	SM	Spiked @ 10 ppb	1 L	4.75	5.64	11.1	4.55
CH Amb To	sea		1 L	ND	ND	ND	ND
EV2 #1	well		1 L	ND	ND	ND	ND
EV2 #2	well	Spiked @ 2 ppb	1 L	0.37	0.76	1.09*	ND
EV2 #3	well	duplicate of EV2 #1	1 L	ND	ND	ND	ND
EV1 #1	well	Spiked @ 2 ppb	1 L	ND	1.28	0.51*	ND
EV1 #2	well		1 L	ND	ND	ND	ND
EV1 #3	well	duplicate of EV1 #2	1 L	ND	ND	ND	ND

Table 2A continued

Name	Type	Comments	Volume	Measured Conc. ( $\mu\text{g/L}$ )			
				Air	Ala	Carb	Cyan
EVF/R #1	well		1 L	ND	ND	ND	ND
EVF/R #2	well	duplicate of EVF/R #1	1 L	ND	ND	ND	ND
EVSM 1&4	SM		800 mL	ND	ND	ND	ND
EVSM 2	SM		900 mL	ND	ND	ND	ND
EV Amb To	sea		1 L	ND	ND	ND	ND
BVCW #1	well	Spiked @ 2 ppb	1 L	0.66	1.06	1.67	ND
BVCW #2	well		1 L	ND	ND	ND	ND
BVCW #3	well	duplicate of BVCW #2	1 L	ND	ND	ND	ND
BV1 #1	well		1 L	ND	ND	ND	ND
BV1 #2	well	duplicate of BV1 #1	1 L	ND	ND	ND	ND
BV2 #1	well		1 L	ND	ND	ND	ND
BV2 #2	well	duplicate of BV2 #1	1 L	ND	ND	ND	ND
BVSM1	SM		1 L	ND	ND	ND	ND
BVSM2	SM		1 L	ND	ND	ND	ND
BVSM3 #1	SM		1 L	ND	ND	ND	ND
BVSM3 #2	SM	duplicate of BVSM3 #1	1 L	ND	ND	ND	ND
BVSM3 #3	SM	duplicate of BVSM3 #1	1 L	ND	ND	ND	ND
BVSM 4	SM		1 L	ND	ND	ND	ND
BV Amb To	sea		1 L	ND	ND	ND	ND
EB #1	blank	equipment blank	1 L	ND	ND	ND	ND

Table 2A continued

Name	Type	Comments	Volume	Measured Conc. (µg/L)				
				Atr	Ala	Carb	Cyan	
EB #2	blank	duplicate of EB#1	1 L	ND	ND	ND	ND	
TB #1	blank		1 L	ND	ND	ND	ND	
TB #2	blank	duplicate of TB #1						
LB #1	blank	Reagent blank	1 L	ND	ND	ND	ND	
LFB #2	blank	Spiked @ 2 ppb	1 L	0.53	0.80	0.91*	ND	

Table 3A  
June Samples

Name	Type	Comments	Volume	Measured Conc. (µg/L)				
				Atr	Ala	Carb	Cyan	
EV1 #1	well		1 L	ND	ND	ND	ND	
EV1 #2	well	duplicate of EV1 #1	1 L	ND	ND	ND	ND	
EV2 #1	well		1 L	ND	ND	ND	ND	
EV2 #2	well	duplicate of EV2 #1	1 L	ND	ND	ND	ND	
EVF/R #1	well	Spiked @ 5 ppb	1 L	1.86	3.05	2.36	1.86	
EVF/R #2	well		1 L	ND	ND	ND	ND	
EVF/R #3	well	duplicate of EVF/R #2	1 L	ND	ND	ND	ND	
EVF/R #4	well	Spiked @ 2 ppb	1 L	ND	0.71	0.60*	0.34	
EVSM 1	SM		1 L	ND	ND	ND	ND	
EVSM 2&4 combined	SM		1 L	ND	ND	ND	ND	
EVSM 3	SM		1 L	ND	ND	ND	ND	
EV Amb To	Sea	seawater at site of SM	1 L	ND	ND	ND	ND	
BV1 #1	well		1 L	ND	ND	ND	ND	
BV1 #2	well	duplicate of BV1 #1	1 L	ND	ND	ND	ND	
BV2 #1	well		1 L	ND	ND	ND	ND	
BVCW #1	well	Spiked @ 2 ppb	1 L	0.29*	2.09	0.93*	1.08	
BVCW #2	well	Spiked @ 5 ppb	1 L	4.77	5.94	4.78	4.61	
BVCW #3	well		1 L	ND	ND	ND	ND	
BVCW #4	well	duplicate of BVCW #3	1 L	ND	ND	ND	ND	

Table 3A Continued

Name	Type	Comments	Volume	Measured Conc. ( $\mu\text{g/L}$ )			
				Atr	Ala	Carb	Cyan
BVSM 1 #1	SM		1 L	ND	ND	ND	ND
BVSM 1 #2	SM		1 L	ND	ND	ND	ND
BVSM 1 #3	SM		600 mL	ND	ND	ND	ND
BVSM 3	SM		1 L	ND	ND	ND	ND
BVSM 4	SM	Some extract spilled	1 L	ND	ND	ND	ND
BV Amb To	Sea	Seawater at site of SM	1 L	ND	ND	ND	ND
WF1 #1	well		1 L	ND	0.66	ND	ND
WF1 #2	well	duplicate of WF1 #1	1 L	ND	0.57	ND	ND
WF2 #1	well	Spiked @ 2 ppb	1 L	ND	0.68	ND	ND
WF2 #2	well		1 L	ND	ND	ND	ND
WFSW #1	well	Spiked @ 5 ppb	1 L	2.62	4.38	2.51	2.76
WFSW #2	well	Spiked @ 2 ppb	1 L	ND	1.07	ND	0.35
WFSW #3	well		1 L	ND	ND	ND	ND
WFSW #4	well	duplicate of WFSW #3	1 L	ND	ND	ND	ND
WFSM 3	SM		1 L	ND	ND	ND	ND
WFSM 4	SM		1 L	ND	ND	ND	ND
CH2 #1	well	Spiked @ 5 ppb	1 L	3.88	4.97	3.63	3.53
CH2 #2	well	Spiked @ 2 ppb	1 L	2.22	3.12	1.92	1.90
CH2 #3	well	Spiked @ 1 ppb	1 L	2.34	2.12	ND	1.14
CH2 #4	well		1 L	ND	ND	ND	ND

Table 3A Continued

Name	Type	Comments	Volume	Measured Conc. ( $\mu\text{g/L}$ )			
				Atr	Ala	Carb	Cyan
CH New #1	well		1 L	ND	ND	ND	ND
CH New #2	well	duplicate of CH New #1	1 L	ND	ND	ND	ND
CHSM 2	SM		1 L	ND	ND	ND	ND
CHSM 3	SM		800 mL	ND	ND	ND	ND
CH Amb To	Sea	Seawater at site of SM	1 L	ND	ND	ND	ND
TB #1	blank	Travel blank	1 L	ND	ND	ND	ND
TB #2	blank	duplicate of TB #1	1 L	ND	ND	ND	ND
EB #1	blank	equipment blank	1 L	ND	ND	ND	ND
EB #2	blank	duplicate of EB #1	1 L	ND	ND	ND	ND

Table 4A  
July Samples

Name	Type	Comments	Volume	Measured Conc. (µg/L)			
				Atr	Ala	Carb	Cyan
EV1 #1	well		1 L	ND	ND	ND	ND
EV1 #2	well	duplicate of EV1 #1	1 L	ND	ND	ND	ND
EV1 #3	well	Spiked @ 2 ppb	1 L	3.58	3.51	ND	1.99
EV1 #4	well	Spiked @ 4 ppb-some of sample spilled	1 L	1.01	5.22	ND	2.53
EV2 #1	well		1 L	ND	ND	ND	0.16*
EV2 #2	well	duplicate of EV2 #1	1 L	ND	ND	ND	ND
EVF/R #1	well		900 mL	ND	ND	ND	0.53
EVSM 1&2 Combined	SM		1 L	ND	ND	ND	ND
EVSM 2	SM		1 L	ND	ND	ND	ND
EV Amb To	Sea	Seawater at site of SM	1 L	ND	ND	ND	ND
BV1 #1	well		1 L	0.59	ND	ND	0.16*
BV1 #2	well	duplicate of BV1 #1	1 L	ND	ND	ND	ND
BV1 #3	well	duplicate of BV1 #1	1 L	ND	ND	ND	ND
BV1 #4	well	duplicate of BV1 #1	1 L	ND	ND	ND	ND
BV2 #1	well		1 L	ND	ND	ND	ND
BV2 #2	well	duplicate of BV2 #1	1 L	ND	ND	ND	ND
BV2 #3	well	Spiked @ 2 ppb	1 L	0.39	4.60	0.85*	0.14*
BV2 #4	well	Spiked @ 4 ppb	1 L	5.11	4.98	0.85*	3.64
BVCW #1	well		1 L	ND	ND	ND	ND



Table 4A Continued

Name	Type	Comments	Volume	Measured Conc. ( $\mu\text{g/L}$ )			
				Atr	Ala	Carb	Cyan
BVCW #2	well	duplicate of BVCW #1	1 L	ND	ND	ND	ND
BV Amb To	Sea	Seawater at site of SM	1 L	ND	ND	ND	ND
WF1 #1	well		1 L	0.63	ND	ND	ND
WF1 #2	well	duplicate of WF1 #1	1 L	ND	ND	ND	ND
WF1 #3	well	Spiked @ 2 ppb	1 L	1.88	3.22	ND	1.70
WF1 #4	well	Spiked @ 4 ppb	1 L	3.79	4.59	ND	3.17
WF2 #1	well		1 L	0.60	ND	ND	ND
WF2 #2	well	duplicate of WF2 #1	1 L	ND	ND	ND	ND
WFSW #1	well		1 L	ND	ND	ND	ND
WFSW #2	well	duplicate of WFSW #1	1 L	ND	ND	ND	ND
WF Amb To	Sea	seawater at site	1 L	ND	ND	ND	ND
CH2 #1	well		1 L	1.52	ND	ND	0.17*
CH2 #2	well	duplicate of CH2 #1	1 L	ND	ND	ND	ND
CH5 #1	well		1 L	ND	ND	ND	ND
CH5 #2	well	duplicate of CH5 #1	1 L	ND	ND	ND	ND
CH5 #3	well	Spiked @ 2 ppb	1 L	2.15	2.35	2.75	1.21
CH5 #4	well	Spiked @ 4 ppb	1 L	3.00	3.93	3.60	2.68
CH New	well		1 L	ND	ND	ND	ND
CHSM 1	SM		1 L	ND	ND	ND	ND
CHSM 2	SM		1 L	ND	ND	ND	ND

Table 4A Continued

Name	Type	Comments	Volume	Measured Conc. ( $\mu\text{g/L}$ )			
				Atr	Ala	Carb	Cyan
CHSM 3	SM		700 mL	ND	ND	ND	ND
CHSM 4	SM		1 L	ND	ND	ND	ND
CH Amb To	Sea	Seawater at site of SM	1 L	ND	ND	ND	ND
ST Last	well		1 L	ND	ND	ND	ND
STSM 1	SM		1 L	ND	ND	ND	ND
STSM 2	SM		1 L	ND	ND	ND	ND
STSM 4	SM		1 L	ND	ND	ND	ND
STSM 5	SM		1 L	ND	ND	ND	ND
STSM 6	SM		1 L	ND	ND	ND	ND
LFB #1	blank	Spiked @ 2 ppb	1 L	1.06	2.84	ND	1.70
LB #2	blank	Reagent blank	1 L	ND	ND	ND	ND
LB #3	blank	duplicate of LB #2	1 L	ND	ND	ND	ND
TB #1	blank	Spiked @ 2 ppb	1 L	2.77	3.98	ND	2.55
TB #2	blank	Travel blank	1 L	ND	ND	ND	ND
EB #1	blank	Spiked @ 2 ppb	1 L	0.37	1.08	1.70	0.38
EB #2	blank	equipment blank	1 L	ND	ND	ND	ND

Table 5A  
August Samples

Name	Type	Comments	Volume	(% Res	Measured Conc. (µg/L)				
					Atr	Ala	Carb	Cyan	Met
EV1 #1	well		1 L	49	ND	ND	ND	ND	ND
EV1 #2	well	duplicate of EV1 #1	1 L	27	ND	ND	ND	ND	ND
EV2 #1	well		1 L	39	ND	ND	ND	ND	ND
EV2 #2	well	duplicate of EV2 #1	1 L	66	ND	ND	ND	ND	ND
EV2 #3	well	Spiked @ 2 ppb	1 L	51	3.07	2.50	ND	7.10	ND
EVF/R #1	well		1 L	22	ND	ND	ND	ND	ND
EVF/R #2	well	duplicate of EVF/R #1	1 L	**	ND	ND	ND	ND	ND
EVSM 1	SM		1 L	62	ND	ND	ND	ND	ND
EVSM 2	SM		700 mL	46	ND	ND	ND	ND	ND
EVSM 3	SM		1 L	41	ND	ND	ND	ND	ND
EVSM 4	SM		800 mL	28	ND	ND	ND	ND	ND
EV Amb To	Sea	Seawater at site of SM	1 L	76	ND	ND	ND	ND	ND
BV1 #1	well		1 L	36	ND	ND	ND	ND	ND
BV1 #2	well	duplicate of BV1 #1	1 L	56	ND	ND	ND	ND	ND
BV2 #1	well		1 L	32	ND	ND	ND	ND	ND
BV2 #2	well	duplicate of BV2 #1	1 L	55	ND	ND	ND	ND	0.07*
BVCW #1	well		1 L	59	ND	ND	ND	ND	ND
BVCW #2	well	duplicate of BVCW #1	1 L	48	ND	ND	ND	ND	ND
BVCW #3	well	Spiked @ 2 ppb	1 L	52	2.74	0.36	0.14*	2.74	N/A
BVCW #4	well	duplicate of BVCW #1	1 L	48	ND	ND	ND	ND	ND

Table 5A Continued

Name	Type	Comments	Volume	(% Res	Measured Conc. ( $\mu\text{g/L}$ )					
					Air	Ala	Carb	Cyan	Met	
BVSM 1-4 combined	SM		900 mL	**	ND	ND	ND	ND	ND	ND
BV Amb To	Sea	Seawater at site of SM	1 L	25	ND	ND	ND	ND	ND	ND
WF1 #1	well		1 L	46	ND	ND	ND	ND	ND	ND
WF1 #2	well	Spiked @ 2 ppb	1 L	57	4.33	3.25	ND	6.98	4.84	
WF1 #3	well	duplicate of WF1 #1	1 L	47	ND	ND	ND	ND	ND	ND
WF1 #4	well	Spiked @ 4 ppb	1 L	35	9.34	5.94	ND	16.5	9.26	
WF2 #1	well		1 L	22	ND	ND	ND	ND	ND	ND
WF2 #2	well	duplicate of WF2 #1	1 L	1.2	ND	ND	ND	ND	ND	ND
WFSW #1	well		1 L	39	ND	ND	ND	ND	ND	0.07*
WFSW #2	well	duplicate of WFSW #1	1 L	21	ND	ND	ND	ND	ND	ND
WFSM 2&4 Combined	SM		1 L	35	ND	ND	ND	ND	ND	ND
WFSM 3	SM		1 L	19	ND	ND	ND	ND	ND	ND
CH2 #1	well		1 L	33	ND	ND	ND	ND	ND	ND
CH2 #2	well	duplicate of CH2 #1	1 L	47	ND	ND	ND	ND	ND	ND
CH3 #1	well		1 L	57	ND	ND	ND	ND	ND	0.33*
CH3 #2	well	duplicate of CH3 #1	1 L	9.7	ND	ND	ND	ND	ND	ND
CH3 #3	well	Spiked @ 2 ppb	1 L	75	ND	7.48	ND	6.75	N/A	
CH3 #4	well	Spiked @ 4 ppb	1 L	8.5	ND	12.7	ND	8.10	N/A	
CH5 #1	well		1 L	20	ND	ND	ND	ND	ND	ND

Table 5A Continued

Name	Type	Comments	Volume	(% Res	Measured Conc. ( $\mu\text{g/L}$ )					
					Atr	Ala	Carb	Cyan	Met	
CH5 #2	well	duplicate of CH5 #1	1 L	38	ND	ND	ND	ND	ND	ND
CHSM 1	SM		1 L	24	ND	ND	ND	ND	ND	ND
CHSM 3	SM		1 L	50	ND	ND	ND	ND	ND	ND
Hoopers Farm #1	well		1 L	50	ND	ND	ND	ND	ND	ND
Hoopers Farm #2	well	duplicate of HF #1	1 L	20	ND	ND	ND	ND	ND	ND
TB	blank	Travel blank	1 L	23	ND	ND	ND	ND	ND	ND
EB	blank	equipment blank	1 L	119	ND	ND	ND	ND	ND	ND
LFB #1	blank	Spiked @ 2 ppb	1 L	76	1.89	0.18*	ND	3.08	2.23	2.23
LB #2	blank	Reagent blank	1 L	12.1	ND	ND	ND	ND	ND	ND
LFB #3	blank	Spiked @ 4 ppb	1 L	5.5	8.55	6.31	ND	11.96	8.85	8.85

White  
c. 1000.e

Table 6A  
October Samples

Name	Type	Comments	Volume	(% Res)	Measured Conc. (µg/L)					
					Atr	Ala	Carb	Cyan	Met	
EV1 #1	well		1 L		ND	ND	ND	ND	ND	ND
EV1 #2	well	duplicate of EV1 #1	1 L	54	ND	ND	ND	ND	ND	ND
EV1 #3	well	Spiked @ 2 ppb	1 L	63	4.54	2.66	7.28	5.98	N/A	N/A
EV1 #4	well	Spiked @ 1 ppb	1 L	35	1.56	1.11	5.76	2.23	N/A	N/A
EV2 #1	well	Spiked @ 0.25 ppb	1 L	41	ND	0.36	N/A	1.62	N/A	N/A
EV2 #2	well		1 L		ND	ND	ND	ND	ND	ND
EV2 #3	well	duplicate of EV2 #2	1 L	84	ND	ND	ND	ND	ND	ND
EV2 #4	well	Spiked @ 1 ppb	1 L	119	2.61	1.62	9.33	3.54	N/A	N/A
EVF/R #1	well		1 L	89	ND	ND	ND	ND	ND	ND
EVF/R #2	well	duplicate of EVF/R #1	1 L	37	ND	ND	ND	ND	ND	ND
EVSM 1 #1	SM	Spiked @ 0.25 ppb	1 L	38	ND	0.62	N/A	1.82	N/A	N/A
EVSM 1 #2	SM	Spiked @ 1 ppb	1 L	75	5.76	1.93	3.93	5.44	N/A	N/A
EVSM 2 #1	SM		1 L	133	ND	ND	ND	ND	ND	ND
EVSM 2 #2	SM		1 L	142	ND	ND	ND	ND	ND	ND
EVSM 3 #1	SM		1 L	31	ND	ND	ND	ND	ND	ND
EVSM 4 #1	SM		1 L	32	ND	ND	ND	ND	ND	ND
EVSM 4 #2	SM	Spiked @ 0.5 ppb	1 L	40	ND	0.71	N/A	2.29	N/A	N/A
EV Amb To	Sea	Seawater at site of SM	1 L	45	ND	ND	ND	ND	ND	ND
BV1 #1	well		1 L	36	ND	ND	ND	ND	ND	ND
BV1 #2	well	duplicate of BV1 #1	1 L	4.1	ND	ND	ND	ND	ND	ND

Table 6A Continued

Name	Type	Comments	Volume	(% Resp	Measured Conc. (µg/L)				
					Atr	Ala	Carb	Cyan	Met
BV1 #3	well	Spiked @ 0.5 ppb	1 L	15	0.76	0.25*	N/A	1.43	N/A
BV1 #4	well	Spiked @ 2 ppb	1 L	65	3.83	2.57	6.72	1.28	N/A
BV2 #1	well		1 L	106	ND	ND	ND	ND	ND
BV2 #2	well	duplicate of BV2 #1	1 L	47	ND	ND	ND	ND	ND
BVCW #1	well		1 L	41	ND	ND	ND	ND	ND
BVCW #2	well	duplicate of BVCW #1	1 L	65	ND	ND	ND	ND	ND
BVCW #3	well	Spiked @ 1 ppb	1 L	26	2.42	0.80	3.90	1.55	N/A
BVCW #4	well	Spiked @ 0.25 ppb	1 L	9.1	ND	0.18*	N/A	0.16*	N/A
BVSM 2	SM		1 L	96	ND	ND	ND	ND	ND
BVSM 4 #1	SM	Spiked @ 2 ppb	1 L	**	2.77	0.91	3.87	1.69	N/A
BVSM 4 #2	SM		1 L	9.7	ND	ND	ND	ND	ND
BV Amb To	Sea	Seawater at site of SM	1 L	56	ND	ND	ND	ND	ND
WF1 #1	well		1 L	85	ND	ND	ND	ND	ND
WF1 #2	well	duplicate of WF1 #1	1 L	71	ND	ND	ND	ND	ND
WF1 #3	well	Spiked @ 0.5 ppb	1 L	79.6	ND	0.58	N/A	0.97	N/A
WF1 #4	well	Spiked @ 2 ppb	1 L	45	2.35	1.72	4.57	3.63	N/A
WF2 #1	well		1 L	56	ND	ND	ND	ND	ND
WF2 #2	well	duplicate of WF2 #1	1 L	43	ND	ND	ND	ND	ND
WF2 #3	well	Spiked @ 0.25 ppb	1 L	64	1.10	0.30*	N/A	2.4	ND
WF2 #4	well	Spiked @ 1 ppb	1 L	75	1.60	1.10	ND	1.90	N/A

Table 6A Continued

Name	Type	Comments	Volume	(% ) Resp	Measured Conc. (µg/L)					
					Air	Ala	Car	Cyn	Met	
WFSW #1	well		1 L	15	ND	ND	ND	ND	ND	ND
WFSW #2	well	duplicate of WFSW #1	1 L	69	ND	ND	ND	ND	ND	ND
WFSM 1&2 Combined	SM		1 L	38	ND	ND	ND	ND	ND	ND
WFSM 1	SM		1 L	11	ND	ND	ND	ND	ND	ND
WFSM 3	SM		1 L	64	ND	ND	ND	ND	ND	ND
WF Amb To	Sea	Seawater at site of SM	1 L		ND	ND	ND	ND	ND	ND
CH2 #1	well		1 L	11.5	ND	ND	ND	ND	ND	ND
CH3 #1	well	Spiked @ 0.25 ppb	1 L	81	ND	0.53	N/A	1.10	N/A	N/A
CH3 #2	well		1 L	21	ND	ND	ND	ND	ND	ND
CH3 #3	well	duplicate of CH3 #2	1 L	51	ND	ND	ND	ND	ND	ND
CH3 #4	well	Spiked @ 1 ppb	1 L	69	ND	1.39	ND	3.65	N/A	N/A
CH5	well		1 L	26	ND	ND	ND	ND	ND	ND
CH5 #1	well		1 L	75	ND	ND	ND	ND	ND	ND
CH5 #2	well	Spiked @ 2 ppb	1 L	52	4.15	2.32	5.49	5.50	N/A	N/A
CHSM 1-4 Combined	SM		1 L	4.1	ND	ND	ND	ND	0.04*	
CHSM 3	SM		1 L	2.5	ND	ND	ND	ND	ND	ND
CH Stream		Stream on Site	1 L	88	ND	ND	ND	ND	ND	ND
CH Amb To	Sea	Seawater at site of SM	1 L	**	ND	ND	ND	ND	ND	ND
ST1 #1	well		1 L	36	ND	ND	ND	ND	ND	ND



Table 6A Continued

Name	Type	Comments	Volume	(% Resp	Measured Conc. (µg/L)					
					Atr	Ala	Carb	Cyan	Met	
ST3 #1	well	Spiked @ 0.25 ppb	1 L	**	ND	0.27*	N/A	0.22*	N/A	
ST3 #2	well		1 L	126	ND	ND	ND	ND	ND	
ST3 #3	well	duplicate of ST3 #1	1 L	**	ND	ND	ND	ND	ND	
ST3 #4	well	duplicate of ST3 #2	1 L	**	ND	ND	ND	ND	ND	
EB #1	blank	equipment blank	1 L	**	ND	ND	ND	ND	ND	
EB #2	blank	duplicate of EB #1	1 L	33	ND	ND	ND	ND	ND	
EB #3	blank	duplicate of EB #1	1 L	77	ND	ND	ND	ND	ND	
TB	blank	Travel blank	1 L	33.5	ND	ND	ND	ND	ND	
LB #1	blank	Reagent blank	1 L	2.6	ND	ND	ND	ND	ND	
LB #2	blank	duplicate of LB #1	1 L	**	ND	ND	ND	ND	ND	
LB #3	blank	duplicate of LB #1	1 L	47	ND	ND	ND	ND	ND	
LFB #1	blank	Spiked @ 0.25 ppb	1 L	86	1.35	0.20*	N/A	0.98	N/A	
LFB #2	blank	Spiked @ 0.5 ppb	1 L	38	ND	0.78	N/A	1.40	N/A	
LFB #3	blank	Spiked @ 1 ppb	1 L	3.3	1.40	0.96	6.77	1.52	N/A	
LFB #4	blank	Spiked @ 2 ppb	1 L	6	4.01	0.16*	5.53	3.80	N/A	

Table 7A  
November Samples

Name	Type	Comments	Volume	(% ) Resp	Measured Conc. (µg/L)					
					Atr	Ala	Car	Cyn	Met	
EV1 #1	well		1 L	65	ND	ND	ND	ND	ND	ND
EV1 #2	well	duplicate of EV1 #1	1 L	80	ND	ND	ND	ND	ND	ND
EV2 #1	well		1 L	110	ND	ND	ND	0.03*	ND	ND
EV2 #2	well	duplicate of EV2 #2	1 L	88	ND	ND	ND	ND	ND	ND
EV2 #3	well	Spiked @ 1 ppb	1 L	9.3	ND	0.69	1.52	1.22	1.22	N/A
EV2 #4	well	Spiked @ 0.25 ppb	1 L	108	ND	0.23*	N/A	0.28	0.28	1.68
EVF/R #1	well		1 L	66	ND	ND	ND	ND	ND	ND
EVF/R #2	well	duplicate of EVF/R#1	1 L		ND	ND	ND	ND	ND	ND
EVF/R #3	well	Spiked @ 0.5 ppb	1 L	91	ND	0.45	N/A	0.90	0.90	3.86
EVF/R #4	well	Spiked @ 2 ppb	1 L	6.5	ND	1.27	0.89*	2.51	2.51	N/A
EVSM 1-4 Combined	SM		1 L	54	ND	ND	ND	ND	ND	ND
EVSM 4	SM		1 L	86	ND	ND	ND	ND	ND	ND
BV1 #1	well	Spiked @ 2 ppb	1 L	49	1.55	1.88	1.67	3.76	3.76	N/A
BV1 #2	well	Spiked @ 1 ppb	1 L	49	1.02	1.61	1.19*	3.32	3.32	N/A
BV1 #3	well	Spiked @ 0.5 ppb	1 L	74	ND	0.42	N/A	0.72	0.72	3.44
BV1 #4	well	Spiked @ 0.5 ppb	1 L	99	ND	0.31*	N/A	0.55	0.55	2.24
BV2 #1	well		1 L	62	ND	ND	ND	ND	ND	ND
BV2 #2	well	duplicate of BV2 #1	1 L	79	ND	ND	ND	ND	ND	ND
BVCW #1	well		1 L	51	ND	ND	ND	ND	ND	ND

Table 7A Continued

Name	Type	Comments	Volume	(% ) Resp	Measured Conc. (µg/L)				
					Atr	Ala	Car	Cyn	Met
BVCW #2	well	duplicate of BVCW #1	1 L	63	ND	ND	ND	ND	•
BVCW #3	well	Spiked @ 0.25 ppb	1 L	78	ND	0.65	N/A	1.93	2.91
BVCW #4	well	Spiked @ 1 ppb	1 L	37	0.75	0.92	1.01*	1.67	N/A
BVSM 1	SM		400 mL	36	ND	ND	ND	ND	ND
BVSM 3 #1	SM		1 L		ND	ND	ND	ND	ND
BVSM 3 #2	SM	Spiked @ 2 ppb	1 L	31	0.76	0.53	2.11	1.56	N/A
BVSM 3 #3	SM	duplicate of BVSM 3 #3	1 L	67	ND	ND	ND	ND	ND
BVSM 4 #1	SM		1 L	11.3	ND	ND	ND	ND	ND
BVSM 4 #2	SM	duplicate of BVSM 4 #1	1 L		ND	ND	ND	ND	ND
WF1 #1	well		1 L	52	ND	ND	ND	ND	ND
WF1 #2	well	duplicate	1 L	93	ND	ND	ND	ND	ND
WF1 #3	well	Spiked @ 0.25 ppb	1 L	100	ND	0.22*	N/A	0.41	2.11
WF1 #4	well	Spiked @ 1 ppb	1 L	67	ND	0.78	7.88	1.22	N/A
WF2 #1	well		1 L	47	ND	ND	ND	ND	ND
WF2 #2	well	duplicate of WF2 #1	1 L	26	ND	ND	ND	ND	ND
WF2 #3	well	Spiked @ 2 ppb	1 L	45	1.19	1.42	1.05*	2.71	N/A
WF2 #4	well	Spiked @ 0.5 ppb	1 L	45	ND	0.25*	N/A	0.43	2.10
WFSW #1	well		1 L	92	ND	ND	ND	ND	ND
WFSW #2	well	duplicate of WFSW #1	1 L	78	ND	ND	ND	ND	ND
WF Amb To	Sea	Seawater at site of SM	1 L	74	ND	ND	ND	ND	ND

\* Note that Metolachlor's spectrum was detected on the GC/MS below the LDO

Table 7A Continued

Name	Type	Comments	Volume	(% ) Resp	Measured Conc. (µg/L)					
					Atr	Ala	Car	Cyn	Met	
CH2	well		1 L		ND	ND	ND	0.02*	ND	
CH2 #1	well	duplicate of CH2	1 L	73	ND	ND	ND	ND	ND	
CH3 #1	well		1 L	86	ND	ND	ND	ND	ND	
CH3 #2	well	duplicate of CH3 #1	1 L	102	ND	ND	ND	ND	ND	
CH3 #3	well	Spiked @ 1 ppb	1 L	**	ND	0.76	1.17*	1.20	N/A	
CH3 #4	well	Spiked @ 0.25 ppb	1 L		ND	0.20*	N/A	0.46	1.82	
CH5 #1	well		1 L	85	ND	ND	ND	0.55	ND	
CH5 #2	well	duplicate of CH5 #2	1 L	88	ND	ND	ND	ND	ND	
CH Amb To	Sea		1 L	38	ND	ND	ND	ND	ND	
STI 4	well		1 L	103	ND	ND	ND	ND	ND	
ST Middle	well		1 L	52	ND	ND	ND	ND	ND	
ST Middle #1	well	duplicate of ST Middle	1 L	115	ND	ND	ND	ND	ND	
EB #1	blank	equipment blank	1 L	93	ND	ND	ND	ND	ND	
EB #2	blank	duplicate of EB #1	1 L	95	ND	ND	ND	ND	ND	
TB #1	blank	Travel blank	1 L	43	ND	ND	ND	ND	ND	
TB #2	blank	duplicate of TB #1	1 L	61	ND	ND	ND	ND	ND	
LB #1	blank	Reagent blank	1 L		ND	ND	ND	ND	ND	
LB #2	blank	duplicate of LB #1	1 L	67	ND	ND	ND	ND	ND	
LB #3	blank	duplicate of LB #1	1 L		ND	ND	ND	ND	ND	
LFB #1	blank	Spiked @ 0.25 ppb	1 L	83	ND	0.23*	N/A	0.47	0.02	

Table 7A Continued

Name	Type	Comments	Volume	(% ) Resp	Measured Conc. (µg/L)				
					Air	Ala	Car	Cyn	Met
LFB #2	blank	Spiked @ 0.5 ppb	1 L	16	ND	0.28*	0.72*	0.52	N/A
LFB #3	blank	Spiked @ 2 ppb	1 L	59	0.98	1.62	N/A	3.14	9.76

Table 8A  
December Samples

Name	Type	Comments	Volume	(% ) Resp	Measured Conc. (µg/L)					
					Atr	Ala	Car	Cyn	Met	
EV1 #1	well	Spiked @ 0.5 ppb	1 L	89	ND	0.29*	N/A	0.70	ND	ND
EV1 #2	well	Spiked @ 0.25 ppb	1 L	95	ND	0.25*	N/A	0.48	1.42	ND
EV1 #3	well		1 L	32	ND	ND	ND	ND	ND	ND
EV1 #4	well	duplicate of EV1 #3	1 L	31	ND	ND	ND	ND	ND	ND
EV2 #1	well		1 L	67	ND	ND	ND	ND	ND	ND
EV2 #2	well	duplicate of EV2 #1	1 L	124	ND	ND	ND	ND	ND	ND
EVF/R #1	well		1 L	76	ND	ND	ND	ND	ND	ND
EVF/R #2	well	Spiked @ 2 ppb	1 L	52	0.84	1.19	2.00	2.82	N/A	N/A
EVF/R #3	well	Spiked @ 1 ppb	1 L	24	0.90	1.17	2.31	2.56	N/A	N/A
EVF/R #4	well	duplicate of EVF/R #3	1 L	18	ND	ND	ND	ND	ND	ND
EVSM 2	SM		500 mL	27	ND	ND	ND	ND	ND	ND
EV Amb To	Sea	Seawater at site of SM	1 L	49	ND	ND	ND	ND	ND	ND
BV1 #1	well	Spiked @ 2 ppb	1 L	37	1.41	1.43	2.10	2.98	N/A	N/A
BV1 #2	well	Spiked @ 1 ppb	1 L	107	0.58	0.86	0.85*	2.04	N/A	N/A
BV1 #3	well		1 L	81	ND	ND	ND	ND	ND	ND
BV1 #4	well	duplicate of BV1 #3	1 L	89	ND	ND	ND	ND	ND	ND
BV2 #1	well		1 L	90	ND	ND	ND	ND	ND	ND
BV2 #2	well	duplicate of BV2 #1	1 L	67	ND	ND	ND	ND	ND	ND
BVCW #1	well	Spiked @ 0.5 ppb	1 L	106	ND	0.29*	N/A	0.63	2.01	ND
BVCW #2	well		1 L	**	ND	ND	ND	ND	ND	ND

Table 8A Continued

Name	Type	Comments	Volume	(% ) Resp	Measured Conc. (µg/L)					
					Air	Ala	Car	Cyn	Met	
BVCW #3	well	Spiked @ 0.25 ppb	1 L	38	ND	0.21*	N/A	N/A	0.60	0.12*
BVCW #4	well	duplicate of BVCW #2	1 L	**	ND	ND	ND	ND	ND	ND
WF1 #1	well		1 L	113	ND	ND	ND	ND	ND	ND
WF1 #2	well	duplicate of WF1 #1	1 L	43	ND	ND	ND	ND	ND	ND
WF1 #3	well	Spiked @ 0.25 ppb	1 L	49	ND	0.19*	N/A	N/A	0.57	1.09
WF1 #4	well	Spiked @ 0.5 ppb	1 L	44	ND	0.35	N/A	N/A	0.57	2.27
WF2 #1	well	Spiked @ 2 ppb	1 L	53	0.25*	0.34	0.61*	0.61*	1.10	N/A
WF2 #2	well		1 L	21	ND	ND	ND	ND	ND	ND
WF2 #3	well	Spiked @ 1 ppb	1 L	24	1.37	1.22	1.75	1.75	2.47	N/A
WF2 #4	well	duplicate of WF2 #3	1 L	44	ND	ND	ND	ND	ND	ND
WFSW #1	well		1 L	135	ND	ND	ND	ND	0.02*	ND
WFSW #2	well	duplicate of WFSW #2	1 L	106	ND	ND	ND	ND	ND	ND
WFSM 3	SM		1 L	**	ND	ND	ND	ND	ND	ND
WF Amb To	Sea	Seawater at site of SM	1 L	36	ND	ND	ND	ND	ND	ND
TB	blank	Travel blank	1 L	56	ND	ND	ND	ND	ND	ND
EB #1	blank	equipment blank	1 L	16	ND	ND	ND	ND	ND	ND
EB #2	blank	duplicate of EB #1	1 L	62	ND	ND	ND	ND	ND	ND
LB #1	blank	Reagent blank	1 L	23	ND	ND	ND	ND	ND	ND
LB #2	blank	duplicate of LB #1	1 L	82	ND	ND	ND	ND	ND	ND

Table 8A Continued

Name	Type	Comments	Volume	(% ) Resp	Measured Conc. (µg/L)				
					Atr	Ala	Car	Cyn	Met
LFB #1	blank	Spiked @ 0.25 ppb	1 L	44	ND	0.20*	1.60	0.33	0.05*
LFB #2	blank	Spiked @ 1 ppb	1 L	61	0.53	0.63	N/A	1.44	4.44



Table 9A  
January Samples

Name	Type	Comments	Volume	(% Resp	Measured Conc. (µg/L)					
					Atr	Ala	Car	Cyn	Met	
EV1 #1	well		1 L	63	ND	ND	ND	ND	ND	ND
EV1 #2	well	duplicate of EV1 #1	1 L	**	ND	ND	ND	ND	ND	ND
EV2 #1	well	Spiked @ 0.25 ppb	1 L	3.6	0.34	0.21*	N/A	0.35	0.09*	0.09*
EV2 #2	well		1 L	4.1	ND	ND	ND	ND	ND	ND
EV2 #3	well	Spiked @ 0.5 ppb	1 L	**	0.22*	0.41	N/A	0.67	0.27*	0.27*
EV2 #4	well	duplicate of EV2 #3	1 L	3.4	ND	ND	ND	ND	ND	ND
EVF/R #1	well		1 L	102	ND	ND	ND	ND	ND	ND
EVF/R #2	well	duplicate of EVF/R #1	1 L	**	ND	ND	ND	ND	ND	ND
EVF/R #4	well	Spiked @ 2 ppb	1 L	70	0.42	1.05	1.88	1.48	N/A	N/A
BV1 #1	well	Spiked @ 1 ppb	1 L	34	0.42	0.46	2.53	0.64	N/A	N/A
BV1 #2	well	Spiked @ 2 ppb	1 L	**	0.75	1.20	6.33	1.81	N/A	N/A
BV1 #3	well		1 L	**	ND	ND	ND	ND	ND	ND
BV1 #4	well	Spiked @ 0.25 ppb	1 L	**	0.37	0.19*	N/A	0.39	0.18*	0.18*
BV2 #1	well		1 L	115	ND	ND	ND	ND	ND	ND
BV2 #2	well	duplicate of BV2 #1	1 L	122	ND	ND	ND	ND	ND	ND
BVCW #1	well		1 L	104	ND	ND	ND	ND	ND	ND
BVCW #2	well	Spiked @ 0.25 ppb	1 L	36	0.35*	0.23*	N/A	0.34	0.10*	0.10*
BVCW #3	well	duplicate of BVCW #1	1 L	**	ND	ND	ND	ND	ND	ND
BVCW #4	well	Spiked @ 0.5 ppb	1 L	**	0.44	0.41	N/A	0.73	0.25*	0.25*
WF1 #1	well		1 L	144	ND	ND	ND	ND	ND	ND

Table 9A Continued

Name	Type	Comments	Volume	(% Resp	Measured Conc. ( $\mu\text{g/L}$ )					
					Air	Ala	Car	Cyn	Met	
WF1 #2	well	duplicate of WF1 #1	1 L	147	ND	ND	ND	ND	ND	ND
WF1 #3	well	Spiked @ 1 ppb	1 L	125	0.71	0.76	2.33	1.04	N/A	N/A
WF1 #4	well	Spiked @ 2 ppb	1 L	**	2.30	1.58	4.72	2.46	N/A	N/A
WF2 #1	well		1 L	116	ND	ND	ND	ND	ND	ND
WF2 #2	well	duplicate of WF2 #1	1 L	122	ND	ND	ND	ND	ND	ND
WF2 #3	well	Spiked @ 0.25 ppb	1 L	**	0.14*	0.19*	N/A	0.32*	0.09*	0.09*
WF2 #4	well	Spiked @ 0.5 ppb	1 L	137	ND	0.41	N/A	0.64	ND	ND
WFSW #1	well		1 L	11	ND	ND	ND	ND	ND	ND
WFSW #2	well	duplicate of WFSW #1	1 L	150	ND	ND	ND	ND	ND	ND
WFSM 1	well		800mL	29	ND	ND	ND	ND	ND	ND
WFSM 2	SM		1 L	30	ND	ND	ND	ND	ND	ND
WFSM 2&3 Combined	SM		800 mL	71	ND	ND	ND	ND	ND	ND
WFSM 3	SM		1 L	48	ND	ND	ND	ND	ND	ND
WFSM 4 #1	SM		1 L	**	ND	ND	ND	ND	ND	ND
WFSM 4 #2	SM	duplicate of WFSM 4 #1	1 L	77	ND	ND	ND	ND	ND	ND
WFSM 4 #3	SM	Spiked @ 2 ppb	1 L	75	0.80	1.31	0.63*	2.10	N/A	N/A
WF Amb To	Sea	Seawater at site of SM	1 L	144	ND	ND	ND	ND	ND	ND
TB #1	blank	Travel blank	1 L	103	ND	ND	ND	ND	ND	ND
TB #2	blank	duplicate of TB #1	1 L	**	ND	ND	ND	ND	ND	ND
EB #1	blank	equipment blank	1 L	60	ND	ND	ND	ND	ND	ND

Table 9A Continued

Name	Type	Comments	Volume	(% Resp	Measured Conc. (µg/L)					
					Air	Ala	Car	Cyn	Met	
EB #2	blank	duplicate of EB #1	1 L	**	ND	ND	ND	ND	ND	ND
LB #1	blank	Reagent blank	1 L	**	ND	ND	ND	ND	ND	ND
LB #2	blank	duplicate of LB #1	1 L	**	ND	ND	ND	ND	ND	ND
LFB #2	blank	Spiked @ 0.5 ppb	1 L	**	0.39	0.42	N/A	0.66	0.28*	0.28*

Table 10A  
February Samples

Name	Type	Comments	Volume	(% Resp	Measured Conc. ( $\mu\text{g/L}$ )					
					Air	Ala	Car	Cyn	Met	
EV1 #1	well		1 L	**	ND	ND	ND	ND	ND	ND
EV1 #2	well	duplicate of EV1 #1	1 L	**	ND	ND	ND	ND	ND	ND
EV1 #3	well	Spiked @ 0.5 ppb	1 L	116	0.45	0.23*	N/A	0.47	0.11*	0.09*
EV1 #4	well	Spiked @ 0.25 ppb	1 L	45	ND	0.07*	N/A	0.16*	0.09*	0.09*
EV2 #1	well		1 L	**	ND	ND	ND	ND	ND	ND
EV2 #2	well	duplicate of EV2 #1	1 L	52	ND	ND	ND	ND	ND	ND
EV2 #3	well	Spiked @ 1 ppb	1 L	37	0.88	0.83	2.52	1.04	N/A	N/A
EV2 #4	well	Spiked @ 2 ppb	1 L	17	0.52	1.22	3.17	ND	N/A	N/A
EVF/R #1	well		1 L	**	ND	ND	ND	ND	ND	ND
EVF/R #2	well	duplicate of EVF/R #1	1 L	130	ND	ND	ND	ND	ND	ND
EV Amb To	Sea	Seawater at site of SM	1 L	48	ND	ND	ND	ND	ND	ND
BV1 #1	well		1 L	142	ND	ND	ND	ND	ND	ND
BV1 #2	well	duplicate of BV1 #1	1 L	59	ND	ND	ND	ND	ND	ND
BV1 #3	well	Spiked @ 0.25 ppb	1 L	48	ND	0.19*	N/A	0.28*	0.06*	0.06*
BV1 #4	well	Spiked @ 0.5 ppb	1 L	30	0.34*	0.40	N/A	0.58	0.28*	0.28*
BV2 #1	well		1 L	96	ND	0.05*	ND	ND	0.04*	0.04*
BV2 #2	well	duplicate of BV2 #1	1 L	82	ND	ND	ND	ND	ND	ND
BVCW #1	well		1 L	**	ND	ND	ND	ND	ND	ND
BVCW #3	well	Spiked @ 0.5 ppb	1 L	22	ND	0.27*	N/A	0.38	0.12*	0.12*
BVCW #4	well	Spiked @ 0.5 ppb	1 L	73	ND	0.38	N/A	0.50	0.22*	0.22*

Table 10A Continued

Name	Type	Comments	Volume	(% Resp	Measured Conc. (µg/L)					
					Atr	Ala	Car	Cyn	Met	
BVSM 1 #1	SM		1 L	141	ND	ND	ND	ND	ND	
BVSM 1 #2	SM	duplicate of BVSM 1 #1	1 L	**	ND	ND	ND	ND	ND	
BVSM 1 #3	SM	Spiked @ 1 ppb	1 L	28	2.38	0.61	ND	ND	N/A	
BVSM 1 #4	SM	Spiked @ 2 ppb	1 L	17	0.50	.91	4.71	1.16	N/A	
BVSM 1&3 Combined	SM		1 L	94	ND	ND	ND	ND	ND	
BVSM 3 #1	SM	Spiked @ 0.5 ppb	1 L	**	0.58	0.66	N/A	1.19	0.68	
BVSM 3 #2	SM		1 L	**	ND	ND	ND	ND	ND	
BVSM 3 #3	SM	duplicate of BVSM 3 #1	1 L	40	ND	ND	ND	ND	ND	
BVSM 3&4 Combined	SM		600 mL	50	ND	ND	ND	ND	ND	
BVSM 4 #1	SM		1 L		ND	ND	ND	ND	ND	
BVSM 4 #2	SM	Spiked @ 0.5 ppb	1 L	24	ND	ND	ND	ND	ND	
BVSM 4 #3	SM	Spiked @ 1 ppb	1 L	118	0.83	1.39	13.0	0.26*	N/A	
BVSM 4 #4	SM	duplicate of BVSM 4 #1	1 L		ND	ND	ND	ND	ND	
BV Amb To	Sea	Seawater at site of SM	1 L	31	ND	ND	ND	ND	ND	
WF1 #1	well		1 L	53	ND	ND	ND	ND	ND	
WF1 #2	well	duplicate of WF1 #1	1 L	122	ND	ND	ND	ND	ND	
WF1 #3	well	Spiked @ 0.5 ppb	1 L	24	0.33*	0.06*	N/A	0.48	0.18*	
WF1 #4	well	Spiked @ 0.25 ppb	1 L	51	ND	0.08*	N/A	0.23*	ND	
WF2 #1	well		1 L		ND	ND	ND	ND	ND	

Table 10A Continued

Name	Type	Comments	Volume	(% Resp	Measured Conc. (µg/L)					
					Atr	Ala	Car	Cyn	Met	
WF2 #2	well	duplicate of WF2 #1	1 L	115	ND	ND	ND	ND	ND	ND
WF2 #3	well	Spiked @ 1 ppb	1 L	**	1.09	0.89	19.3	1.31	N/A	N/A
WFSW #1	well		1 L	41	ND	ND	ND	ND	ND	ND
WFSW #2	well	duplicate of WFSW #1	1 L	20	ND	ND	ND	ND	ND	ND
WFSM 1	SM		750 mL	**	ND	ND	ND	ND	ND	ND
WFSM 2	SM		1 L	58	ND	ND	ND	ND	ND	ND
WFSM 3	SM		1 L	46	ND	ND	ND	ND	ND	ND
WFSM 3&4 Combined	SM		1 L	**	ND	ND	ND	ND	ND	ND
WFSM 4	SM		1 L	141	ND	ND	ND	ND	ND	ND
WF Amb To	Sea	Seawater at site of SM	1 L	68	ND	ND	ND	ND	ND	ND
CH2 #1	well		1 L	65	ND	ND	ND	ND	ND	ND
CH3 #1	well	Spiked @ 0.5 ppb	1 L	57	0.32*	0.22*	N/A	0.31*	0.08*	0.08*
CH3 #2	well		1 L	**	ND	ND	ND	ND	ND	ND
CH3 #3	well	duplicate of CH3 #2	1 L	102	ND	ND	ND	ND	ND	ND
CH3 #4	well	Spiked @ 0.25 ppb	1 L	107	ND	0.22*	N/A	0.34	0.06*	0.06*
CH5 #1	well		1 L	54	ND	ND	ND	ND	ND	ND
CH5 #2	well	duplicate of CH5 #1	1 L	83	ND	ND	ND	ND	ND	ND
CH5 #3	well	Spiked @ 0.25 ppb	1 L	72	ND	0.06*	N/A	0.19*	ND	ND
CHSM 2	SM		1 L	18	ND	ND	ND	ND	ND	0.32*

Table 10A Continued

Name	Type	Comments	Volume	(% ) Resp	Measured Conc. (µg/L)					
					Air	Ala	Car	Cyn	Met	
CHSM 3	SM		1 L	21	ND	ND	ND	ND	ND	ND
CHSM 4	SM		1 L	52	ND	ND	ND	ND	ND	ND
CH Amb To	Sea	Seawater at site of SM	1 L	60	ND	ND	ND	ND	ND	ND
EB #1	blank	equipment blank	1 L	130	ND	ND	ND	ND	ND	ND
EB #2	blank	duplicate of EB #1	1 L	33	ND	ND	ND	ND	ND	ND
TB	blank	Travel blank	1 L	109	ND	ND	ND	ND	ND	ND
LB #1	blank	laboratory blank	1 L	29	ND	ND	ND	ND	ND	ND
LB #2	blank	duplicate of LB #1	1 L	**	ND	ND	ND	ND	ND	ND
LB #3	blank	duplicate of LB #1	1 L	**	ND	ND	ND	ND	ND	ND
LB #4	blank	duplicate of LB #1	1 L	31	ND	ND	ND	ND	ND	ND
LFB #1	blank	Spiked @ 0.25 ppb	1 L	129	0.55	0.30*	N/A	0.44	0.22*	0.43*
LFB #2	blank	Spiked @ 0.5 ppb	1 L	105	0.56	0.54	N/A	0.73	0.43*	0.43*
LFB #3	blank	Spiked @ 1 ppb	1 L	127	0.80	0.81	7.53	1.12	N/A	N/A
LFB #4	blank	Spiked @ 2 ppb	1 L	49	0.92	1.07	7.43	1.29	N/A	N/A

## Appendix B

### Chronology

The date of all the important procedures performed in the study are listed in table B1.



Table B1  
Chronology of Data Collection

April 1992	
3/28-3/29	Collected samples
3/30-4/2	Extracted samples
4/9	Injected samples into the GC/FID
4/15-4/16	Injected samples into the GC/MS
4/23	Injected samples into the GC/MS
May 1992	
5/5-5/8	Collected samples
5/9-5/11	Extracted samples
5/18-5/19	Injected samples into the GC/MS
5/30	Injected calibration standards into the GC/MS
June 1992	
6/18-6/20	Collected samples
6/21-6/22	Extracted samples
6/25-6/27	Injected samples into the GC/FID
6/30	Injected samples into the GC/MS
7/2-7/4	Injected samples into the GC/ECD
7/1	Extracted check standards
7/1	Injected calibration standards into the GC/ECD
7/8	Injected check standards into the GC/MS
8/2	Injected samples with possible pesticides into the GC/MS
July 1992	
7/18-7/20	Collected samples
7/21-7/22	Extracted samples
7/26-7/28	Injected samples into the GC/ECD
8/10	Injected samples into the GC/ECD

Table B1 continued

8/11	Injected samples into the GC/FID
8/19	Injected Calibration standard into the GC/MS
August 1992	
8/24-8/26	Collected samples
8/26-8/27	Extracted samples
8/30	Extracted samples
9/2-9/4	Injected samples into the GC/ECD
9/5	Injected samples into the GC/MS
9/9	Injected samples into the GC/MS
9/9	Injected samples into the GC/ECD with DB-5 column
9/10-9/11	Injected samples into the GC/ECD with DB-210 column
9/17-9/18	Injected samples into the GC/ECD with DB-210 column
9/22	Injected samples into the GC/MS
10/6-10/7	Injected calibration standards into the GC/ECD with DB-210 column
October 1992	
10/9-10/11	Collected samples
10/12-10/14	Extracted samples
10/17	Injected Calibration standards into the GC/MS
10/25-10/27	Injected samples into the GC/ECD with DB-5 column
10/28-10/31	Injected samples into the GC/ECD with DB-210 column
11/5-11/6	Injected calibration standards into the GC/ECD with DB-210 column
11/9-11/10	Injected calibration standards into the GC/ECD with DB-210 column

Table B1 continued

November 1992	
11/13-11/15	Collected samples
11/16-11/18	Extracted samples
12/1-12/2	Injected samples into the GC/ECD with DB-210 column
December 1992	
12/4-12/6	Collected samples
12/7-12/8	Extracted samples
12/10-12/12	Injected samples into the GC/ECD with DB-210 column
12/31, 1/3, 1/6	Injected samples into the GC/ECD with the DB-5 column
1/4	Autosampler installed on GC/ECD
January 1993	
1/9-1/11	Collected samples
1/12-1/13	Extracted samples
1/19-1/20	Injected samples into the GC/ECD with DB-5 column
1/21-1/22	Injected calibration standards into the GC/ECD with DB-5 column
1/23-1/24	Injected calibration standards into the GC/ECD with DB-210 column
1/24-1/25	Injected samples into the GC/ECD with DB-210 column
1/26-1/27	Re-injected samples with pesticides detected into the GC/ECD with DB-210 column
1/29-1/30	Re-injected samples with pesticides detected into the GC/ECD with DB-5 column
February 1993	
2/11-2/14	Collected samples
2/15-2/16	Extracted samples

Table 1B Continued

2/17-2/18	Injected samples into the GC/ECD with DB-5 column
2/19-2/21	Injected samples into the GC/ECD with DB-210 column

## Appendix C

### Statistical Analysis

For the project, only standard statistical tests were performed, including mean, standard deviation, and linear regression. For the instrument limit of detection (LOD) and method detection limit (MDL), the standard deviation of the samples in question was multiplied by the value of the t-test at a 99% confidence level. The LOD and MDL's were given in the methods and materials section under tables 3 and 4. The following tables contain the least squares regression equations for the pesticides, separated by instrument, column, and time.

**Table 1C  
Regression Output for the GC/MS in June, 1992**

Pesticide	R <sup>2</sup>	Regression Output, where x = ng injected on column
Atrazine	0.97	806853x - 131959
Alachlor	0.97	458920x - 281977
Carbofuran	0.84	586335x - 1357451

\* Note y predicts pesticide concentration in  $\mu\text{g/L}$

Table 2C  
Regression Output for the GC/MS in Oct., 1992

Pesticide	R <sup>2</sup>	Regression Output, where x = ng injected on column
Alachlor	0.96	414939x - 741467
Atrazine	0.91	1146072x - 4260664
Carbofuran	0.85	664297x - 3934340
Metolachlor	0.97	1062405 - 1720167

\* Note y predicts pesticide concentration in  $\mu\text{g/L}$

Table 3C  
Regression Output for GC/FID in June 1992

Pesticide	R <sup>2</sup>	Regression Output, where x = ng injected on column
Alachlor	0.55	855x - 19.8
Atrazine	0.63	586x - 980
Carbofuran	0.53	1285x - 3982
Cyanazine	0.42	298x - 131

\* Note y predicts pesticide concentration in  $\mu\text{g/L}$



**Table 4C**  
**Regression Output for GC/ECD with DB-5 Column in Aug. 1992**

Pesticide	R <sup>2</sup>	Regression Output, where x = ng injected on the column
Alachlor	0.70	72943x + 144953
Atrazine	0.97	4694x + 7615
Carbofuran	0.46	3520x - 9747
Cyanazine	0.80	58382x -18546

\* Note y predicts pesticide concentration in  $\mu\text{g/L}$

Table 5C  
Regression Output for GC/ECD with DB-5 column in Jan. 1993

Pesticide	R <sup>2</sup>	Regression Output, where x = ng inject on column
Alachlor	0.992	465934x + 87653
Atrazine	0.994	59391x - 41172
Carbofuran	0.97	4859x - 9983
Cyanazine/Metolachlor*	0.88	934673x - 624817

\* Note that cyanazine and metolachlor were not resolved on the DB-5 column.

\* Note y predicts pesticide concentration in  $\mu\text{g/L}$

Table 6C  
Regression Output for the GC/ECD with DB-210 column in Oct. 1992

Pesticide	R <sup>2</sup>	Regression Output, where x = ng injected on the column
Alachlor	0.97	205166x + 11270
Atrazine	0.85	6064x - 12837
Cyanazine	0.86	71000x + 43649
Metolachlor	0.88	105340x + 19405

\* Note y predicts pesticide concentration in  $\mu\text{g/L}$

**Table 7C**  
**Regression Output for the GC/ECD with DB-210 column in Nov. 1992**

<b>Pesticide</b>	<b>R<sup>2</sup></b>	<b>Regression Output, where x = ng injected on column</b>
<b>Alachlor</b>	<b>0.992</b>	<b>306973x + 19426</b>
<b>Atrazine</b>	<b>0.81</b>	<b>11868x + 20666</b>
<b>Carbofuran</b>	<b>0.78</b>	<b>11099x + 13368</b>
<b>Cyanazine</b>	<b>0.98</b>	<b>118721x + 9292</b>
<b>Metolachlor</b>	<b>0.97</b>	<b>145449x + 43303</b>

\* Note y predicts pesticide concentration in  $\mu\text{g/L}$

Table 8C  
Regression Output for GC/ECD with DB-210 column in Jan. 1992

Pesticide	R <sup>2</sup>	Regression Output, where x = ng injected on the column
Alachlor	0.99	734129x + 48699
Atrazine	0.70	33827x + 11668
Cyanazine	0.94	50939x - 96479
Metolachlor	0.997	366665x + 72882

\* Note y predicts pesticide concentration in  $\mu\text{g/L}$

Table 9C  
Response factor Output for the GC/MS in Feb., 1992

Pesticide	Response Factor for Phenanthrene-d <sub>10</sub>
Alachlor	0.0946
Atrazine	0.1236
Carbofuran	0.0949
Cyanazine	0.0391
Metolachlor	0.2564

## Appendix D

### Nutrient Data

The seepage meter discharge, nutrient flux in the seepage, and nutrient concentration data were calculated by Dr. William Reay. The upland well data was shown in Table 1D, and the submarine groundwater discharge data was included in Table 2D.

Table 1D  
 Discharge, Sediment Nutrient Flux, and Nitrate Concentration  
 in the Seepage Water at the Study Sites

Agricultural site 1				
Month	Discharge (L/m <sup>2</sup> *hr)	Conc. <sup>a</sup> NO <sub>3</sub> <sup>-</sup> (mg/L as N)	Flux NO <sub>3</sub> <sup>-</sup> NH <sub>4</sub> <sup>+</sup> (mg/m <sup>2</sup> *hr)	
May	1.21	1.04	4.20	0.24
June	0.62			
July	0.12	0.03	- 0.04	0.06
Aug	0.43	0.14	0.11	0.32
Oct	1.27	2.83	14.82	0.09
Nov	0.95	4.05	20.15	- 0.02
Feb	2.09	0.91	5.49	0.21
Agricultural site 2				
May	0.87	0.63	1.91	0.05
June	0.69	0.25	0.75	0.02
July	1.18	0.06	0.02	- 0.02
Oct	0.56	0.35	0.79	0.03
Nov	0.23	0.47	0.13	0.02
Feb	0.76	1.80	5.39	0.03
Agricultural site 3				
May	0.46	0.01	- 0.19	0.18
June	1.04	0.01	- 0.22	0.05
July	0.36	0.02	0.02	- 0.01
Aug	0.81	0.01	0.04	1.29
Oct	1.47	3.09	11.66	0.13
Nov	0.30	0.06	- 1.19	0.11
Dec	0.21	0.06	- 0.52	
Jan				
Feb	0.10	0.02	- 0.25	0.02



Table 1D Continued

Agricultural site 4				
Month	Discharge (L/m <sup>2</sup> *hr)	Conc. <sup>a</sup> NO <sub>3</sub> <sup>-</sup> (mg/L as N)	Flux	
			NO <sub>3</sub> <sup>-</sup> (mg/m <sup>2</sup> *hr)	NH <sub>4</sub> <sup>+</sup> (mg/m <sup>2</sup> *hr)
May	0.40	0.05	0.06	0.09
June	0.38	0.01		
July	0.19	0.03	- 0.03	0.30
Oct	0.48	0.33	1.03	0.29
Nov	0.09	0.24	0.41	0.13
Dec	0.30	0.06	0.19	
Jan	0.68	1.66	4.82	0.13
Feb	0.75	0.49	1.33	0.06

a Values are for nitrate concentrations measured in seepage water, which is a combination of groundwater and bay water. Ambient nitrate concentrations in the bay water ranged from <0.01 to 1.47 mg/L as N.

**Table 2D**  
**Upland Groundwater Nutrient Quality at Study Sites**

<b>Agricultural site 2</b>			
<b>Month</b>	<b>NO<sub>3</sub><sup>-</sup>, (mg/L as N)</b>	<b>PO<sub>4</sub><sup>-3</sup>, (mg/L)</b>	<b>NH<sub>4</sub><sup>+</sup>, (mg/L as N)</b>
May	13.09	0.005	0.04
June	14.83	0.006	0.03
July	13.76	0.017	0.05
Oct	13.78	0.003	0.07
Nov	14.22	0.011	0.04
Dec	13.96		
Feb	16.60	<0.001	0.05
<b>Agricultural site 3</b>			
May	13.73	0.015	0.03
June	11.17	0.027	0.05
July	7.86	0.038	0.04
Aug	20.33	0.039	0.06
Oct	19.57	0.042	0.038
Nov	15.68	0.022	0.02
Dec	17.32	0.023	
Jan	12.93	0.017	0.029
Feb	18.6	0.014	0.01
<b>Agricultural site 1</b>			
May	14.21		0.031
June	18.42		0.038
July	12.86	0.034	0.013
Aug	13.27	0.012	0.034
Oct	11.71	0.008	0.051
Nov	17.23	0.020	0.016
Jan	15.14	0.002	0.043

Table 2D Continued

Agricultural site 1, continued			
Month	NO <sub>3</sub> <sup>-</sup> , (mg/L as N)	PO <sub>4</sub> <sup>-3</sup> , (mg/L)	NH <sub>4</sub> <sup>+</sup> , (mg/L as N)
Feb	17.48	0.03	0.03
Agricultural site 4			
May	9.98	0.005	0.030
June	11.71	0.004	0.05
July	10.92	0.035	0.04
Oct	8.82	0.005	0.14
Nov	13.98	0.017	0.02
Jan	20.60	0.011	0.018
Feb	10.56	0.005	0.01

## Appendix E

### Soil Sample Data

Starting in July 1992, Douglas Schicho, a Master's degree candidate at VPI & SU, took soil samples at all the sites as the water samples were being collected. Samples were taken from the area of the monitored wells and from the near shore sediment at each site. The soil samples were soxhlex extracted and injected on the GC/MS. Table 1E contained the LOD for the pesticides, and the data from the soil samples was included in Table 2E.

#### Abbreviations

BV: Agricultural site 1  
CH: Agricultural site 2  
EV: Agricultural site 3  
WF: Agricultural site 4  
F: Field sample  
OS: Off-shore (sediment) sample

Table 1E  
Instrument LOD for the Soil Samples, after Schicho

Pesticide	LOD, in mg/kg (ppm)
Atrazine	1.1
Alachlor	2.5
Cyanazine	1.9
Metolachlor	0.3

\* Note that LOD did not include the extraction concentration factor

Table 2E  
Soil Sample Data, after Schicho (all numbers are in mg/kg)

Extraction date	Sample	Sample Type	Carb	Atr	Ala	Met	Cyan
7/15/92	CHF 6/19	FIELD SAMPLE	nd	nd	nd	0.1105	nd
7/14/92	EVF 6/18	FIELD SAMPLE	nd	nd	nd	0.0269	nd
7/14/92	BVF 6/18	FIELD SAMPLE	nd	nd	0.1452	0.0295	nd
7/14/92	CHOS 6/19	SEDIMENT	nd	nd	nd	nd	nd
7/14/92	EVOS 6/18	SEDIMENT	nd	nd	nd	nd	nd
7/15/92	BVOS 6/18	SEDIMENT	nd	nd	nd	nd	nd
7/15/92	WFOS 6/19	SEDIMENT	nd	nd	nd	nd	nd
7/25/92	WFF 7/18	FIELD SAMPLE	nd	nd	nd	nd	nd
7/25/92	BVF 7/17	FIELD SAMPLE	nd	nd	0.4842	nd	nd
7/26/92	WFOS 7/19	FIELD SAMPLE	nd	nd	nd	nd	nd
7/25/92	BVOS 7/18	SEDIMENT	nd	nd	nd	nd	nd
7/25/92	EV#3Surf 7/18	SEDIMENT	nd	nd	nd	nd	nd
7/26/92	EVSurf1 7/18	SEDIMENT	nd	nd	nd	nd	nd
7/26/92	EVOS 7/18	SEDIMENT	nd	nd	nd	nd	nd
7/26/92	EV3ft#2 7/18	SEDIMENT	nd	nd	nd	nd	nd
8/4/92	EVLot#1 6' 7/18	FIELD SAMPLE	nd	nd	nd	nd	nd
8/4/92	EVLot#3 7/18	FIELD SAMPLE	nd	nd	nd	nd	nd
8/4/92	EV 4FT1 7/18	FIELD SAMPLE	nd	nd	nd	nd	nd

Table 2E Continued

Extraction date	Sample	Sample Type	Carb	Atr	Ala	Met	Cyan
8/4/92	EV WT#2 7/18	FIELD SAMPLE	nd	nd	nd	nd	nd
8/4/92	EVSurf2 7/18	SEDIMENT	nd	nd	nd	nd	nd
8/5/92	EV 3ft3 7/18	SEDIMENT	nd	nd	nd	nd	nd
10/21/92	BVF 10/10	FIELD SAMPLE	nd	nd	0.162	nd	nd
10/25/92	BVF 10/10	MATRIX SPIKE	5.693	2.706	3.456	0.2236	2.456
10/21/92	WFF 10/11	FIELD SAMPLE	nd	nd	nd	0.0083	nd
10/21/92	CHF 10/11	FIELD SAMPLE	nd	nd	nd	0.0614	nd
10/25/92	EVF 10/10	FIELD SAMPLE	nd	nd	nd	nd	nd
10/21/92	LB 10/21	LAB BLANK	nd	nd	nd	nd	nd
10/22/92	BVOS 10/10	SEDIMENT	nd	nd	nd	nd	nd
10/22/92	WFOS 10/11	SEDIMENT	nd	nd	nd	nd	nd
10/25/92	EVOS 10/10	SEDIMENT	nd	nd	nd	nd	nd
10/25/92	CHOS 10/11	SEDIMENT	nd	nd	nd	nd	nd
11/18/92	BVF 11/14	FIELD SAMPLE	nd	nd	nd	nd	nd
11/18/92	EVF 11/14	FIELD SAMPLE	nd	nd	nd	nd	nd
11/19/92	CHF 11/15	FIELD SAMPLE	nd	nd	nd	0.0547	nd
11/19/92	WFF 11/15	FIELD SAMPLE	nd	nd	nd	nd	nd
11/19/92	WFOS 11/15	SEDIMENT	nd	nd	nd	nd	nd
11/19/92	CHOS 11/15	SEDIMENT	nd	nd	nd	nd	nd

Table 2E Continued

Extraction date	Sample	Sample Type	Carb	Atra	Alac	Met	Cyan
11/18/92	EVOS 11/14	SEDIMENT	nd	nd	nd	nd	nd
11/18/92	BVOS 11/14	SEDIMENT	nd	nd	nd	nd	nd
12/8/92	EVF 12/5	FIELD SAMPLE	nd	nd	nd	0.0074	nd
12/8/92	BVF 12/5	FIELD SAMPLE	nd	nd	0.0198	nd	nd
12/9/92	WFF 12/6	FIELD SAMPLE	nd	nd	nd	0.0071	nd
12/8/92	EVOS 12/5	SEDIMENT	nd	nd	nd	nd	nd
12/8/92	BVOS 12/5	SEDIMENT	nd	nd	nd	nd	nd
12/9/92	WFOS 12/6	SEDIMENT	nd	nd	nd	nd	nd
12/9/92	WFOS 12/6	MATRIX SPIKE	nd	5.430	1.502	8.4971	2.7955
1/12/93	WFF 1/11	FIELD SAMPLE	nd	nd	nd	0.0087	nd
1/12/93	BVF 1/10	FIELD SAMPLE	nd	nd	nd	0.0172	nd
1/12/93	EVF 1/10	FIELD SAMPLE	nd	nd	nd	nd	nd
1/12/93	WFOS 1/11	SEDIMENT	nd	nd	nd	nd	nd
1/25/93	EVF 1/10 #4	REP. MAT. SPIKE	nd	0.9236	1.0351	1.0095	1.3128
1/25/93	EVF 1/10 #3	REP. MAT. SPIKE	nd	1.0206	1.0203	1.0071	1.7006
1/25/93	EVF 1/10 #2	REP. MAT. SPIKE	nd	0.7464	0.8051	0.7844	0.8785
1/25/93	EVF 1/10 #1	REP. MAT. SPIKE	nd	1.0759	1.1046	1.053	1.8114
2/17/93	BVF 2/12	FIELD SAMPLE	nd	nd	nd	nd	nd
2/17/93	EVF 2/12	FIELD SAMPLE	nd	nd	nd	nd	nd



Table 2E Continued

Extraction date	Sample	Sample Type	Carb	Atra	Ala	Met	Cyan
2/18/93	CHF 2/13	FIELD SAMPLE	nd	nd	nd	0.0067	nd
2/18/93	WFF 2/13	FIELD SAMPLE	nd	nd	nd	nd	nd
2/17/93	EVOS 2/12	SEDIMENT	nd	nd	nd	nd	nd
2/17/93	BVOS 2/12	SEDIMENT	nd	nd	nd	nd	nd
2/18/93	WFOS 2/13	SEDIMENT	nd	nd	nd	nd	nd

## Appendix F

### Ohmicron Sample Data

Each month from May to February 1992, an independent environmental company, Ohmicron Corporation, received one water sample from each well and seepage meter. Ohmicron used an immunoassay procedure to test the samples for the pesticides included in the study. Table 1F listed the LOD's for the pesticides used in the study. The sampling data was incorporated in tables 2F through 10F.

Abbreviations were:

BV: Agricultural Site 1

CH: Agricultural Site 2

EV: Agricultural Site 3

WF: Agricultural Site 4

ST: Wetlands Site 1

nd: not detected

EB: Equipment blank

LB: Laboratory blank

LFB: Laboratory fortified blank (pesticides added)

TB: Travel blank

Table 1F  
Ohmicron Limits of Detection

Pesticide	Concentration, $\mu\text{g/L}$
Alachlor	0.05
Atrazine	0.04
Carbofuran	0.06
Cyanazine	0.04
Metolachlor	0.06

\* Note that the pesticides' MDLs were not provided, but are assumed to be higher.

Table 2F  
May Ohmicron Sample Results

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Metol	Carb
WFAmbTo	1a		nd	nd	nd	nd	nd
CHAmbTo	2a		nd	nd	nd	nd	nd
BVAmbTo	3a		nd	nd	nd	nd	nd
EVAmbTo	4a		nd	nd	nd	nd	nd
LFB #2	5a	2 µg/L	1.19	1.21	1.28	1.18	0.46
EB #2	6a		nd	nd	nd	nd	nd
EB #1	7a		nd	nd	nd	nd	nd
BVSM 3#2	8a		nd	nd	nd	nd	nd
BVSM 1	9a		nd	nd	nd	nd	nd
BVSM 2	10a		nd	nd	nd	nd	nd
BVSM 3	11a		nd	nd	nd	nd	nd
BVSM 4	12a		nd	nd	nd	nd	nd
BVCW #2	13a		nd	nd	nd	nd	nd
Tvl Blank	14a		nd	nd	nd	nd	nd
BV2 #2	15a		nd	nd	0.15	nd	nd
Lab Blank #1	16a		nd	nd	nd	nd	nd
BV1 #2	17a		nd	nd	0.18	nd	nd
BV2 #1	18a		nd	nd	0.1	nd	nd
BV1 #1	19a		nd	nd	0.23	nd	nd
CHNew #1	20a		nd	nd	nd	0.1	nd
BVCW #1	21a	2 µg/L	1.48	0.16	2.66	2.61	0.1
BVCW #3	22a		nd	0.06	nd	nd	nd
EV2 #1	23a		nd	nd	nd	nd	nd
CH2	24a		nd	nd	nd	nd	0.33
WFSM 1-4	25a		nd	nd	nd	nd	nd
EVF/R #1	26a		nd	nd	nd	nd	nd
EV2 #2	27a	2 µg/L	2.13	0.16	2.65	2.66	nd
EV2 #3	28a		nd	nd	nd	nd	nd
EV1 #1	29a	2 µg/L	1.26	0.11	2.31	2.49	nd

Table 2F Continued

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
EV1 #2	30a		nd	nd	nd	nd	nd
EVF/R #2	31a		nd	nd	nd	nd	nd
EV1 #3	32a		nd	nd	nd	nd	nd
CHSM 3	33a		nd	nd	nd	nd	nd
EVSM 1&4	34a		nd	nd	nd	nd	nd
CHSM 4	35a	10 µg/L	8.5	0.3	9.5	13.6	0.16
EVSM 2	36a		nd	nd	nd	nd	nd
WF	37a		nd	nd	nd	nd	nd
WFSW	38a		nd	nd	nd	nd	nd
CH5	39a		nd	nd	nd	nd	nd
CH	40a		nd	nd	nd	nd	nd

**Table 3F**  
**June Ohmicron Sample Results**

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
EVF/R	1b		nd	nd	nd	nd	nd
EV2	2b		nd	nd	nd	nd	nd
BVCW	3b		nd	nd	nd	nd	nd
CHNew	4b		nd	nd	nd	nd	nd
WFSW #2	5b		nd	nd	nd	nd	nd
WFSW #1	6b		nd	nd	nd	nd	nd
BV1	7b		nd	nd	0.19	nd	nd
WF1 #1	8b		nd	nd	nd	nd	nd
BVSM 1	9b		nd	nd	nd	nd	nd
WF2 #2	10b		nd	nd	nd	nd	nd
CHSM 2	11b		nd	nd	nd	nd	nd
CH2	12b		nd	nd	nd	nd	nd
WF2 #1	13b		nd	nd	nd	nd	nd
EV1	14b		nd	nd	nd	nd	nd
BV2	15b		nd	nd	0.1	nd	nd
BVSM 4	16b		nd	nd	nd	nd	nd
WF1 #2	17b		nd	nd	nd	nd	nd

Table 4F  
July Ohmicron Sample Results

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
EV1 #4	1c	4 µg/L	4.9	7.1	6.6	0.06	6.76
EV1 #3	2c	2 µg/L	9.4	12	11.5	0.16	9.88
WF1 #4	3c	4 µg/L	14.5	15.6	19.4	0.16	11.92
WF1 #3	4c	2 µg/L	3.69	5.7	5.8	0.06	3.65
WF1	5c		nd	nd	nd	nd	nd
BV2	6c		nd	nd	0.09	nd	nd
EV1	7c		nd	nd	nd	nd	nd
LB #3	8c		nd	nd	nd	nd	nd
BV2 #3	9c	2 µg/L	4.3	7.4	7.8	0.07	6.36
CH5 #3	10c	2 µg/L	4.1	8.9	7.2	0.11	5.64
EVF/R	11c		nd	nd	nd	nd	nd
WFSW	12c		nd	nd	nd	nd	nd
LFB #1	13c	2 µg/L	nd	nd	nd	nd	nd
BV2 #4	14c	4 µg/L	11.3	15.4	14.6	0.17	10.16
LB #2	15c		nd	nd	nd	nd	nd
LB #1	16c		nd	nd	nd	nd	nd
CH5 #4	17c	4 µg/L	10.4	14.4	14.2	0.18	9.64
EV2	18c		nd	nd	nd	nd	nd
CH5	19c		nd	nd	nd	nd	nd
CH2	20c		nd	nd	nd	nd	nd
WF2	21c		nd	nd	nd	nd	nd
WFSM 2	22c		nd	nd	nd	nd	nd
BV1	23c		nd	nd	0.27	0.05	nd
CHSM 4	24c		nd	nd	nd	nd	nd
WFSM 1	25c		nd	nd	nd	nd	nd
CHNew	26c		nd	nd	nd	nd	nd
CHSM 1	27c		nd	0.05	nd	nd	nd
BVCW	28c		nd	nd	nd	nd	nd

Table SF  
August Ohmicron Sample Results

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
BV2	1d		0.13	nd	0.61	nd	nd
Hopper	2d		nd	nd	0.06	nd	nd
WF1 #4	3d	4 µg/L	3.8	4.5	4.67	6.92	4.94
EV1	4d		nd	nd	nd	nd	nd
EVF/R	5d		nd	nd	0.09	nd	nd
WFSM 4	6d		0.36	0.06	0.21	nd	nd
EV2	7d		nd	nd	nd	nd	nd
BVCW #1	8d		0.36	nd	0.64	nd	nd
CHSM 3	9d		0.5	0.06	0.19	nd	nd
WFSM 3	10d		0.51	0.09	0.28	nd	nd
EB	11d		0.07	nd	nd	nd	nd
CH3	12d		nd	nd	nd	nd	nd
BVSM 1	13d		0.14	nd	0.11	nd	nd
CHSM 1	14d		0.47	nd	0.09	nd	nd
WFSM 2	15d		0.24	nd	0.06	nd	nd
LFB #3	16d	4 µg/L	8.3	7.9	6.9	1.42	4.56
BVCW	17d		0.12	nd	0.07	nd	nd
CH5	18d		nd	nd	nd	nd	nd
CH2 #1	19d		nd	nd	nd	nd	nd
LFB #1	20d	2 µg/L	3.32	3.8	3.83	0.37	2.5
WFSW	21d		nd	nd	nd	nd	nd
WF1	22d		nd	nd	nd	nd	nd
WF1 #2	23d	2 µg/L	2.45	2.43	2.95	4.94	2.8
NOLABEL	24d		0.11	nd	nd	nd	nd



Table 6F  
October Ohmicron Sample Results

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
CH5	1e		nd	nd	nd	nd	nd
EV2 #4	2e	1 µg/L	1.18	1.13	1.15	nd	1.05
BVSM 3	3e		nd	nd	nd	nd	nd
BV1 #4	4e	2 µg/L	2.22	1.92	2.46	N/A	2.0
WFSM 2	5e		nd	nd	nd	nd	nd
CH3 #4	6e	1 µg/L	1.26	1.13	1.15	N/A	1.1
WF1 #3	7e	0.5 µg/L	0.66	0.66	0.63	nd	0.62
WF1 #4	8e	2 µg/L	2.37	2.14	2.27	nd	2.09
CHSM 1	9e		0.08	nd	nd	nd	nd
EV2	10e		nd	nd	nd	nd	nd
WF1	11e		nd	nd	nd	nd	nd
CHSM 2	12e		0.11	nd	nd	nd	nd
LB #1	13e		nd	nd	nd	nd	nd
BVSM 2	14e		nd	nd	nd	nd	nd
CH3	15e		nd	nd	nd	nd	nd
WFSM 1	16e		nd	nd	nd	nd	nd
EV1 #4	17e	1 µg/L	1.48	1.3	1.3	N/A	1.32
CHSM 4	18e		0.1	nd	nd	nd	nd
EVSM 1	19e		nd	nd	nd	nd	nd
CHSM 3	20e		nd	nd	nd	nd	nd
BVCW	21e		nd	nd	0.1	nd	nd
BV2	22e		nd	nd	0.09	nd	nd
BVSM 4	23e		nd	nd	nd	nd	nd
EVFR	24e		nd	nd	nd	nd	nd
EVSM 2	25e		nd	nd	nd	nd	nd
BVSM 4 #1	26e	2 µg/L	2.37	2.06	2.16	nd	2.16
ST3	27e		nd	nd	nd	nd	nd
BVCW #3	28e	1 µg/L	1.31	1.19	1.34	nd	1.28

Table 6F Continued

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
CH2	29e		0.12	nd	nd	nd	nd
WF2	30e		0.05	nd	nd	nd	nd
EVSM 3	31e		nd	nd	nd	nd	nd
BV1 #3	32e	0.5 µg/L	0.75	0.65	0.85	N/A	0.71
EV1	33e		nd	nd	nd	nd	nd
WPSW	34e		nd	nd	nd	nd	nd
EVSM 1 #1	35e	0.25 µg/L	0.35	0.35	0.28	nd	0.37
CH3 #1	36e	0.25 µg/L	0.4	0.31	0.25	nd	0.42
WFSM 3	37e		nd	nd	nd	nd	nd
EVSM 4	38e		nd	nd	nd	nd	nd
CH5 #2	39e	2 µg/L	2.52	2.17	1.91	nd	2.36

Table 7F  
November Ohmicron Sample Results

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
CH3	1f		nd	nd	nd	nd	nd
EV2	2f		nd	nd	nd	nd	nd
CH5	3f		nd	nd	nd	nd	nd
WFSM 1	4f		nd	nd	nd	nd	nd
EVF/R	5f		nd	nd	nd	nd	nd
CH2	6f		0.1	nd	nd	nd	nd
WF2	7f		nd	nd	nd	nd	nd
BVSM 2	8f		nd	nd	nd	nd	nd
WFSW	9f		nd	nd	nd	nd	nd
EV1	10f		nd	nd	nd	nd	nd
BV2	11f		nd	nd	0.12	nd	nd
EVSM 2	12f		nd	nd	nd	nd	nd
ST1	13f		nd	nd	nd	nd	nd
TB #2	14f		0.06	nd	nd	nd	nd
WF1 #3	15f	0.25 µg/L	0.27	0.24	0.43	2.17	nd
WF2 #4	16f	0.5 µg/L	0.59	0.57	0.8	3.86	nd
BVCW #3	17f	0.25 µg/L	0.24	0.26	4.4	3.58	nd
WF2 #2	18f		nd	nd	nd	nd	nd
EV2 #4	19f	0.25 µg/L	0.24	0.24	0.41	2.25	nd
WF2 #1	20f		nd	nd	nd	nd	nd
EVFR #3	21f	0.5 µg/L	0.5	0.54	0.8	4.14	nd
LFB #2	22f	0.5 µg/L	0.6	0.62	0.8	nd	0.24
BVSM 3 #2	23f	2 µg/L	2.29	2.06	3.06	nd	2.12
BV1 #3	24f	0.5 µg/L	0.54	0.56	1.01	3.5	nd
WF1 #4	25f	1 µg/L	1.2	1.23	1.62	0.13	0.94
WF2 #3	26f	2 µg/L	2.24	2.03	2.57	nd	2.0
LFB #1	27f	0.25 µg/L	0.2	0.27	0.28	nd	nd
TB #1	28f		0.05	nd	nd	nd	nd

Table 7F Continued

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
CH3 #4	29f	0.25 µg/L	0.24	0.26	0.38	2.23	nd
CH3 #3	30f	1 µg/L	1.3	1.24	1.53	nd	1.11
EVF/R #4	31f	2 µg/L	2.23	2.03	2.8	nd	2.04
LFB #3	32f	2 µg/L	2.16	1.91	2.66	–	–
BV1 #4	33f	0.5 µg/L	0.57	0.55	0.99	4.17	nd
EV2 #3	34f	1 µg/L	1.3	1.16	1.57	nd	nd
EB #2	35f		0.07	nd	nd	nd	nd
BVSM	36f		nd	nd	nd	nd	nd
BV1 #1	37f	2 µg/L	2.5	2.05	3.16	nd	2.05
EVSM 3	38f		nd	nd	nd	nd	nd
EB #1	39f		nd	nd	nd	nd	nd
BV1 #2	40f	1 µg/L	2.43	2.04	2.97	nd	1.82
BVCW #4	41f	1 µg/L	1.39	1.19	4.54	nd	0.94
WFSM 3	42f		nd	nd	nd	nd	nd
EVSM 1	43f		nd	nd	nd	nd	nd
BV1	44f		nd	nd	0.19	nd	nd
EVSM 4	45f		nd	nd	nd	nd	nd
BVSM 4	46f		0.05	nd	nd	nd	nd
WF1	47f		0.06	nd	nd	nd	nd
BVCW	48f		nd	nd	4.46	nd	nd

– Data missing

Table 8F  
December Ohmicron Sample Results

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
WFAmbTo	1g		nd	nd	nd	nd	nd
EVF/R	2g		nd	nd	nd	nd	nd
BVCW #3	3g	0.25 µg/L	0.51	0.42	4.06	3.92	nd
EVF/R #2	4g	2 µg/L	2.38	2.07	2.62	0.1	2.07
EV1	5g		nd	nd	nd	nd	nd
EV2	6g		nd	nd	nd	nd	nd
EVSM 2	7g		nd	nd	nd	nd	nd
BV1	8g		nd	nd	0.12	nd	nd
BV2	9g		nd	nd	0.18	nd	nd
LB #2	10g		nd	nd	nd	nd	nd
WF1	11g		nd	nd	nd	nd	nd
LFB #2	12g	1 µg/L	1.32	1.2	1.54	1.84	nd
WF2	13g		nd	nd	nd	nd	nd
WF2 #1	14g	2 µg/L	21.6	16.1	21	0.58	16.5
WFSM 3	15g		nd	nd	nd	nd	nd
BVCW	16g		nd	nd	nd	nd	nd
SI2 #1	17g		nd	nd	nd	nd	nd
EVSM 1	18g		nd	nd	nd	nd	nd
WFSM 2	19g		0.07	nd	nd	nd	nd
LB #1	20g		nd	nd	nd	nd	nd
BV1 #2	21g	1 µg/L	1.36	1.28	1.56	0.1	1.47
LFB #1	22g	0.25 µg/L	0.41	0.35	0.35	nd	0.27
BVCW #1	23g	0.5 µg/L	0.41	0.3	4.14	2.52	nd
WFSW	24g		nd	nd	nd	nd	nd
BV1 #1	25g	2 µg/L	2.37	2.24	2.91	0.11	2.24

Table 9F  
January Ohmicron Sample Results

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
WF1 #3	1h	1 µg/L	1.34	1.04	1.15	nd	1.15
BVCW #2	2h	0.25 µg/L	0.29	0.33	1.57	0.49	nd
BV1 #3	3h	0.25 µg/L	0.28	0.31	0.37	0.39	nd
BV1 #1	4h	1 µg/L	17.7	16.6	20.4	0.76	–
WFSM 4 #3	5h	2 µg/L	2.33	2.04	2.65	nd	2.19
EVF/R	6h		nd	nd	nd	nd	nd
WF1 #4	7h	2 µg/L	2.52	2.08	2.45	nd	2.27
WF1	8h		nd	nd	nd	nd	nd
WFSM 4	9h		nd	nd	nd	nd	nd
EVF/R #4	10h	2 µg/L	2.92	2.17	2.67	0.16	2.42
EV2 #1	11h	0.25 µg/L	0.48	0.35	0.29	0.49	nd
BV2	12h		nd	nd	0.15	nd	nd
EV1	13h		nd	nd	nd	nd	nd
BVCW #4	14h	0.5 µg/L	0.79	0.59	2.01	0.85	nd
EB	15h		nd	nd	nd	nd	nd
WFSM 1	16h		nd	nd	nd	nd	nd
BV1 #4	17h	2 µg/L	1.97	2.26	2.79	0.14	1.83
EV2 #3	18h	0.5 µg/L	0.58	0.64	0.73	0.73	nd
EV2	19h		nd	nd	nd	nd	nd
BV1	20h		nd	nd	0.15	nd	nd
WFSM 3	21h		nd	nd	nd	nd	nd
WF2 #4	22h	0.5 µg/L	0.68	0.61	0.75	0.7	nd
TB #1	23h		nd	nd	nd	nd	nd
WF2 #3	24h	0.25 µg/L	0.34	0.27	0.38	0.31	nd
LFB #1	25h	1 µg/L	1.17	1.16	1.48	nd	0.85
BVCW	26h		nd	nd	1.22	nd	nd
WFSM 2	27h		nd	nd	nd	nd	nd

– Data missing

Table 9F Continued

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
LFB #2	28h	0.5 µg/L	0.62	0.62	0.72	0.7	nd
WF2	29h		nd	nd	nd	nd	nd
WFSW	30h		nd	nd	nd	nd	nd

Table 10F  
February Ohmicron Sample Results

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
CHSM 2	1i		nd	nd	nd	nd	nd
BV1 #4	2i	0.5 µg/L	0.66	0.7	0.84	0.93	nd
CH5	3i		nd	nd	nd	nd	nd
BV2	4i		nd	nd	nd	nd	nd
BVSM 4	5i		nd	nd	nd	nd	nd
EV1	6i		nd	nd	nd	nd	nd
WFSM 1	7i		nd	nd	nd	nd	nd
BV1	8i		nd	nd	0.16	0.1	nd
LB #2	9i		nd	nd	nd	nd	nd
LB #3	10i		nd	nd	nd	nd	nd
CH3 #4	11i	0.25 µg/L	0.33	0.35	0.37	0.49	nd
WFSM 2	12i		nd	nd	nd	nd	nd
WFSM 3	13i		nd	nd	nd	nd	nd
CH2	14i		0.25	nd	nd	nd	nd
BVCW #4	15i	0.5 µg/L	0.7	0.72	1.34	0.95	nd
BVSM 4 #3	16i	1 µg/L	1.34	1.36	1.57	0.05	1.04
CH5 #3	17i	0.25 µg/L	0.36	0.38	0.38	0.49	nd
BVSM 4 #4	18i	2 µg/L	2.51	2.26	3	0.11	2.07
LFB #2	19i	0.5 µg/L	0.75	0.74	0.66	0.77	nd
LFB #4	20i	2 µg/L	2.41	2.48	2.65	0.11	1.1
WFSM 4	21i		nd	nd	nd	nd	nd
BVSM 3	22i		nd	nd	nd	nd	nd
CH3 #1	23i	0.5 µg/L	0.7	0.76	0.75	0.91	nd
LB #1	24i		nd	nd	nd	nd	nd
LFB #1	25i	0.25 µg/L	0.28	0.31	0.3	0.3	nd
EV2 #4	26i	2 µg/L	2.38	1.99	2.53	nd	2.09
BVSM 3 #1	27i	0.5 µg/L	0.69	0.65	0.65	0.66	nd
LB #4	28i		nd	nd	nd	nd	nd



Table 10F Continued

Virginia Tech ID	Ohmicron ID	Spike Level	Atr	Cyan	Ala	Met	Carb
WF1 #3	29i	0.5 µg/L	0.79	0.64	0.66	0.63	nd
EV2	30i		nd	nd	nd	nd	nd
WF1 #4	31i	0.25 µg/L	0.38	0.33	0.32	0.37	nd
CH3	32i		nd	nd	nd	nd	nd
WF1	33i		nd	nd	nd	nd	nd
WFSW	34i		nd	nd	nd	nd	nd
EV2 #3	35i	1 µg/L	1.43	1.2	1.34	nd	1.13
BVCW	36i		nd	nd	0.54	nd	nd
EVF/R	37i		nd	nd	nd	nd	nd
BVCW #3	38i	0.25 µg/L	0.5	0.43	1.09	0.5	nd
BVSM 1	39i		nd	nd	nd	nd	nd
EB #2	40i		nd	nd	nd	nd	nd
EV1 #4	41i	0.25 µg/L	0.34	0.35	0.31	0.36	nd
WF2 #3	42i	1 µg/L	1.41	1.13	1.31	0.05	1.19
CHSM 3	43i		nd	nd	nd	nd	nd
WF2	44i		nd	nd	nd	nd	nd
BV1 #3	45i	0.25 µg/L	0.36	0.31	0.47	0.42	nd
EV1 #3	46i	0.5 µg/L	0.67	0.61	0.59	0.67	nd
BVSM 1 #4	47i	2 µg/L	2.38	1.91	2.9	1.09	2.13
CHSM 1	48i		nd	nd	nd	nd	nd
BVSM 1 #3	49i	1 µg/L	1.42	1.18	1.52	nd	1.03
CHSM 4	50i		nd	nd	nd	nd	nd
BVSM 2	51i		nd	nd	nd	nd	nd
LFB #3	52i	1 µg/L	1.27	1.11	1.23	nd	1.02

## Appendix G

### Quality Assurance/Quality Control

To insure proper QA/QC in the study, a laboratory blank, equipment blank, and travel blank were included for every 20 samples. Additionally, a laboratory spike, which was milli-Q water with the pesticides added, was included in every 20 samples. Two matrix spikes were made from each site's samples from April to June 1992, and 4 matrix spikes were made from each site's samples from August 1992, to February 1993. The samples were spiked at 4 and 10  $\mu\text{g/L}$  in April 1992, 2, and 4  $\mu\text{g/L}$  in June through August 1992, and then at 0.25, 0.5, 1, and 2  $\mu\text{g/L}$  for the rest of the study.

Acenaphthrene- $\text{d}_{10}$  and Phenanthrene- $\text{d}_{10}$  were added as surrogate to all the samples, but they did not produce sharp peaks when the GC/ECD was used, so 4-chlorobiphenyl was added to all samples after July 1992. The overall recovery rate of 4-chlorobiphenyl was 74%. The recovery rate for each sample was listed in Appendix A.

A graphical comparison of the added concentration of the spiked samples vs. the measured concentration was included as Figures 1G through 10G. Figures 1G-5G were Matrix spikes, and 6G-10G were laboratory spikes. The actual regression output for the matrix spikes was listed in table 1G, with table 2G containing the laboratory spikes.

**Table 1G**  
**Regression Output for Martix Spikes**

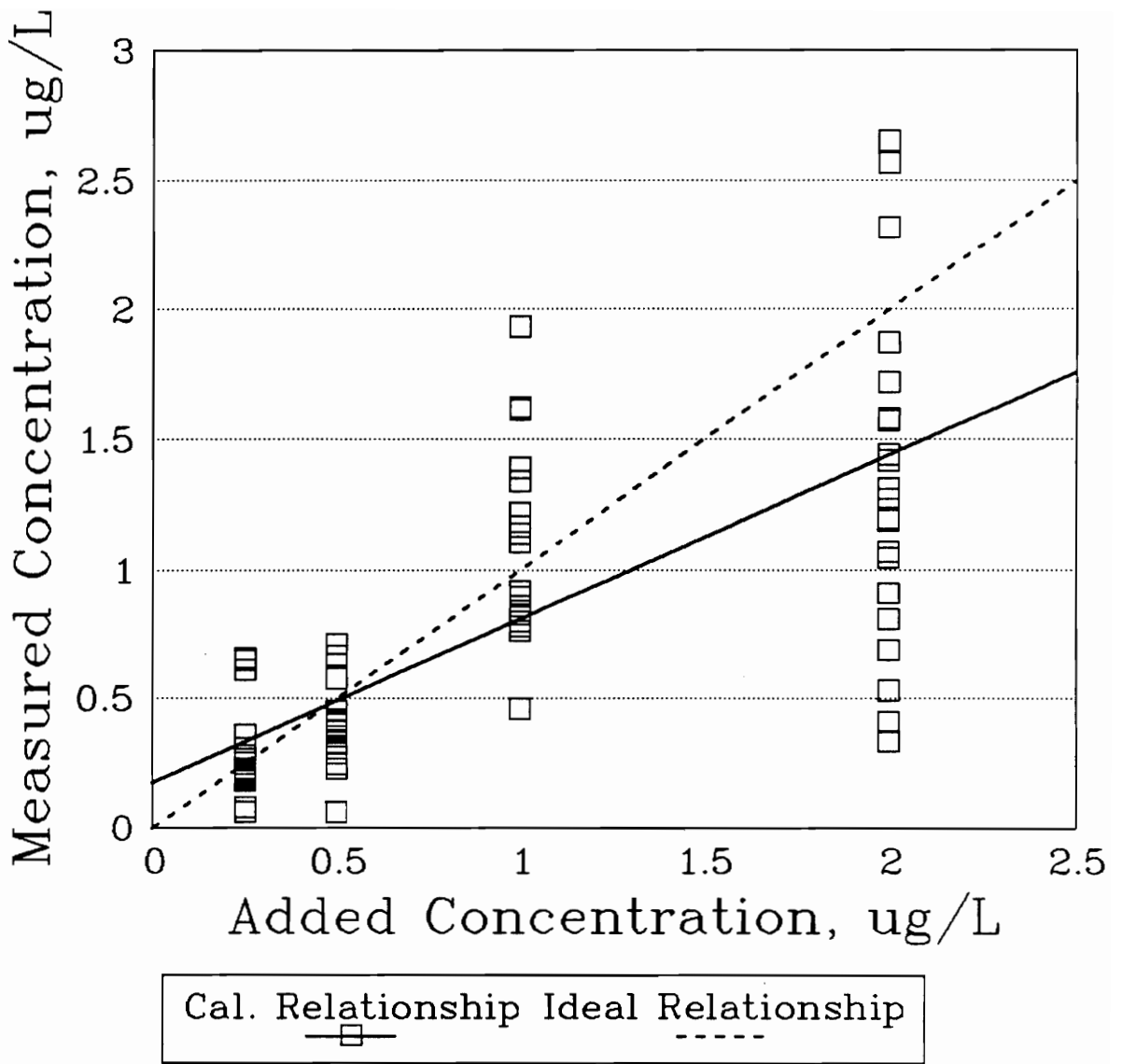
Pesticide	R <sup>2</sup>	Regression Output, where x = added conc. in µg/L
Atrazine	0.11	0.601*x + 0.428
Alachlor	0.53	0.632*x + 0.176
Carbofuran	0.002	-0.409*x + 4.78
Cyanazine	0.34	1.077*x + 0.481
Metolachlor	0.033	1.624*x + 0.254

\* Note that the regression output is the pesticide's concentration in µg/L.

Table 2G  
Regression Output for Laboratory Spikes

Pesticide	R <sup>2</sup>	Regression output, where x = added conc. in µg/L
Atrazine	0.33	0.866*x + 0.143
Alachlor	0.19	0.255*x + 0.317
Carbofuran	0.03	0.724*x + 3.81
Cyanazine	0.74	1.30*x + 0.177
Metolachlor	0.49	0.78*x + 0.113

\* Note that the regression output is the pesticide's concentration in µg/L.



**Figure 1G**  
**Measured vs. Added Concentration**  
**for Alachlor Matrix Spikes**

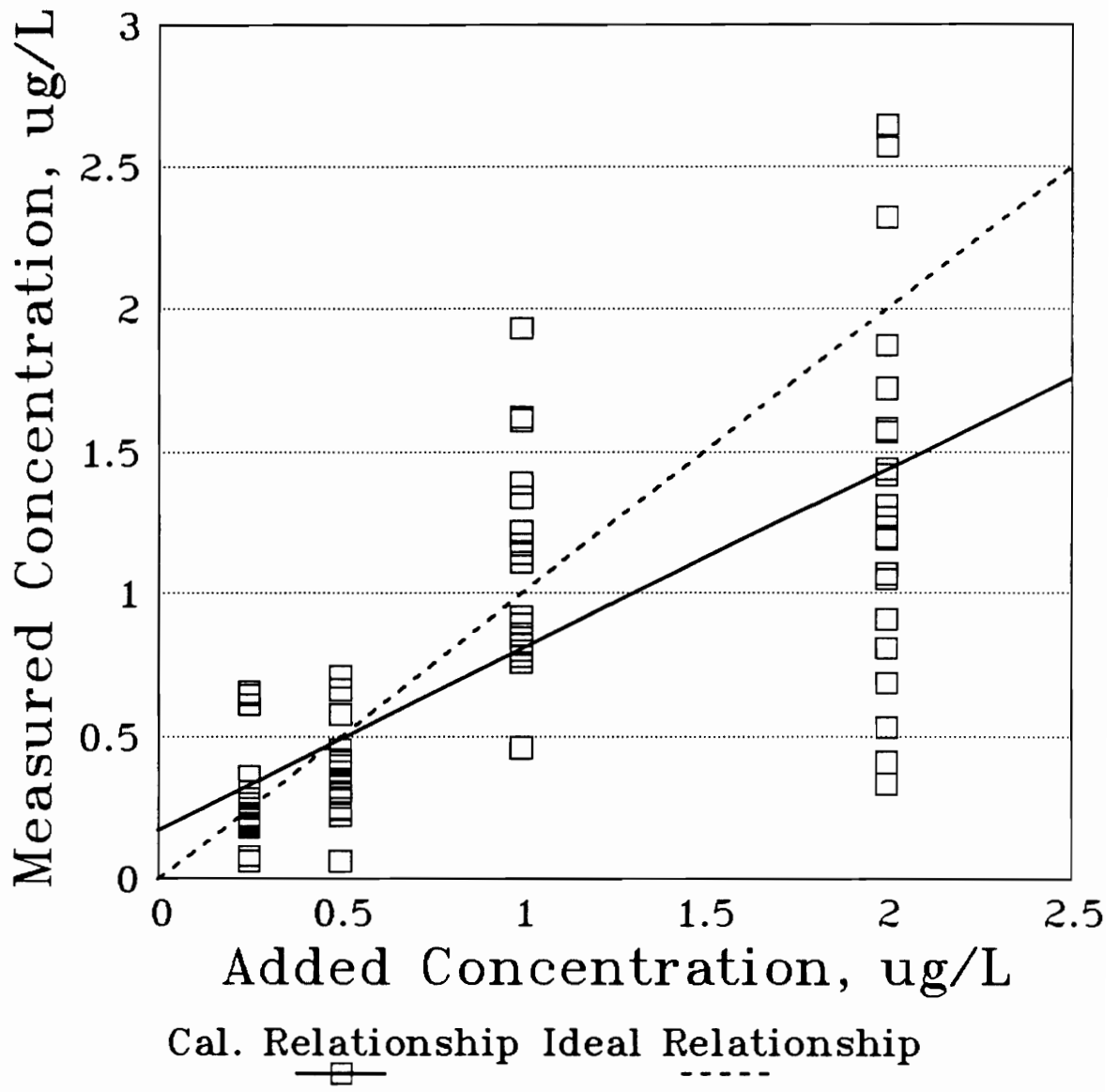


Figure 2G  
 Measured vs. Added Concentration  
 for Atrazine Matrix Spikes

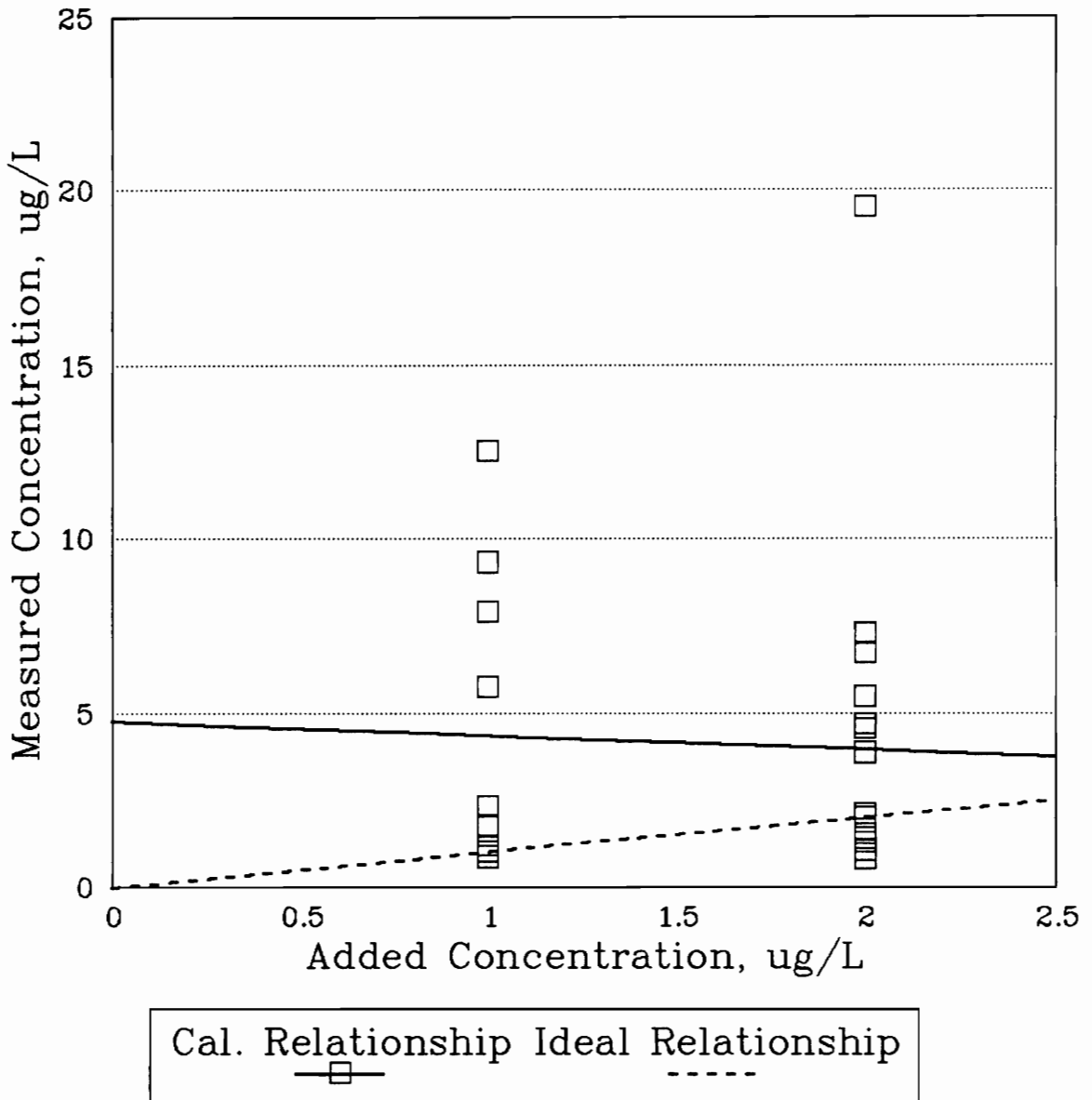


Figure 3G  
 Measured vs. Added Concentration  
 for Carbofuran Matrix Spikes

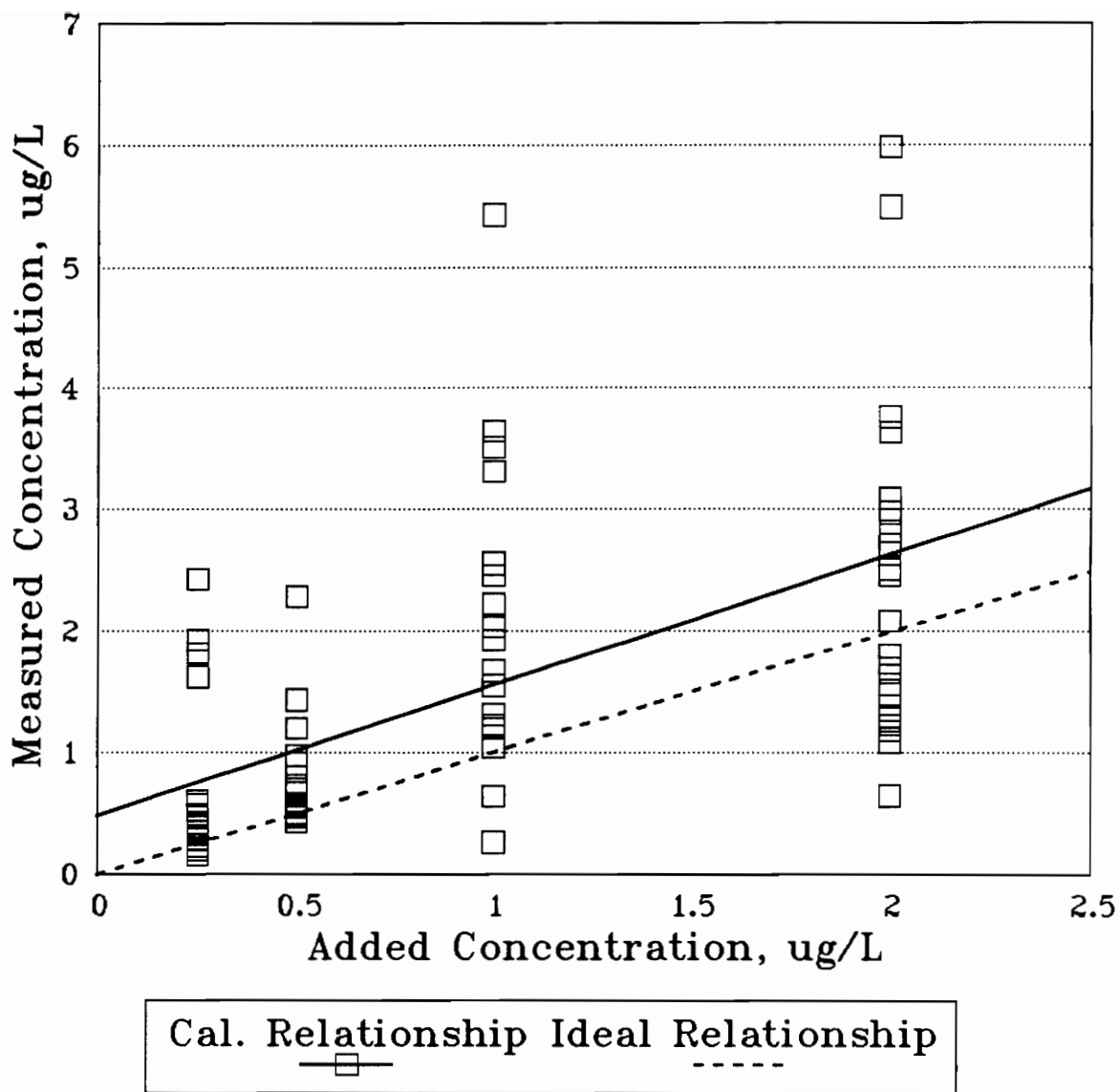


Figure 4G  
 Measured vs. Added Concentration  
 For Cyanazine Matrix Spikes



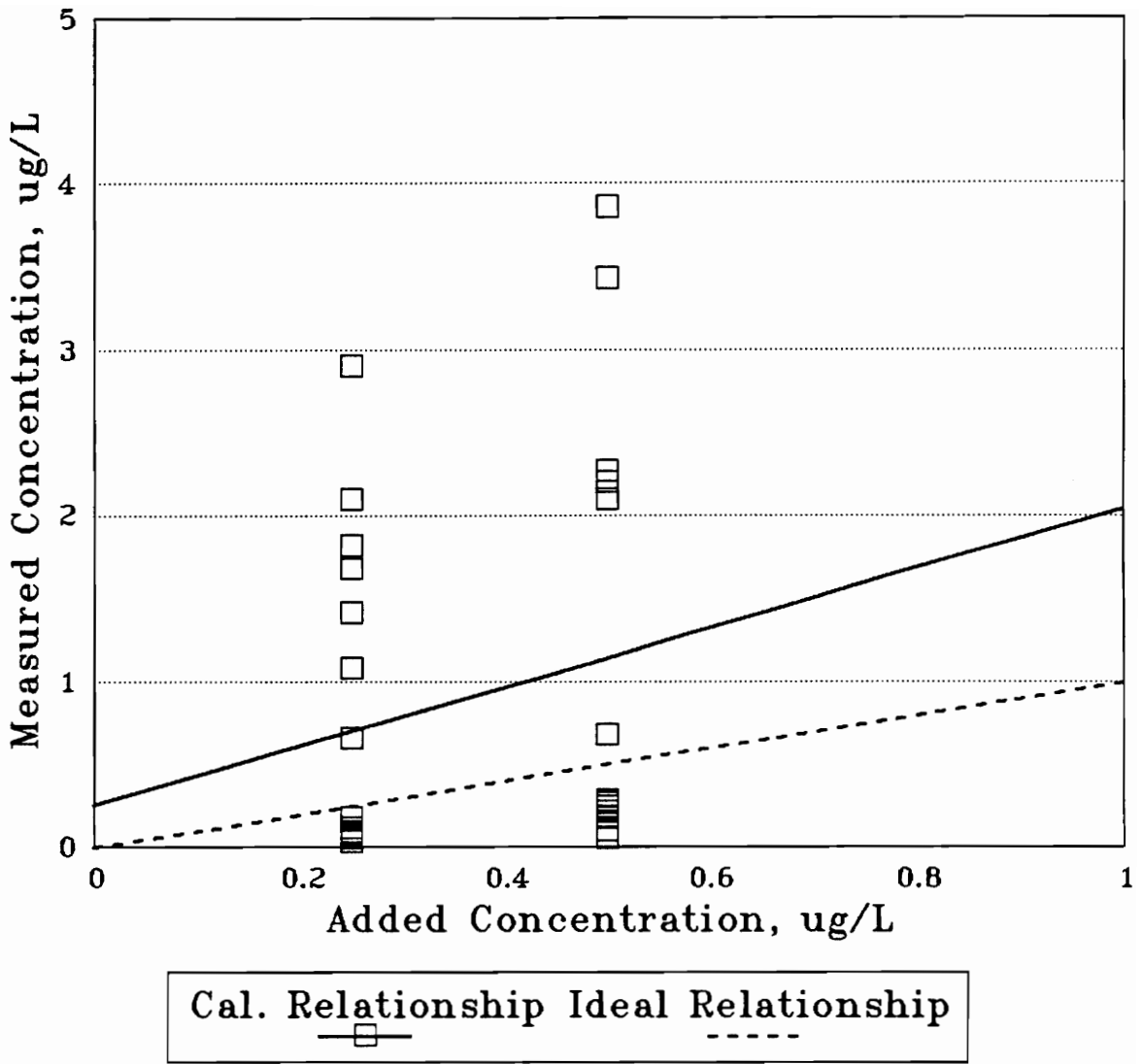
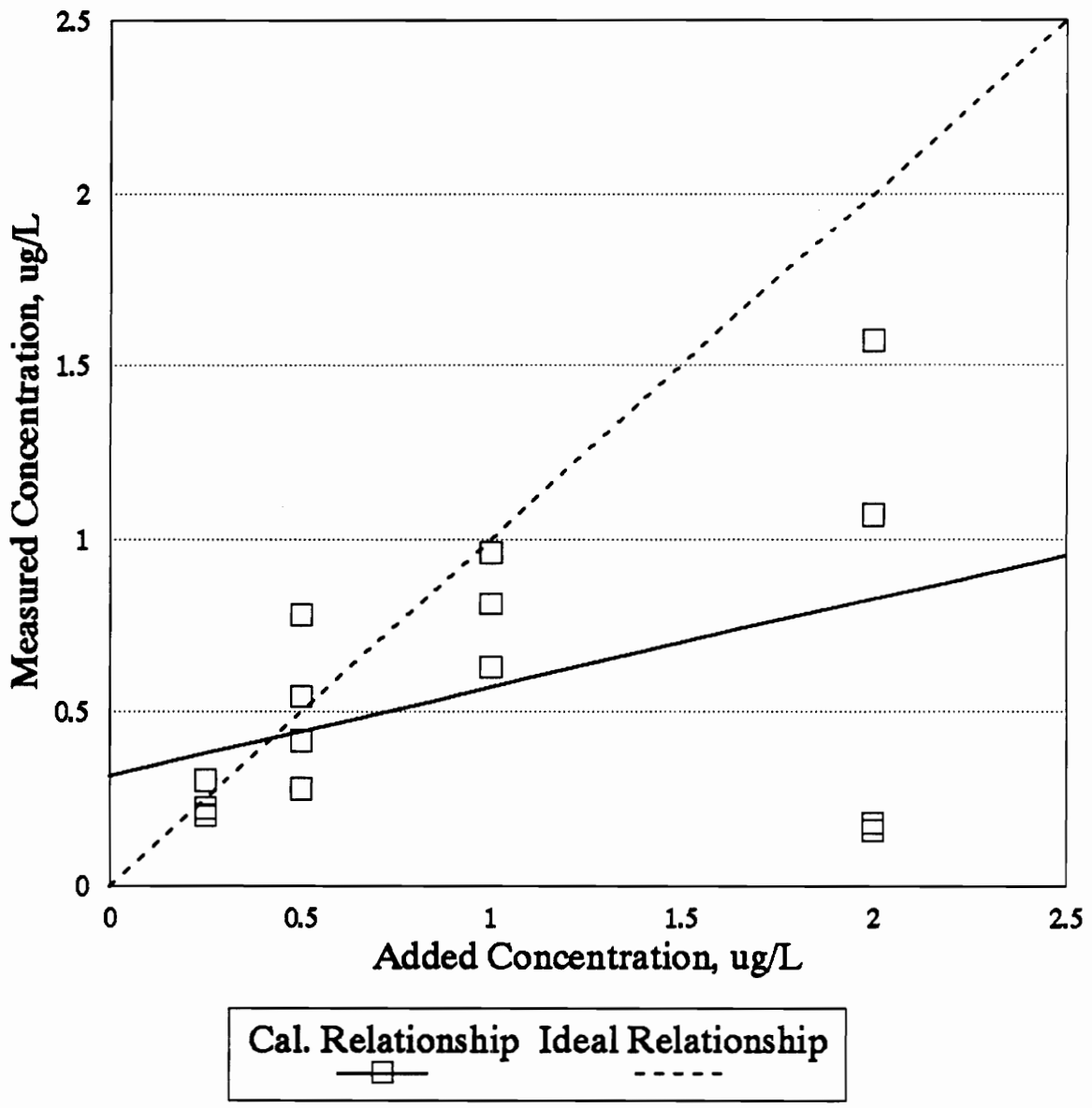
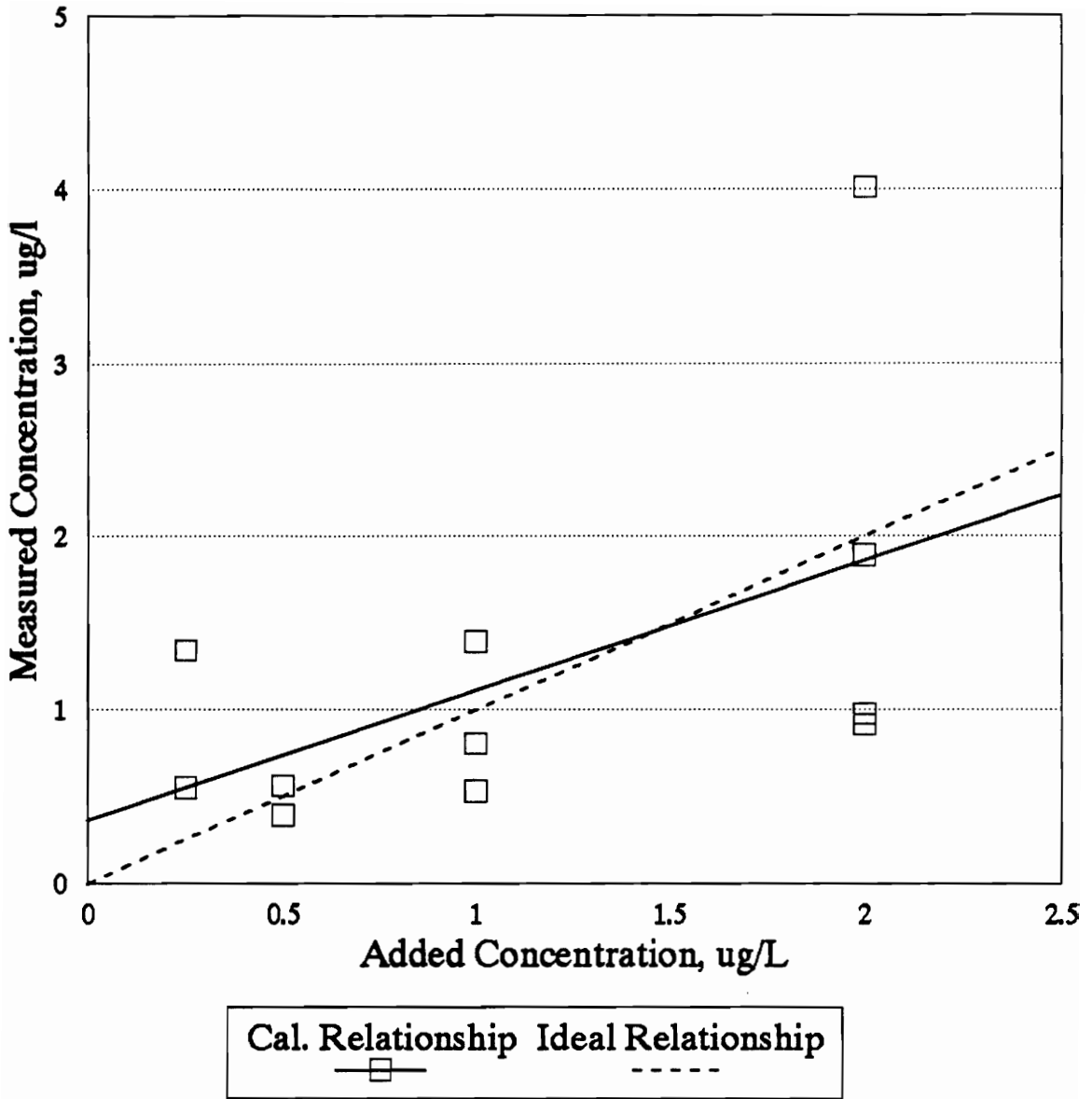


Figure 5G  
 Measured vs. Added Concentration  
 For Metolachlor Matrix spikes



**Figure 6G**  
**Measured vs. Added Concentration**  
**For Alachlor Laboratory Spikes**



**Figure 7G**  
**Measured vs. Added Concentrations**  
**For Atrazine Laboratory Spikes**

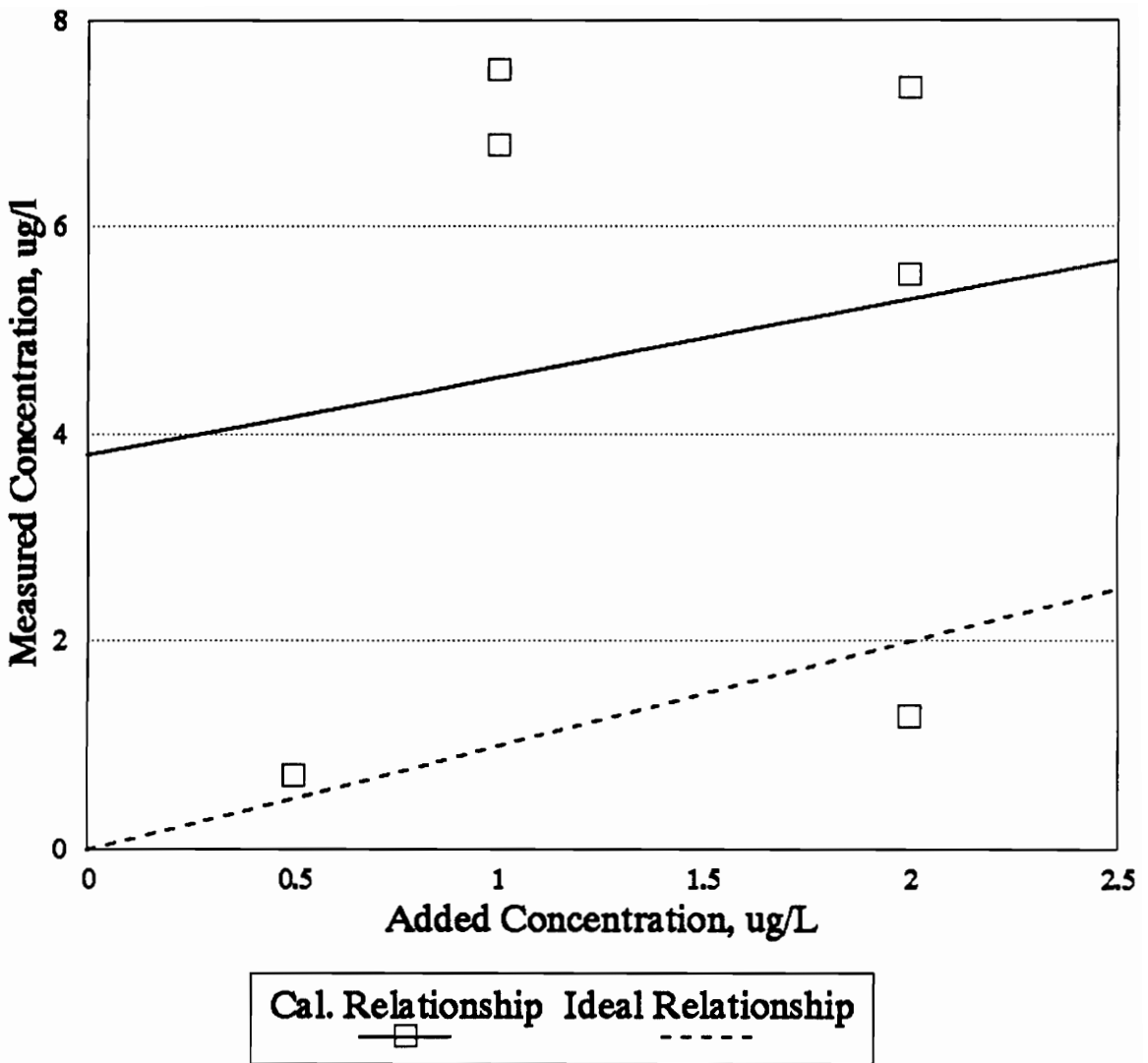
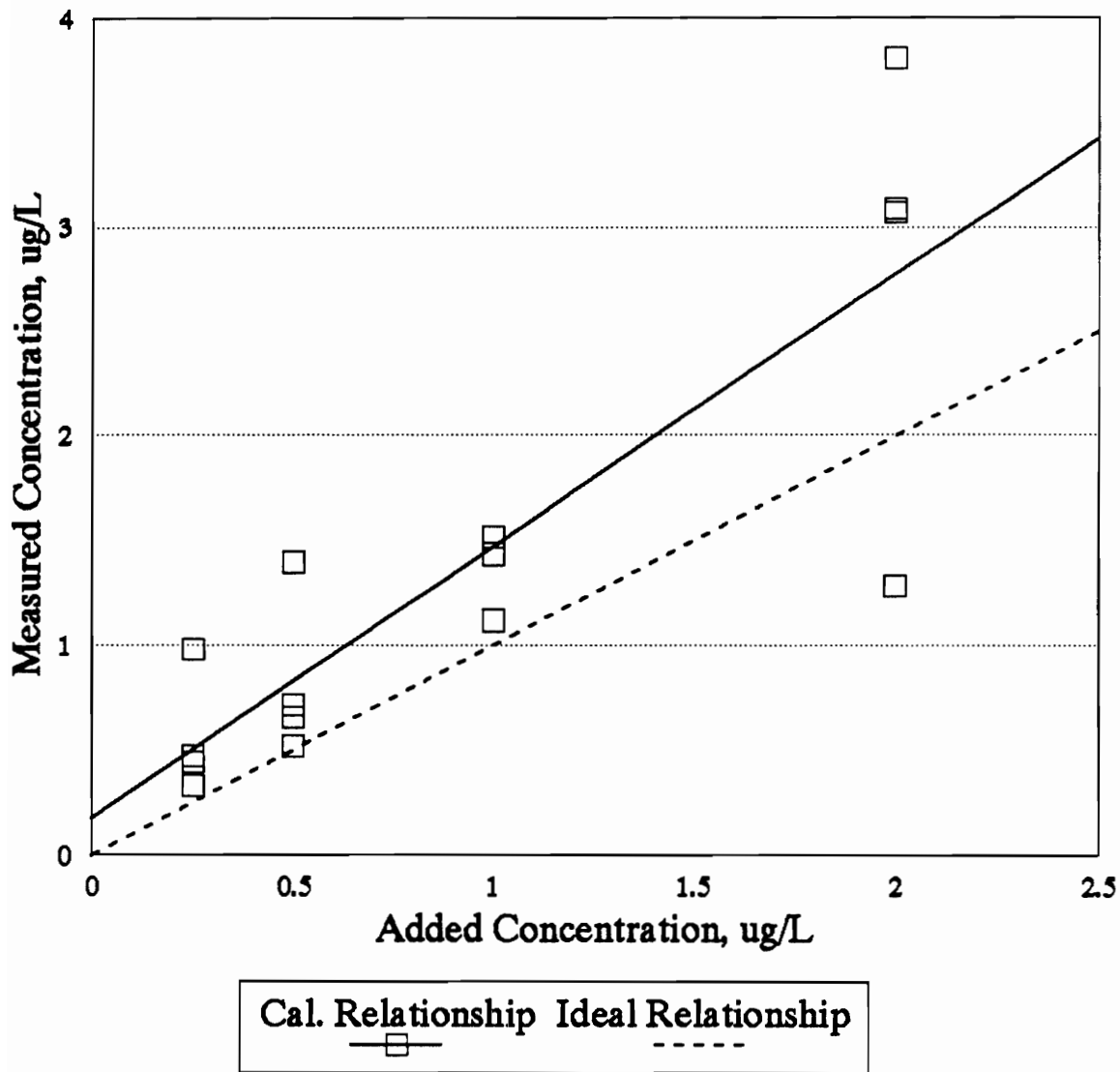
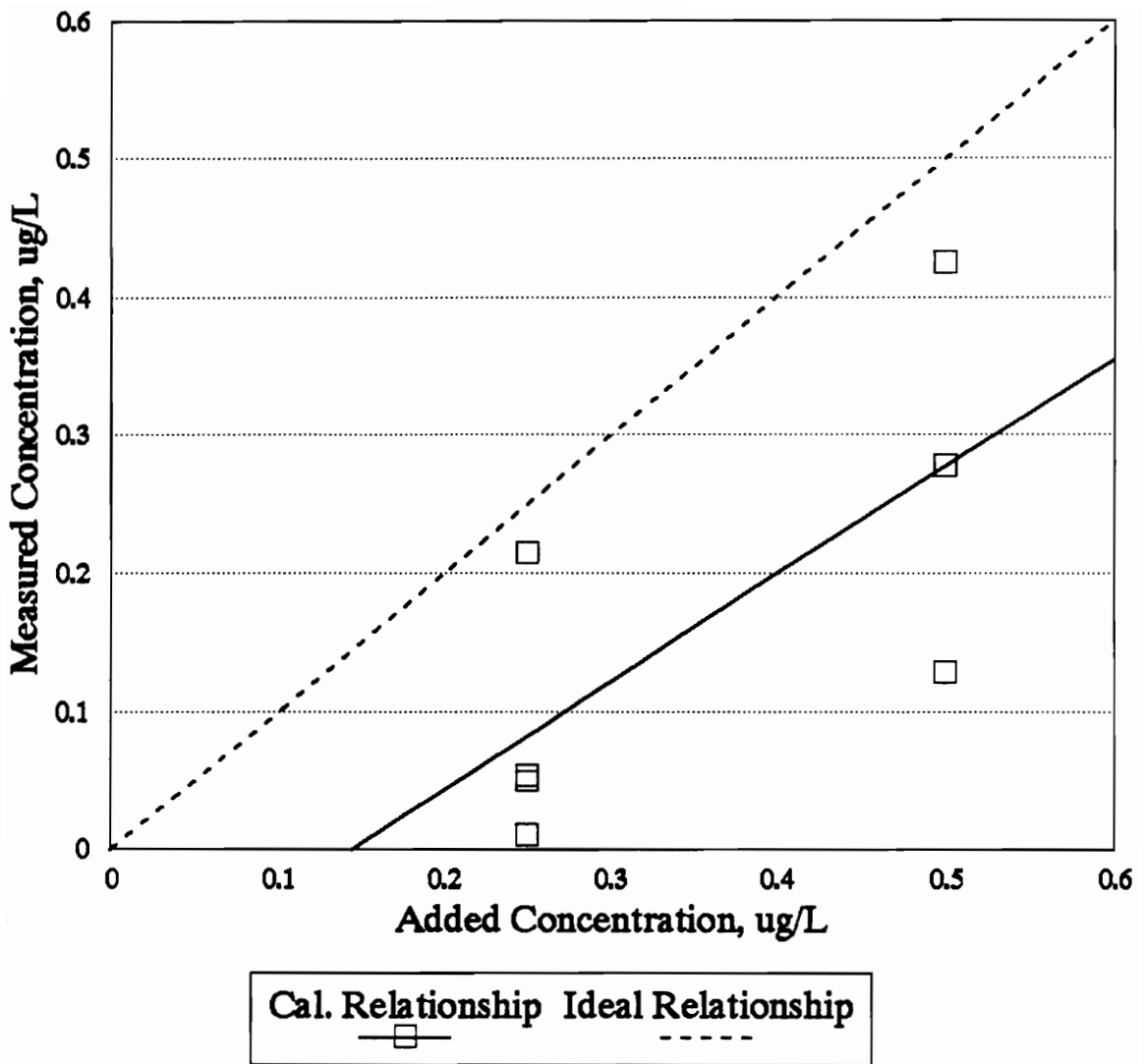


Figure 8G  
 Measured vs. Added Concentrations  
 For Carbofuran Laboratory Spikes



**Figure 9G**  
**Measured vs. Added Concentrations**  
**For Cyanazine Laboratory Spikes**



**Figure 10G**  
**Measured vs. Added Concentrations**  
**For Metolachlor Laboratory Spikes**

VITA

Thomas W. Hubbard was born on November 8, 1963 in Jacksonville, IL. He graduated *summa cum laude* from Jacksonville High School in 1981, and received a B. S. degree in Civil Engineering, *cum laude* from Bradley University in 1991.

A handwritten signature in black ink that reads "Thomas W. Hubbard". The signature is written in a cursive style with a large, sweeping initial 'T'.